

## ANTIOXIDANT ACTIVITY OF *TARENNA POLYCARPA* (MIQ.) KOORD. LEAF

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### ABSTRACT

**Objective:** The objective of this study is to evaluate the antioxidant activity of *Tarenna polycarpa* (Miq.) koord. Ex Valenton. Leaf extract and fractions.

**Methods:** Antioxidant activity was examined by DPPH method.

**Results:** Ethanol extract, n-hexane fraction, and ethyl acetate fraction with DPPH assay measured as half maximal inhibitory concentration were 55.21, 109.73, and 42.04 µg/mL, respectively.

**Conclusions:** The results reveal that *T. polycarpa* extract and fractions have strong antioxidant potential. Our further study is to isolate compounds responsible for antioxidant components.

**Keywords:** Antioxidant, *Tarenna polycarpa*, Leaf, Extract, Fractions.

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### INTRODUCTION

Oxidation is one of important processes in living organisms. Free radicals are increasing from metabolism or environmental sources interact continuously with biological system [1,2]. Reactive species are molecules or atoms which have an electronic instability and most reactive [3]. The uncontrolled production of oxygen free radicals and the unbalanced mechanism of antioxidant protection result in the onset of many diseases such as cancer, diabetes, Alzheimer's, heart diseases, and aging [2,4-6].

*Tarenna polycarpa* (Miq.) Koord. Ex Valenton. (morbesi-besi) has been used as immunostimulator in Tapanuli and Sibolga [7]. *Tarenna asiatica* has the same genus with *T. polycarpa* which contained of flavonoid as antioxidant and antimicrobial [8,9]. The aim of this study was to determine antioxidant activities of *T. polycarpa* (Miq.) Koord. Ex Valenton leaves.

### MATERIALS AND METHODS

#### Plant and chemicals material

Fresh leaves of *T. polycarpa* (Miq.) were collected from Sibolga Sambas, Sibolga City, Sumatera Utara province, Indonesia. *T. polycarpa* (Miq.) was identified in Research Center for Biology, Indonesian Institute of Science, Bogor, and the voucher specimen was deposited in herbarium. Chemicals used were distilled water, 1,1-diphenyl-2-picrylhydrazyl (DPPH) (Sigma), methanol (Merck).

#### Preparation of ethanol extract (EE)

The air-dried and powdered leaves of *T. polycarpa* (Miq.) (1 kg) were extracted by cold maceration with ethanol 80% (7.5 L) for 5 days with soaked, filtrate was collected, and residue was washed with 2.5 L of ethanol 80%, and filtrate was collected in the same bottle and then evaporated under reduced pressure to give a viscous extract and then freeze-dried to dry [10-13].

#### Preparation of n-hexane (NHF) and ethyl acetate fraction (EAF)

The EE was added with 100 mL aquadest to yield liquid form of ethanolic extract. The extract was fractioned with NHF and ethyl

acetate, respectively, and fractions were evaporated under reduced pressure to give a dry fraction [14].

#### Free radical scavenging activity test

The free radical scavenging activity was measured by 1,1-diphenyl-2-picrylhydrazyl (DPPH) method of Blois (1958). 0.2 mM solution of DPPH in methanol was prepared, and 100 µl of this solution (EE, NHF, and EAF) was added to various concentrations; 25, 50, 75, and 100 µg/ml. After 60 min absorbance was measured at 516 nm. All the tests were performed in triplicate and percentage of inhibition was calculated by comparing the absorbance values of the control and test samples [1,2,4,15].

$$\text{Percentage of inhibition} = \frac{\text{Abs control} - \text{Abs test}}{\text{Abs control}} \times 100\%$$

#### Statistical analysis

Data were expressed as a mean ± standard deviation. Analysis of variance (ANOVA) with the Tukey *post hoc* test was used for multiple comparison. All statistics were analyzed using the SPSS 20 software.

### RESULTS AND DISCUSSION

#### Antiradical activity

Antiradical power of the plant samples was measured in terms of hydrogen-donating ability using DPPH which is a stable, nitrogen-centered free radical and produces deep purple color in methanol solution. Antioxidants either transfer an electron or a hydrogen atom to DPPH, thus neutralizing its free radical character [16]. DPPH test, which is based on the ability of DPPH, a stable free radical, to decolorize in the presence of antioxidants, is a direct and reliable method for determining radical scavenging action [17]. The DPPH assay has been largely used as a quick, reliable, and reproducible parameter to search the *in vitro* general antioxidant activity of pure compounds as well as plant extracts [18]. The reducing capacity of compounds could serve as an indicator of potential antioxidant property [19]. It is very

important to point out that a low half maximal inhibitory concentration ( $IC_{50}$ ) value reflects a high antioxidant activity of the fraction since the concentration necessary to inhibit the radical oxidation in 50% is low.  $IC_{50}$  for EE, NHF, and EAF in DPPH assay was 55.21, 109.73, and 42.04  $\mu\text{g/mL}$ , respectively.

#### AUTHORS CONTRIBUTION

RPS : Doing all the research. MN: Editing manuscript. Ros: Preparing manuscript and analysis data.

#### CONFLICT OF INTEREST

There is no conflict of interest to be reported.

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