

**PHYTOCHEMICAL, GAS CHROMATOGRAPHY-MASS SPECTROMETRY PROFILING AND FREE RADICAL SCAVENGING ACTIVITY OF *EXOCARPOS LONGIFOLIUS* (SANTALACEAE) EXTRACTS**

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Received: 14 March 2018, Revised and Accepted: 04 July 2018

**ABSTRACT**

**Objective:** The objectives of the study were to investigate phytochemical screening and gas chromatography-mass spectrometry (GC-MS) profiling of twigs and leaves of *Exocarpos longifolius* and its 2,2-diphenyl-1-picrylhydrazyl free radical scavenging activity.

**Methods:** Phytochemical screening covering saponins, alkaloids, terpenoids, flavonoids, and tannins were carried out by standard methods. The free radical scavenging activity was performed by thin-layer chromatography - bioautography, followed by determining the IC<sub>50</sub> values of active extracts. Chemical compounds of active extract were examined by gas chromatography/mass spectrometry.

**Results:** The phytochemical screening revealed that all extracts have been shown to contain tannins and terpenoids, saponins were found in the methanol extract of twigs, and alkaloids were found in the ethyl acetate extracts of twigs and leaves, while flavonoids were found in the methanol extract and ethyl acetate extracts of twigs and leaves of *E. longifolius*. Ethyl acetate extract of twigs showed the best of IC<sub>50</sub> value (15.65 µg/mL), while ethyl acetate extract of leaves and methanol extract of twigs have strong free radical scavenging activity with IC<sub>50</sub> values were 78.59 and 67.24 µg/mL, respectively. The GC-MS analysis of ethyl acetate extract of twigs has shown the presence of 12 identified compounds, and the main chemical compound is Stigmast-4-en-3-one (21.91%).

**Conclusions:** The ethyl acetate extract of *E. longifolius* showed a very powerful free radical scavenging activity and revealed chemical compounds had been known to have biological activity. It might be used as a natural antioxidant.

**Keywords:** *Exocarpos longifolius*, Phytochemical compounds, 2,2-diphenyl-1-picrylhydrazyl, Stigmast-4-en-3-one.

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**INTRODUCTION**

Plants have been used for such a long time in various aspects of human life, including for their primary health care. Utilization of plants as herbal medicine is related to their biologically active compounds, because plants are capable of producing varieties of chemical compounds categorized as secondary metabolites. Kong *et al.* [1] indicated that plants are rich source of secondary metabolites, such as alkaloids, flavonoids, saponins, steroids, terpenoids, polysaccharides, and tannins. Many of these secondary metabolites have pharmacological and biological activities [2,3] such as free radical scavengers or antioxidant.

Free radicals (superoxide and hydroxyl radicals) are produced in the body during metabolic processes, and in a state of excess will lead to negative effects, including cell death and tissue damage [4], atherosclerosis, coronary heart disease, cancer, and premature aging [5]. On the other hand, the body does not have large amounts of antioxidant reserves, so that the body needs exogenous antioxidants to protect it more efficiently against oxidative stress caused by free radical [6].

Antioxidants are components that can donate an electron to a free radical, and neutralize it [7]. The exogenous antioxidant can be divided into synthetic and natural antioxidants. Natural antioxidants have advantages over synthetic antioxidants which can be obtained easily and economically and have a slight or negligible side effect [8]. Recently, the interests in natural products and environment-friendly antioxidants are looking for as a substitute and various plants contained a massive pool of bioactive compounds [9].

Some species of Santalaceae have been reported to have biological activities. *Exocarpos latifolius* has yielded exocarpic acid and its analogs

as antimycobacterial compounds [10]. Some species of Santalaceae have been reported as potential free radical scavengers such as *Viscum album*, *Osyris quadripartita*, and *Quinchamali chilensis* which correlated with its total phenol [8,11,12].

*Exocarpos longifolius* is a small tree or shrub belongs to Santalaceae. *E. longifolius* distributed in Southeast Asia, Australia, and the Pacific Islands. Several plant species from Santalaceae have been known to be potential as an antioxidant. However, the biological potential of *Exocarpos* as free radical scavenger has not been studied widely. Therefore, this study focused on the chemical compounds and the free radical scavenging activity of various extracts of *E. longifolius*.

**METHODS****Plant material**

*Exocarpos longifolius* or Kayu Sulaeman (common name) was collected from Batudulang, Batu Lanteh District, West Sumbawa, West Nusa Tenggara. The plant was identified and authenticated at the Herbarium Bogoriense, Botany Division, Research Center for Biology, Indonesian Institute of Sciences.

**Sample preparation and extraction**

Twigs and leaves were separated, washed thoroughly under tap water to remove dirt. The samples were cut into small pieces and dried in the oven at 40°C. The dried samples were ground into powder.

Samples were subsequently extracted with different solvent polarity, namely hexane, chloroform, ethyl acetate, and methanol successively. Samples were macerated with hexane thrice. The filtrate was concentrated with the rotary evaporator (Heidolph WB2000). The

same steps follow for other solvents. Dried crude extract weighted and extraction yield were calculated as follows: (Weight of extract/weight of original sample)×100%

#### Phytochemical analysis

Qualitative phytochemical screening was carried out on ether, chloroform, ethyl acetate, and methanol extract of *E. longifolius* for the identification of major phytochemical constituents such as saponins, tannins, alkaloids, terpenoids, and flavonoids [13].

#### Saponins

The extract is diluted with distilled water and shaken vigorously. The appearance of persistent froth for 1 min indicates the presence of saponins.

#### Tannins

The extract was added with distilled water and then added with ferric chloride (FeCl<sub>3</sub>). The appearance of a dark green color indicates the presence of tannins.

#### Alkaloids

The extract was added with 2% sulfuric acid, stand for 2 min. A few drops of Dragendroff's reagent were added. The appearance of orange precipitate indicates the presence of alkaloids.

#### Terpenoids

The extract was added with chloroform and added with concentrated sulfuric acid from the side of the test tube. The appearance of the reddish brown rings at the junction of two layers indicates the presence of terpenoids.

#### Flavonoids

The extract added with methanol and heated at 50°C. After that, it added with magnesium and 4 drops of hydrochloric acid. Appearance of orange-red color indicates the presence of flavonoids

#### Bioautographic analysis for 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging activity

Ten microliters of hexane, chloroform, ethyl acetate, and methanol extracts (10 mg/ml) and (+)-catechin as standard were transferred on thin-layer chromatography (TLC) plate (Silica gel GF<sub>254</sub>). The TLC plate was air-dried and then sprayed with 0.2% DPPH in methanol [14]. Observations of DPPH free radical scavenging activity were examined at 30 min after DPPH spraying. The yellowish-white spot around the extract against a purple background on TLC-plate indicating DPPH free radical scavenging activity.

Active extracts with free radical scavenging activity were further analyzed. 10 µL of ethyl acetate or chloroform extract was transferred on TLC plate, then developed in mobile phase dichloromethane: methanol (10:1) to separate the chemical compounds. The active methanol extract was developed in chloroform: methanol:water (6:4:1). The TLC plates were subsequently dried then sprayed with 0.2% DPPH in methanol. The plates were incubated at room temperature for 30 min. The chemical

compounds with DPPH free radical scavenging activity were detected as yellow spots against a purple background.

#### Determination of IC<sub>50</sub>

The extracts with DPPH free radical scavenging activity were further analyzed for their IC<sub>50</sub> values by serial dilution method in a 96 well microplate according to the modified method recommended by Zhou *et al.* [15]. The test was conducted in triplicates. The working solution of extract was prepared at a concentration of 1000 µg/ml in methanol. Catechin was used as positive control and prepared at a concentration of 1000 µg/ml. The DPPH solution in methanol (0.1 mmol) was freshly prepared. Each well within the first row (row A) was added with 100 µL of working solution (1000 µg/ml). The following rows were added with 50 µL methanol. Then, 50 µL of solution in the first row was taken out and transferred into the second row and homogenized. After homogenized, 50 µL was taken out and transferred into the next row resulting in serial dilution. In the past row, 50 µL was taken out and discarded. After the serial dilution is completed, each well was added with 80 µL 0.1 mmol DPPH. The microplate was incubated for 30 min at 37°C in the dark condition. After the incubation period was completed, the absorbance of each sample was measured at 517 nm (Varioskan Flash, Thermo Scientific). The value of IC<sub>50</sub> was calculated using a linear curve between the concentration of the test sample and the percentage of inhibition.

#### Gas chromatography-mass spectrometry (GC-MS) analysis

The chemical compounds of ethyl acetate extract of twigs were identified by GC-MS analysis. It was carried out using a GC-MS (Shimadzu 2010), equipped with Rtx-5MS capillary column (5% diphenyl/95% dimethylpolysiloxane, 30 m, 0.25 mm ID, 0.25µm df, max. 350°C, RESTEK, USA). The carrier gas was helium at a flow rate of 2.0 ml/min. The volume of the injected sample was 5 µL. The injector temperature was 250°C, and the interface temperature was 300°C. The initial temperature of the column was 100°C, and the final temperature was 270°C. The temperature program: Initial temperature 100°C, held for 3 min and then ramping at the rate of 10°C/min up to 270°C and held for 18 min [16]. Identification of chemical components in the extract based on interpretation on mass spectrum of GC - MS using the database of NIST 11 (National Institute Standard and Technology,US) and WILEY 8.

## RESULTS

#### Qualitative phytochemical analysis

The qualitative phytochemical analysis was carried out on different solvent from leaves and twigs of *E. longifolius* (Table 1). Phytochemical screening revealed similar chemical compounds of the leaves and the twigs of *E. longifolius*. Tannins and terpenoids were present in the hexane and chloroform extracts, while ethyl acetate extracts revealed the presence of tannins, alkaloids, terpenoids, and flavonoids (Table 1).

#### Detection of DPPH free radical scavenging activity

The results of DPPH free radical scavenging activity assay of *E. longifolius* twigs and leaves were presented in Fig. 1.

Fig. 1 showed that all of the extracts had active compounds as DPPH free radical scavenger indicated by a yellowish white band. The active

Table 1: Qualitative phytochemical analysis of twigs and leaves of *E. longifolius*

Compounds	Twigs				Leaves		
	Hexane	Chloroform	Ethyl acetate	Methanol	Hexane	Chloroform	Ethyl acetate
Saponin	-	-	-	+	-	-	-
Tannin	+	+	+	+	+	+	+
Alkaloid	-	-	+	-	-	-	+
Terpenoids	+	+	+	+	+	+	+
Flavonoids	-	-	++	+	-	-	++

-: Secondary metabolites not detected, +: Secondary metabolites detected. *E. longifolius*: *Exocarpos longifolius*

extracts were further analyzed to determine their potential as free radical scavengers.  $IC_{50}$  values of extracts were presented in Table 2. Three extracts exhibited *in vitro* strong free radical scavengers ( $IC_{50}$  value <100  $\mu$ g/ml) [17], which were ethyl acetate and methanol extracts of the twigs and ethyl acetate extract of the leaves.

#### GC-MS profiling of ethyl acetate extract of twigs

The GC-MS analysis of ethyl acetate extract of the twigs *E. longifolius* was shown in Fig. 2.

GC-MS analysis of ethyl acetate extract of *E. longifolius* of twigs revealed the presence of 12 identified compounds and stigmast-4-en-3-one (21.91%) as the main compound. The identified compounds could be classified into various chemical classes such as fatty acids (methyl palmitate; linoleic acid, methyl ester; alpha linoleic acid, methyl ester; and sebacic acid, ethyl methyl ester); steroids (1-dehydrotestosterone, methyloxime, trimethylsilyl ether; Androst-11-en-17-one, 3-formyloxy-, (3.alpha.,5.alpha.)-; and stigmast-4-en-3-one); acyclic diterpene (phytol, acetate); alcohol (tetracontane-1,40-diol); benzofuran derivative (2-isopropenyl-4,4,7a-trimethyl-2,4,5,6,7,7a-hexahydro-benzofuran-6-ol); acetamide derivative (N1-(8-Methyl-8-azabicyclo[3.2.1]oct-3-yl)-2-(2-methoxyphenyl)acetamide); and thiophene derivative (4-Acetyloxyimino-6,6-dimethyl-3-methylsulfanyl-4,5,6,7-tetrahydro-benzo[c]thiophene-1-carboxylic acid methyl ester). Some of these compounds have been reported by other researchers to have biological activity (Tables 3 and 4).

#### DISCUSSIONS

Phytochemical screening is important to ascertain the chemical components in the extract and also as the first step in predicting the potential of active compounds in plant [24]. Priyanga *et al.* and Sowmya *et al.* indicated that the chemical compounds such as alkaloids, tannins, flavonoids, steroids, terpenoids, and phenolic in plants are natural bioactives [25,26]. Saponins are sedimenting and thickened red blood cells, may bind cholesterol [27], and overcome inflammation [28]. Alkaloids might be used as an analgesic, antispasmodic, and antibacterial [29]. Terpenoids increase glutathione-S-transferase and apoptotic against cancer cells [30], antibacteria by weakening the cell membrane and resulting in the dissolution of microbial cell walls [28].

Table 2:  $IC_{50}$  of twigs and leaves *E. longifolius* extracts

No	Part of the plant	Solvent	$IC_{50}$
1	Twigs	Hexane	>385
2		Chloroform	>385
3		Ethyl acetate	15,65
4		Methanol	67,24
5	Leaves	Hexane	>385
6		Chloroform	>385
7		Ethyl acetate	78,59

*E. longifolius*: *Exocarpos longifolius*

Table 3: Identified chemical compounds of ethyl acetate twigs extract of *E. longifolius*

Retention time (min)	Area	Area percentage (%)	Compounds
14,18	56351	0.85	Phytol, acetate
15,11	107075	1.62	Methyl palmitate
16,79	86467	1.31	linoleic acid, methyl ester
16,86	261350	3.96	Alfa linoleic acid, methyl ester
18,20	57836	0.88	Sebacic acid, ethyl methyl ester
18,36	49588	0.75	2-Isopropenyl-4,4,7a-trimethyl-2,4,5,6,7,7a-hexahydro-benzofuran-6-ol
19,55	45983	0.70	1-Dehydrotestosterone, methyloxime, trimethylsilyl ether
19,93	56051	0.85	Androst-11-en-17-one, 3-formyloxy-, (3.alpha., 5.alpha.)-
20,55	160571	2.43	Tetracontane-1,40-diol
21,55	197451	2.99	N1-(8-Methyl-8-azabicyclo[3.2.1]oct-3-yl)-2-(2-methoxyphenyl) acetamide
21,58	49929	0.76	4-Acetyloxyimino-6,6-dimethyl-3-methylsulfanyl-4,5,6,7-tetrahydro-benzo[c]thiophene-1-carboxylic acid methyl ester
21,70	1449611	21.91	Stigmast-4-en-3-one

*E. longifolius*: *Exocarpos longifolius*

Flavonoids have many biological activities, including anti-inflammatory, antibacterial, antiallergic [31,32], antimicrobial, antioxidant, and cytotoxic properties [33].

DPPH free radical is commonly used for evaluation of antioxidant activity because it is a stable radical. DPPH free radical scavenging activity assay has been used widely for plant materials and pure compounds [34] because it's fast, simple [35] and the results are reliable [36]. In the presence of antioxidant compounds, DPPH (purple colored) will be reduced to  $\alpha,\alpha$ -diphenyl- $\beta$ -picryl hydrazine (yellow colored) [37]. The yellow color produced through reduction of DPPH against a purple background on the TLC [38]. The intensity of the color indicates the potential of DPPH free radical scavenging activity. The active extracts as DPPH free radical scavengers may be related to the flavonoids content

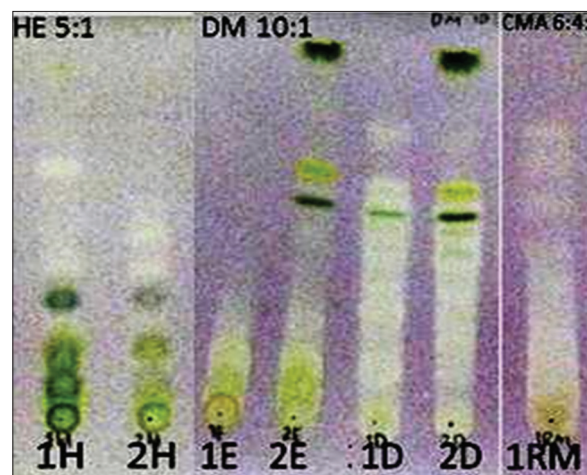


Fig. 1: Chromatogram of *Exocarpos longifolius* extracts sprayed with 0.2% 2,2-diphenyl-1-picrylhydrazyl (DPPH) in methanol.

The yellowish-white band shows the active compounds as DPPH free radical scavengers. (1: Twig, 2: Leaves, H: Hexane, D: Chloroform, E: Ethyl acetate, and RM: Methanol)

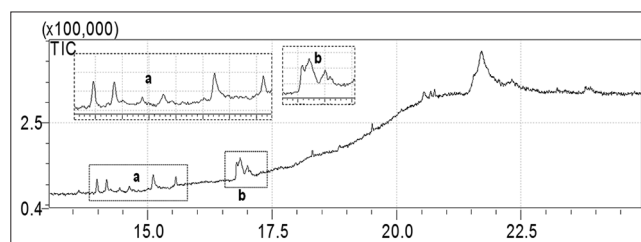


Fig. 2: Gas chromatography-mass spectrometry chromatogram of ethyl acetate extract of *Exocarpos longifolius* of twigs



Table 4: Bioactivity of compounds in *E. longifolius* ethyl acetate extract

No	Name of compound	Activity	Ref.
1	Phytol, acetate	Antimycobacterial	[18]
2	Methyl palmitate	Acaricidal activity	[19]
3	Linoleic acid, methyl ester	Increase Zinc availability, acidifier, inhibit production of uric acid	[20]
		Anticancer	[21]
		Anti-inflammatory, antihistamine, and anti-arthritis properties. It also possesses hepatoprotective and hypocholesterolemic properties	[22]
4	Alpha-linolenic acid, methyl ester	Increase zinc availability, acidifier, inhibit production of uric acid	[20]
5	Sebacic acid, ethyl methyl ster	Acidifier, inhibit production of uric acid	[20]
6	2-Isopropenyl-4,4,7a-trimethyl-2,4,5,6,7,7a-hexahydro-benzofuran-6-ol	Oligosaccharide provider	[20]
7	1-Dehydrotestosterone, methyloxime, trimethylsilyl ether	-	-
8	Androst-11-en-17-one, 3-formyloxy-, (3.alpha., 5.alpha.)-	Alpha-amylase inhibitor, alpha-glucosidase inhibitor	[20]
9	Tetracontane-1,40-diol	-	-
10	N1-(8-Methyl-8-azabicyclo[3.2.1]oct-3-yl)-2-(2-methoxyphenyl) acetamide	Antitumor, anaphylactic, inhibit production of tumor necrosis factor	[20]
11	4-Acetyloxyimino-6,6-dimethyl-3-methylsulfanyl-4,5,6,7-tetrahydro-benzo[c]thiophene-1-carboxylic acid methyl ester	Acidifier, inhibit production of uric acid	[20]
12	Stigmast-4-en-3-one	Hypoglycemic effect	[23]

*E. longifolius*: *Exocarpos longifolius*

in the extract. Flavonoids are able as a free radical scavenger [39], act as hydrogen donor [40] and inhibit the enzyme responsible for free radicals generation [41]. Tannins in the ethyl acetate extracts might also contribute to free radical scavenging activity and antibacterial activity [42]. The DPPH radical scavenger activity of the extract serves as an important indicator of its potential as an antioxidant. An antioxidant prevents the oxidation process of the molecule by protecting the key cellular components from free radicals [43].

The active extract was analyzed further to identify its chemical compounds by GC-MS analysis. The compounds from the extract are identified based on interpretation on mass spectrum of GC - MS using the database of National Institute Standard and Technology (NIST11) and WILEY 8. The twelve compounds were characterized and identified (Table 3).

## CONCLUSIONS

Based on the result, it can be concluded that the ethyl acetate extract of *E. longifolius* twigs had strong free radicals scavengers' activity, and the main compounds were Stigmast-4-en-3-one. Further study needs to be done to fractionate, isolate and elucidate the potential natural free radicals scavengers from *E. longifolius*.

## ACKNOWLEDGMENT

The authors thank Dra. Florentina Indah Windradi for her kind contribution of sample collection from West Nusa Tenggara.

## AUTHOR'S CONTRIBUTION

Praptiwi, designed the work and performed the laboratory analysis for extraction and bioassay, participated in drafting the article and contributed to the final manuscript. Ahmad Fathoni, performed the laboratory analysis in phytochemical analysis and GC-MS, performed for data analysis and interpretation, and contributed to the final manuscript.

## CONFLICTS OF INTEREST

The authors declare that they have no conflicts of interests.

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