

DEVELOPMENT AND VALIDATION OF STABILITY INDICATING REVERSED PHASE HIGH-PRESSURE LIQUID CHROMATOGRAPHY METHOD FOR SIMULTANEOUS ESTIMATION OF METFORMIN AND EMPAGLIFLOZIN IN BULK AND TABLET DOSAGE FORMGEETHA SWARUPA P¹, LAKSHMANA RAO K², PRASAD KRS¹, SURESH BABU K^{2*}¹Department of Chemistry, K. L. University, Guntur, Andhra Pradesh, India. ²Department of Chemistry, Mallareddy Engineering College, Hyderabad, Telangana, India. Email: babuiict@gmail.com

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ABSTRACT**Objective:** To develop accurate, fast, simple, and precise reversed-phase high-pressure liquid chromatography method for simultaneous determination of the binary mixture of metformin (MET) and empagliflozin (EMPA) in dosage forms.**Methods:** The method uses a mobile phase consisting of phosphate buffer, acetonitrile, methanol (15:80:5 v/v/v), an octadecyl silica C-18 column (4.6 mm × 250 mm, 5 μm particle size) in isocratic mode, detection wavelength of 227 nm, and a flow rate of 1 mL/minutes.**Results:** The measured retention times for MET and EMPA and were 2.528 and 4.140 minutes, respectively. The percentage recoveries of MET and EMPA were 101.12% and 100.55%, respectively. The relative standard deviation for assay of tablets was found to be <2%. The correlation coefficient for MET and EMPA was found to be 0.9990. The limit of detection and quantification for MET was 0.10 μg/mL and 0.31 μg/mL and for EMPA 0.01 μg/mL and 0.03 μg/mL.**Conclusion:** The method was fast, accurate, precise, and sensitive hence it can be employed for routine quality control of tablets containing both drugs in quality control laboratories and pharmaceutical industries.**Keywords:** Empagliflozin, Forced degradation method, Metformin, Reversed-phase high-performance liquid, Stability-indicating assay.**INTRODUCTION**

Metformin (MET) is chemically named as 4 1-carbamimidamido-N, N-dimethylmethanimidamide and it is as shown in Fig. 1. It is used as a biguanide antihyperglycemic agent used for treating non-insulin-dependent diabetes mellitus (NIDDM) [1,2]. It improves glycemic control by decreasing hepatic glucose production, decreasing glucose absorption, and increasing insulin-mediated glucose uptake. MET may induce weight loss and is the drug of choice for obese NIDDM patients. Use of MET is associated with modest weight loss.

Empagliflozin (EMPA) is chemically named as (2S,3R,4R,5S,6R)-2-[4-chloro-3-({4-[(3S)-oxolan-3-yloxy]phenyl)methyl}phenyl)-6-(hydroxymethyl)oxane-3,4,5-triol and it is as shown in Fig. 2. EMPA sodium glucose co-transporter-2 (SGLT-2) inhibitor indicated as an adjunct to diet and exercise to improve glycemic control in adult patients with Type 2 diabetes. SGLT-2 cotransporters are responsible for reabsorption of glucose from the glomerular filtrate in the kidney. The glucuretic effect resulting from SGLT-2 inhibition reduces renal absorption and lowers the renal threshold for glucose, therefore resulting in increased glucose excretion [3,4]. In addition, it contributes to reduced hyperglycemia and also assists weight loss and blood pressure reduction.

Literature survey revealed that few analytical methods are reported for analysis of both the drugs alone as well as in combination using ultraviolet (UV) spectrophotometry [5], high-pressure liquid chromatography (HPLC) [6-11], and ultra-pressure liquid chromatography [12].

The analytical method employed for the quantitative determination of drugs in formulation plays a significant role in the evaluation and interpretation of combined tablet dosage forms. Hence, an attempt has been made to develop new method for simultaneous estimation and validation of MET and EMPA in tablet formulation in accordance with the International Conference on Harmonization (ICH) guidelines [13].

METHODS**Chemicals**

The reference samples of MET and EMPA were provided as gift samples from Spectrum Pharma Research Solutions, Hyderabad.

HPLC grade acetonitrile, HPLC grade methanol, and all other chemicals were obtained from Merck chemical division, Mumbai. HPLC grade water obtained from Milli-Q water purification system was used throughout the study. Commercial tablets (Wormazan; Dosage: MET - 850 mg and EMPA - 5 mg) were purchased from the local pharmacy.

Methodology*Preparation of phosphate buffer pH 4.8*

Accurately weighed 1.36 g of potassium dihydrogen orthophosphate in a 1000 mL of volumetric flask add about 900 mL of milli-Q water added and degas to sonicate and finally make up the volume with water then added 1 mL of orthophosphoric acid then PH adjusted to 4.8 with dil. Orthophosphoric acid solution.

Preparation of mobile phase

Mixture of phosphate buffer (pH - 4.8), acetonitrile, and methanol (15:80:5 v/v/v).

Preparation of diluent

A mixture of water and methanol in the ratio 50:50 [%v/v].

Chromatographic parameters

Mobile phase: Phosphate buffer: Acetonitrile: Methanol (15:80:5% v/v/v) pH adjusted to 4.8

Column: Octadecyl silica C18, 250 mm × 4.6 mm, 5 μm
Wavelength: 227 nm

Flow rate: 1.0 mL/minutes
 Injection volume: 10 µL
 Run time: 10 minutes
 Detector: 2996 photodiode array detector.

Blank

A mixture of water and methanol in the ratio 50:50 [%v/v].

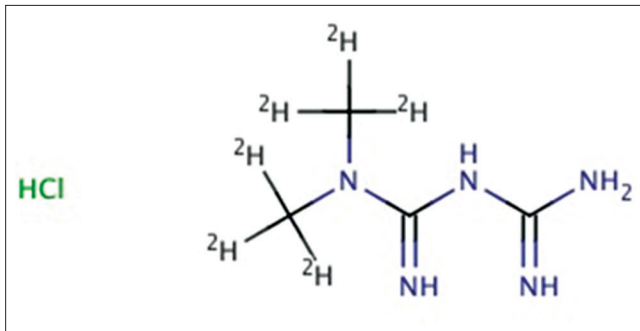


Fig. 1: Chemical structure of metformin

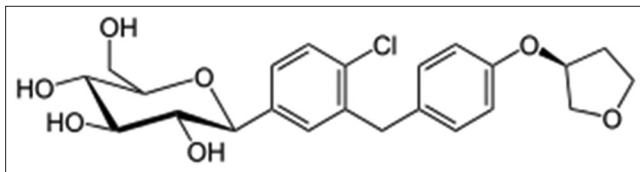


Fig. 2: Chemical structure of empagliflozin

Standard solution

To prepare the standard solution, 1000 ppm of MET and 12.5 ppm of EMPA was taken in to 10 mL of clean dry volumetric flask, add 7 mL of diluent (water:methanol, 50:50), then sonicated for 10 minutes and make up the volume with diluent.

Sample preparation

The chromatogram was shown in (Fig.3). One tablet was weighed, powdered and then was transferred into a 100 mL volumetric flask, 10 mL of diluent added and sonicated for 25 minutes, further the volume made up with diluent and filtered. From the filtered solution, 1 mL was pipeted out into a 10 mL volumetric flask and made up to 10 mL with diluent.

Label claim: 850 mg MET, 5 mg EMPA.

Placebo preparation

The placebo was prepared by weighing accurately 85 mg of MET and (5 mg of EMPA) was taken into a 100 mL volumetric flask and add 50 mL of diluent and sonicated for 15 minutes and the diluted to the mark. The chromatogram was shown in (Fig.4).

Method development

To establish and validate an efficient method for analysis of these drugs in pharmaceutical formulations, preliminary tests were performed. Different chromatographic conditions were employed for the analysis of the MET and EMPA in both bulk and pharmaceutical dosage form. The pure drugs of MET and EMPA were injected into the HPLC system and run using standard organic solvents commonly used for HPLC studies. Water, acetonitrile, and methanol were tested separately and in combination to find the best conditions for the separation of MET and EMPA. In focus to develop good symmetrical peak, water was replaced by phosphate buffer and it was observed that phosphate

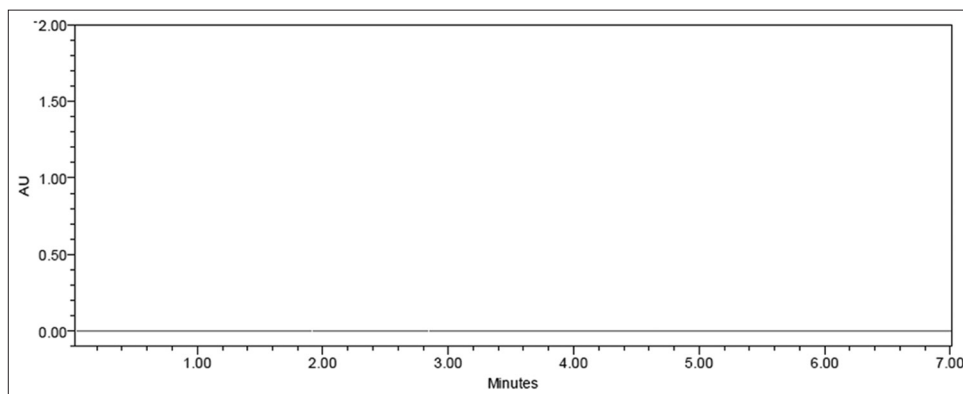


Fig. 3: Chromatogram of blank

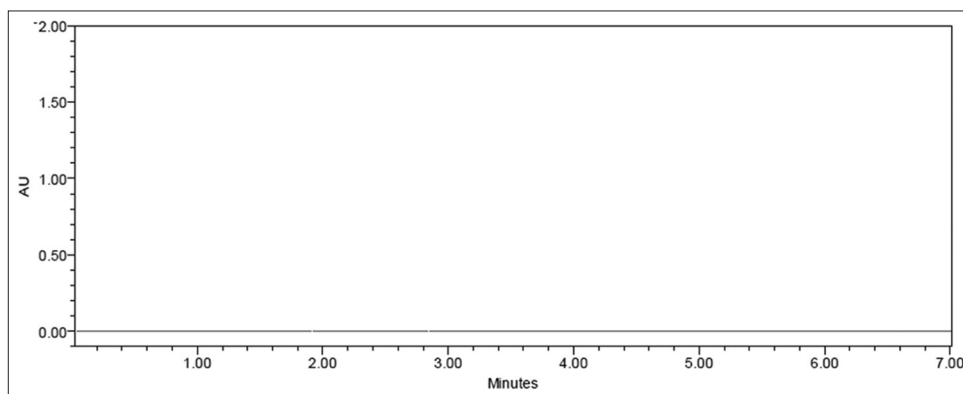


Fig. 4: Chromatogram of placebo

buffer, acetonitrile and methanol gave satisfactory results. This mobile phase system was tried with different proportions and with different flow rates. Finally, the optimal condition of the mobile phase was chosen as phosphate buffer: Acetonitrile: Methanol in the ration of 15:80:5% v/v/v. This composition of the mobile phase resolved the two drugs very well. All measurements were carried out at ambient temperature of the column. To optimize the flow rate, various flow rates were used. The optimal flow rate was 1 mL/minute for the presented work. The chromatogram was shown in (Fig.5).

Validation of quantitative HPLC method

The validation of the method was carried out as per ICH Guidelines [13]. The parameters assessed were specificity, linearity, precision, accuracy, stability, limit of detection (LOD), and limit of quantification (LOQ).

Specificity

Specificity is the ability of the analytical method to measure the analyte response in the presence of interferences including degradation products and related substances.

Accuracy

The accuracy was determined by calculating percentage recoveries of MET and EMPA. It was carried out by adding known amounts of each analyte corresponding to three concentration levels (50, 100, and 150%) of the labelled claim to the excipients. At each level, six determinations were performed, and the accuracy results were expressed as percent analyte recovered by the proposed method.

Precision

The precision of an analytical method is usually expressed as the standard deviation (SD). The repeatability studies were carried out by estimating response of MET and EMPA six times. The intra- and inter-day precision studies (intermediate precision) were carried out by estimating the corresponding responses three times on the same day and on three different days for three different concentrations, and the results are reported in terms of relative SD (RSD).

Linearity

The purpose of the test for linearity is to demonstrate that the entire analytical system (including detector and data acquisition) exhibits a linear response and is directly proportional over the relevant concentration range for the target concentration of the analyte. The linear regression data for the calibration plot is indicative of a good linear relationship between peak area and concentration over a wide range. The correlation coefficient was indicative of high significance.

Robustness

Robustness of the method was investigated under a variety of conditions including changes of composition of buffer in the mobile phase, flow rate, and temperature. This deliberate change in the method has no

affect on the peak tailing, peak area and theoretical plates and finally the method was found to be robust.

LOD and LOQ

The LOD can be defined as the smallest level of analyte that gives a measurable response and LOQ was determined as the lowest amount of analyte that was reproducibly quantified. These two parameters were calculated using the formula based on the SD of the response and the slope. LOD and LOQ were calculated by using equations,

$$\text{LOD} = 3.3 \times s/S \text{ and}$$

$$\text{LOQ} = 10 \times s/S,$$

Where s = Standard deviation, S = Slope of the calibration curve.

Assay of MET and EMPA in tablet

Assay of marketed product was carried out using the developed method. Sample solutions were prepared and injected into reversed-phase (RP)-HPLC system. The sample solution was scanned at 227 nm. The percentage drug estimated was found to be 101.12 for MET and 100.10 for EMPA. The chromatogram showed two single peaks of MET and EMPA was observed with retention times of 2.528 and 4.140 minutes Fig. 6.

Forced degradation studies

Stress studies are performed according to ICH guidelines under conditions of peroxide, acid, alkali, dry heat and UV studies.

Peroxide degradation studies

To 1 mL of stock solution of MET and EMPA, 1 mL of 20% hydrogen peroxide (H_2O_2) was added separately. The solutions were kept for 30 minutes at 60°C. For HPLC study, the resultant solution was diluted to obtain 850 $\mu\text{g}/\text{mL}$ and 5 $\mu\text{g}/\text{mL}$ solution and 10 μl were injected into the system and the chromatograms were recorded to assess the stability of the sample.

Acid degradation studies

To 1 mL of stock solution MET and EMPA, 1 mL of 2 N hydrochloric acid was added and refluxed for 30 minutes at 60°C. The resultant solution was diluted to obtain 850 $\mu\text{g}/\text{mL}$ and 5 $\mu\text{g}/\text{mL}$ solution and 10 μl solutions were injected into the system and the chromatograms were recorded to assess the stability of the sample.

Alkali degradation studies

To 1 mL of stock solution MET and EMPA, 1 mL of 2N sodium hydroxide was added and refluxed for 30 minutes at 60°C. The resultant solution was diluted to obtain 850 $\mu\text{g}/\text{mL}$ and 5 $\mu\text{g}/\text{mL}$ solution and 10 μl were injected into the system and the chromatograms were recorded to assess the stability of the sample.

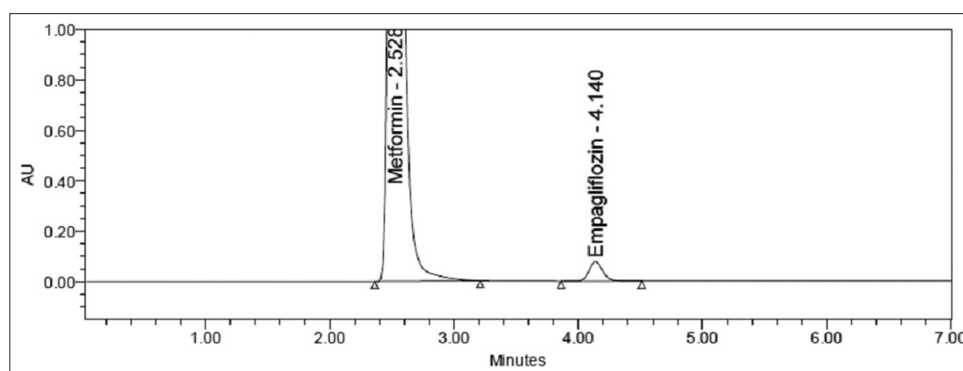


Fig. 5: Standard chromatogram of metformin and empagliflozin

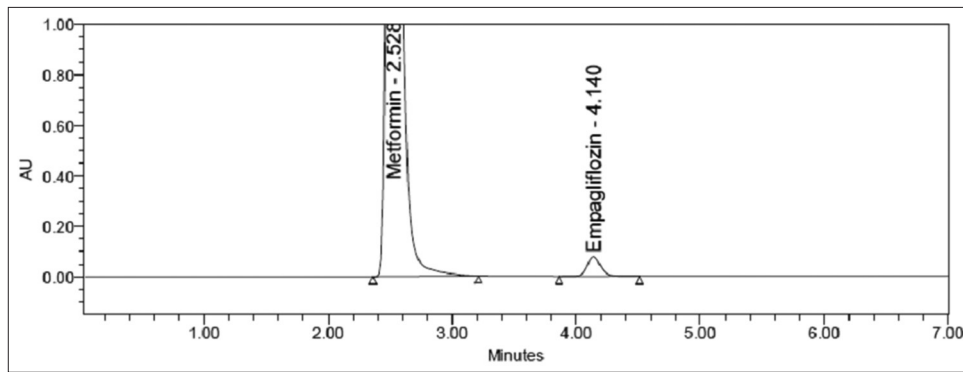


Fig. 6: A typical chromatogram of metformin and empagliflozin in tablet dosage form

Dry heat degradation studies

The standard drug solution was placed in oven at 105°C for 6 hrs to study dry heat degradation. For HPLC study, the resultant solution was diluted to 850 µg/mL and 5 µg/mL solution and 10 µl were injected into the system and the chromatograms were recorded to assess the stability of the sample.

UV degradation studies

The photochemical stability of the drug was also studied by exposing the 850 µg/mL and 5 µg/mL solution to UV light by keeping the beaker in UV chamber for 7 days or 200 Watt hrs/m² in photostability chamber. For HPLC study, the resultant solution was diluted to obtain 850 µg/mL and 5 µg/mL solutions and 10 µl were injected into the system, and the chromatograms were recorded to assess the stability of sample.

RESULTS AND DISCUSSION

From this study, it was found that a simple, precise, accurate, sensitive, and efficient stability indicating RP-HPLC method has been developed and validated for simultaneous estimation of MET and EMPA in bulk and pharmaceutical dosage form. To establish and validate an efficient method for analysis of these drugs in pharmaceutical formulations, optimized chromatographic conditions used for the estimation of MET and EMPA are given as in Table 1.

Linearity was established by analyzing different concentrations of MET and EMPA. The calibration curve was plotted with the area obtained versus concentration of both MET and EMPA Figs. 7 and 8. In the present study, six concentrations were chosen ranging between 10 and 60 µg/mL of MET and 30-50 µg/mL of EMPA.

The regression equation and correlation coefficient for MET and EMPA was found to be:

$y=9338x + 641.8$ and $R^2=0.9990$ and

$y=48884x + 1054$ and $R^2=0.9990$, respectively, and results were given in Table 2.

The accuracy of the method was proved by performing recovery studies on the commercial formulation at 50, 100 and 150% level. Percentage recoveries of MET and EMPA ranges from 100.01% to 100.10% in simultaneous equation method, and the results were shown in Table 3.

Precision was evaluated by a known concentration of MET and EMPA was injected six times and corresponding peaks were recorded and percentage RSD was calculated and found within the limits. The low percentage RSD value was indicated that the method was precise and reproducible and the results were shown in Table 4.

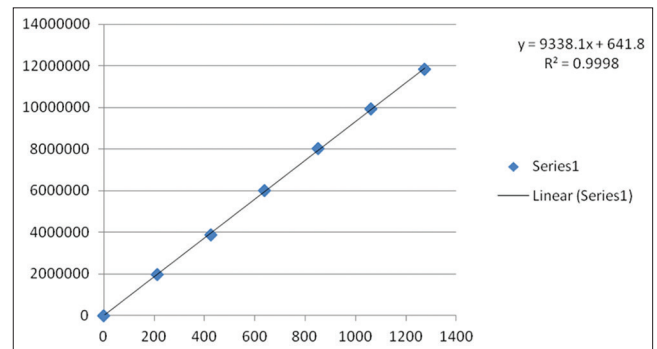


Fig. 7: Linearity curve of metformin

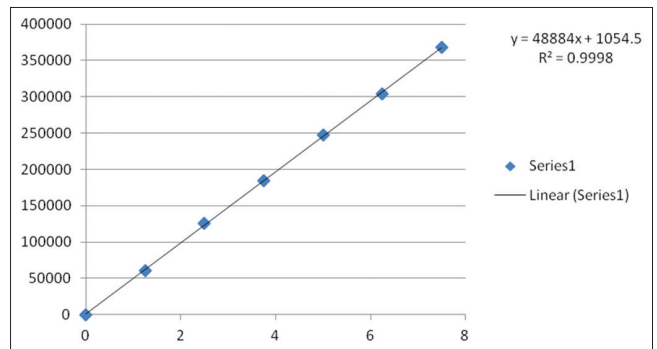


Fig. 8: Linearity curve of empagliflozin

Table 1: Optimized chromatographic conditions

Parameter	Condition
Mobile phase	Phosphate buffer: Acetonitrile: Methanol (15:80:5%V/V/V) pH adjusted to 4.8
Column	ODS C18, 250×4.6 mm, 5µ
Wave length	227 nm
Flow rate	1.0 mL/min
Injection volume	10 µL
Run time	10 min
Diluent	Water: Methanol (50:50)

ODS: Octadecyl silica

Robustness of the method is the ability of the method to remain unaffected by small deliberate changes in parameters like flow rate, mobile phase composition and column temperature. To study the effect of flow rate of the mobile phase, it was changed to 0.1 units from 1.0 mL to 0.9 mL and 1.1 mL. The effect of column temperature also checked by changing temperature to ±5°C. This deliberate change in the above parameters has no significant effect on the chromatographic behavior

Table 2: Linearity data for MET and EMPA

Serial number	MET		EMPA	
	Concentration (µg/ml)	Peak area	Concentration (µg/ml)	Peak area
1	212.5	1,978,270	1.25	60,899
2	425	3,888,258	2.5	125,922
3	637.5	6,025,370	3.75	184,711
4	850	6,025,370	5	247,276
5	1062.5	8,010,015	6.25	303,966
6	1275	11,848,754	7.5	367,820

MET: Metformin, EMPA: Empagliflozin

Table 3: Accuracy data for MET and EMPA

Drug	Spiked level (%)	Percentage recovery	Percentage RSD
MET	50	100.64	1.41
	100	100.89	0.41
	150	100.83	0.96
EMPA	50	101.34	0.26
	100	99.95	0.41
	150	100.34	0.96

RSD: Relative standard deviation, MET: Metformin, EMPA: Empagliflozin

Table 4: Precision method of proposed RP-HPLC method

Drug	Mean area	Percentage RSD
MET	7,993,232	0.59
EMPA	245,611	1.3

RSD: Relative standard deviation, RP-HPLC: Reversed phase high-pressure liquid chromatography, MET: Metformin, EMPA: Empagliflozin

Table 5: Robustness data for MET and EMPA

Parameters	Changed condition	Mean peak area		USP plate count	
		MET	EMPA	MET	EMPA
Flow rate (mL/minute)	0.9ml	25,943,381	625,201	2711.17	12,942.5
	1.0ml	7,984,114	245,123	2867.83	9433
	1.1ml	23,407,466	570,391	2632.33	13,144.5
Temperature (±5)	25°C	26,407,278	624,757	2402.17	13,202.2
	30°C	7,984,114	245,123	2867.83	9433
	35°C	23,507,485	566,202	2628	13,219.5
Mobile phase (±5%)	10:85:45%v/v	26,295,148	624,135	2405.33	13,209.3
	15:80:5%v/v	7,984,114	245,123	2867.83	9433
	20:75:5%v/v	24,446,476	598,229	2774.67	9406.67

MET: Metformin, EMPA: Empagliflozin

of the samples and results were given in Table 5 and the chromatograms are shown in the Figs. 9-14.

LOD and LOQ of MET and EMPA were evaluated based on RSD of the response and slope of the calibration curve. The detection limits were found to be 0.10 µg/mL and 0.01 µg/mL for MET and EMPA, respectively. The quantitation limits were found to be 0.31 µg/mL and 0.03 µg/mL for MET and EMPA, respectively. The results were given in Table 6. Moreover, the chromatograms are shown in Figs. 15 and 16.

Forced degradation studies were performed to establish the stability indicating property and specificity of the proposed method. Degradation studies were carried out under conditions of hydrolysis, dry heat, oxidation, UV light and photolysis and the drug substances were degraded in all conditions.

Acid and base hydrolysis was performed by exposing the drug substances with 2 N HCl and 2 N NaOH at 60°C for 30 minutes and it was showed degradation of MET and EMPA with degraded products peak at retention time 2.61 minutes and 3.02 minutes, respectively.

Degradation studies under oxidative conditions were performed by heating the drug sample with 20% H₂O₂ at 60°C and degraded product peaks were observed. Both MET and EMPA are sensitive to acid and alkali and there was no degradation occurs under UV light and thermal conditions. The results of forced degradation studies were given in Table 7 and chromatograms are shown in Figs. 17-21.

CONCLUSION

A new stability - indicating RP-HPLC method has been developed for estimation of MET and EMPA in bulk and pharmaceutical dosage form. The developed method was validated and it was found to be simple, sensitive, precise, and robust and it can be used for the

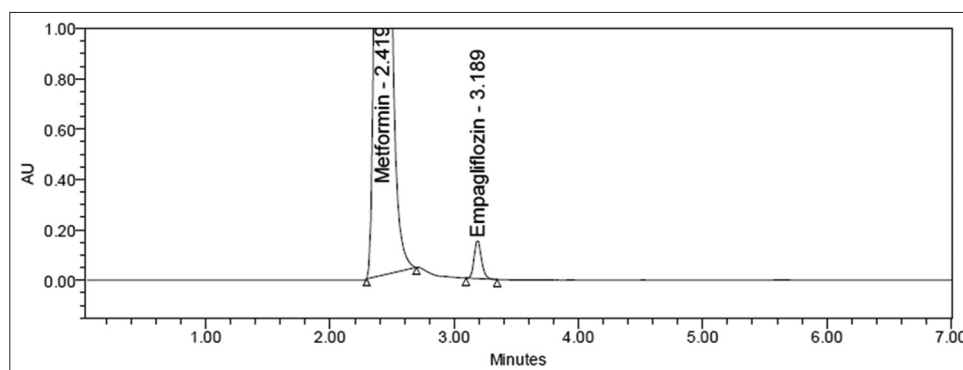


Fig. 9: Chromatogram of flow minus

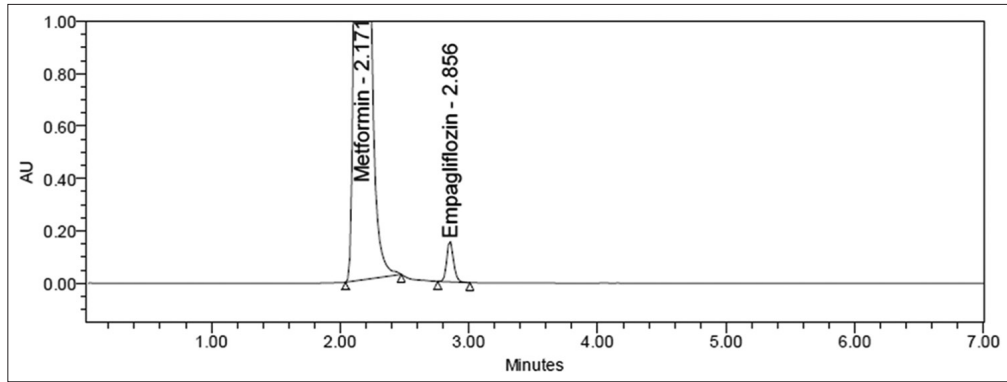


Fig. 10: Chromatogram of flow plus

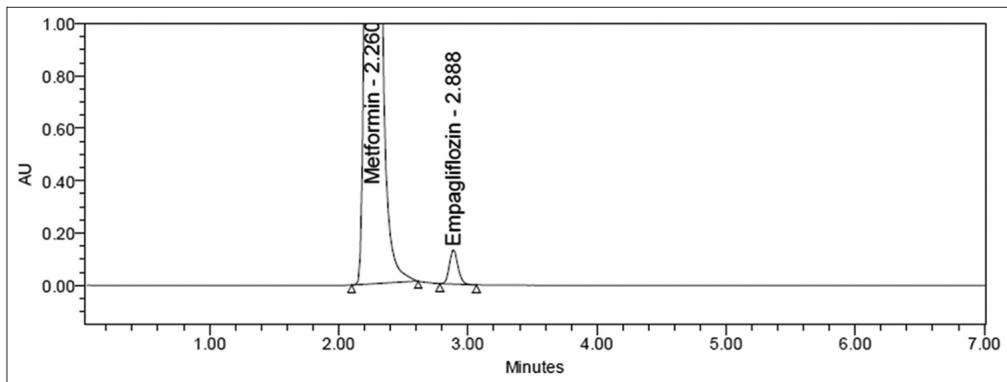


Fig. 11: Chromatogram of mobile-phase minus

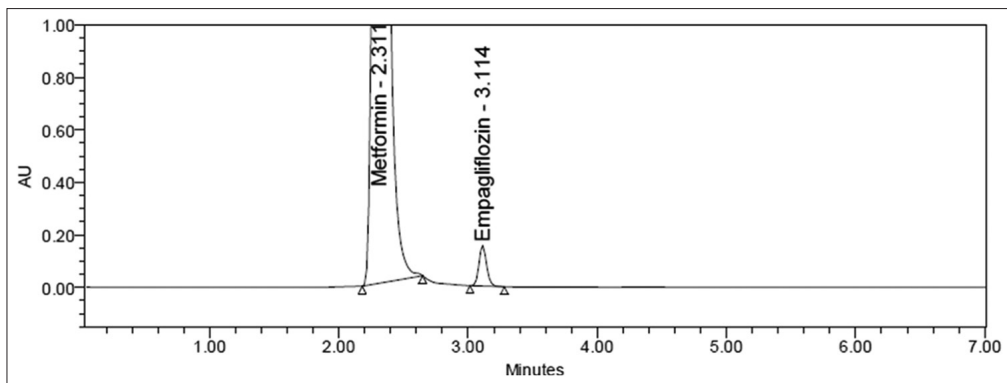


Fig. 12: Chromatogram of mobile-phase plus

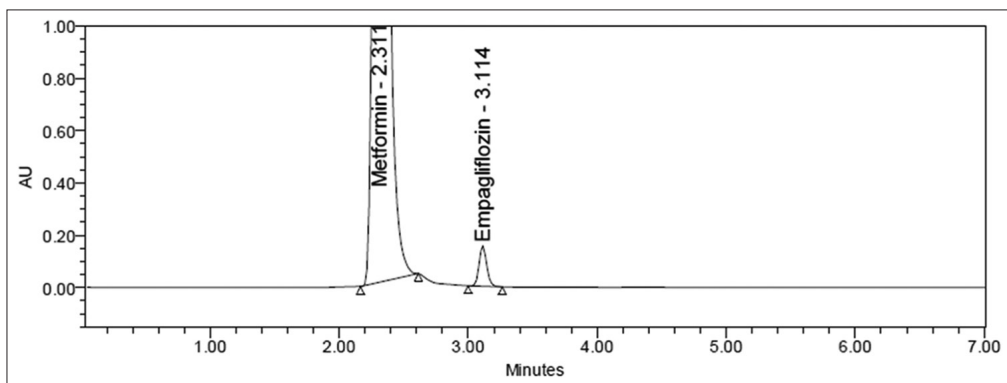


Fig. 13: Chromatogram of temperature minus

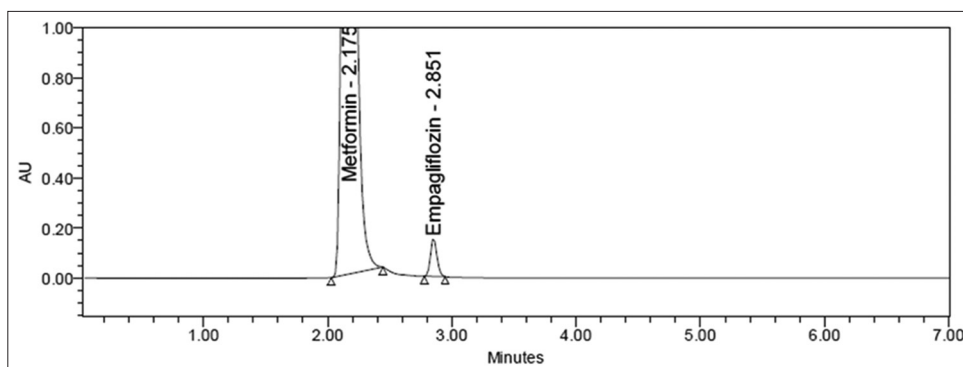


Fig. 14: Chromatogram of temperature plus

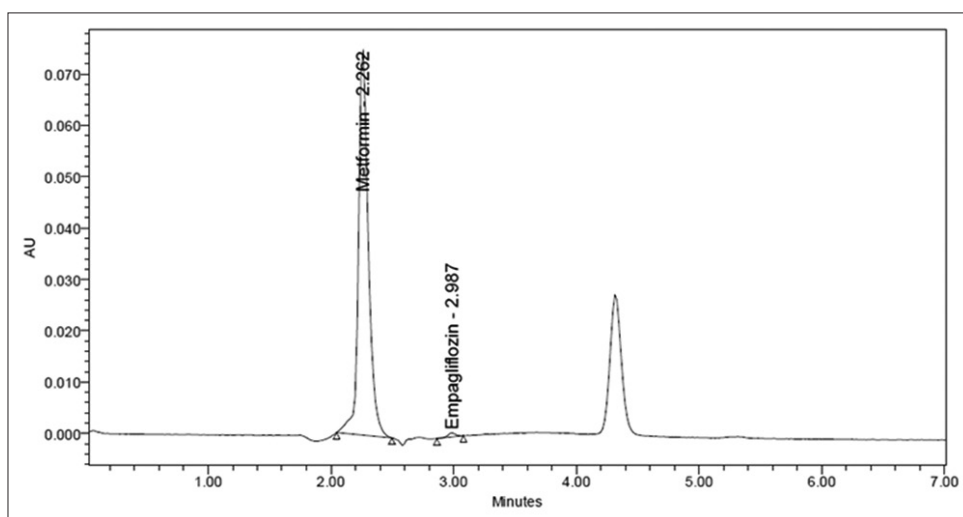


Fig. 15: Chromatogram of limit of detection

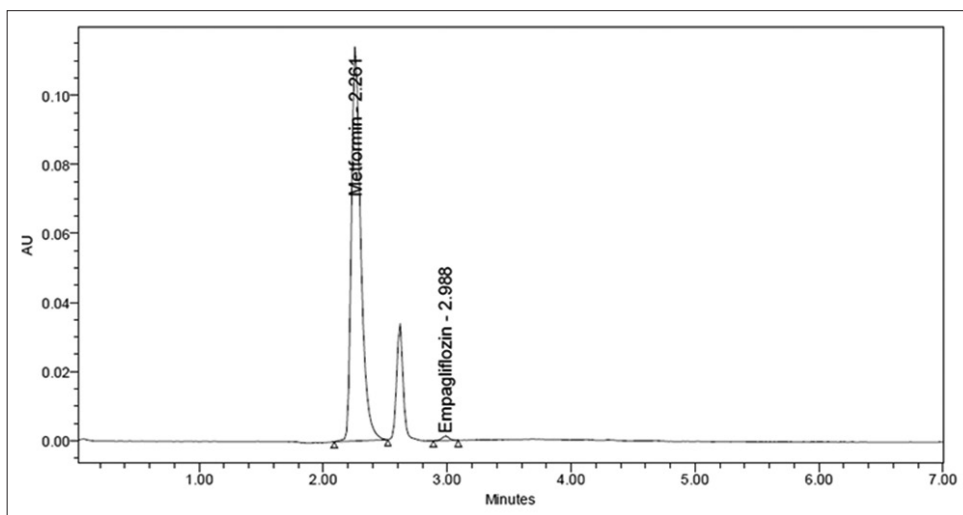


Fig. 16: Chromatogram of limit of quantification

Table 6: LOD and LOQ data for MET and EMPA

Drug	LOD ($\mu\text{g/ml}$)	LOQ ($\mu\text{g/ml}$)
MET	0.10	0.31
EMPA	0.1	0.03

LOD: Limit of detection, LOQ: Limit of quantification, MET: Metformin, EMPA: Empagliflozin

routine analysis of MET and EMPA in both bulk and pharmaceutical dosage forms. The forced degradation studies were carried out in accordance with ICH guidelines, and the results revealed suitability of the method to study stability of MET and EMPA under various degradation conditions such as acid, base, oxidative, thermal, UV and photolytic degradations. Finally, it was concluded that the method is simple, sensitive and has the ability to separate the

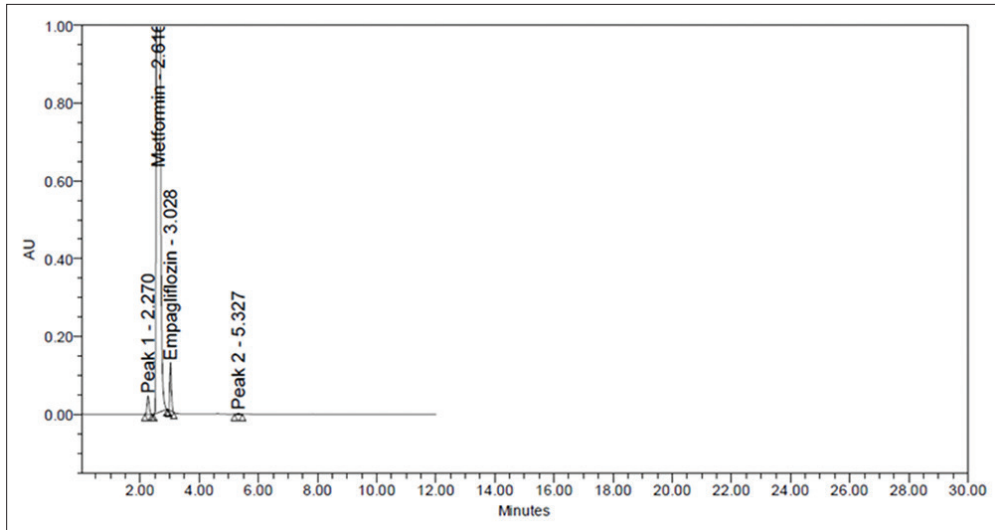


Fig. 17: Chromatogram of acid degradation

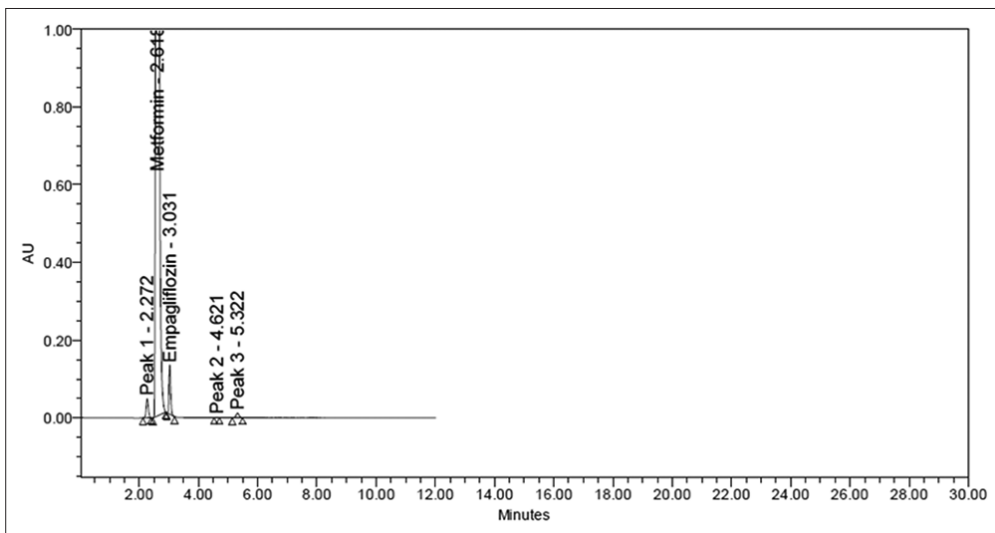


Fig. 18: Chromatogram of alkali degradation

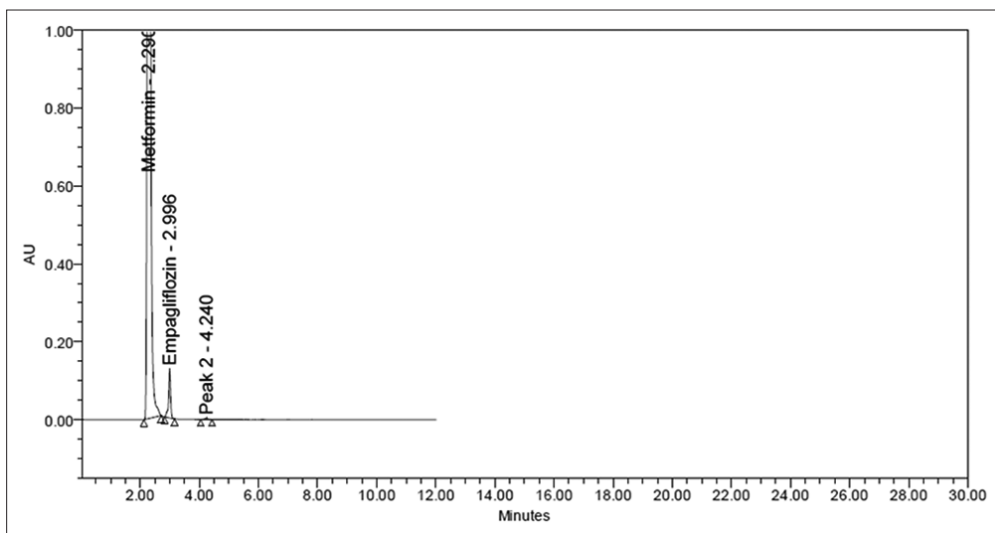


Fig. 19: Chromatogram of peroxide degradation

Table 7: Results of forced degradation studies

Serial number	Injection	MET		EMPA	
		Percentage assay	Percentage degradation	Percentage assay	Percentage degradation
1	Acid degradation	95.28	2.61	95.60	3.02
2	Base degradation	96.54	2.61	96.15	3.02
3	Peroxide	97.79	2.29	97.85	2.90
4	Thermal degradation	97.90	2.28	98.60	3.0
5	UV degradation	98.16	2.28	98.60	3.0
6	Neutral degradation	99.01	2.29	99.82	3.0

MET: Metformin, EMPA: Empagliflozin, UV: Ultraviolet

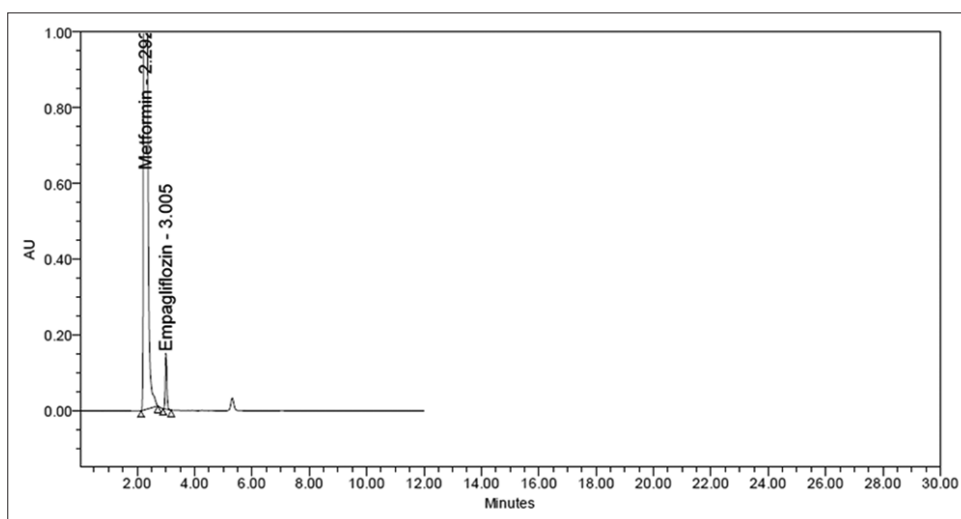


Fig. 20: Chromatogram of thermal degradation

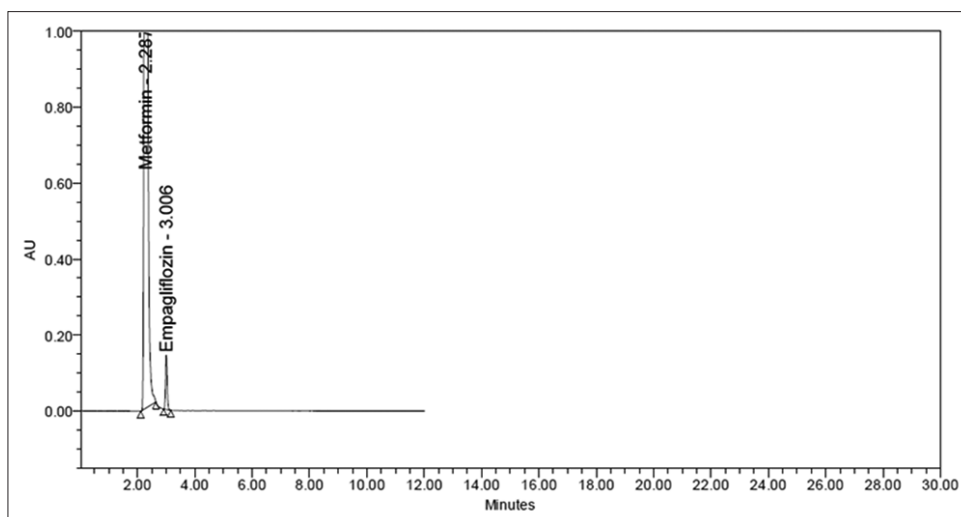


Fig. 21: Chromatogram of ultraviolet degradation

drug from degradation products and excipients found in the dosage form.

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