

ANTIOXIDANT POTENTIAL OF PIPERIDINE CONTAINING COMPOUNDS-A SHORT REVIEW

MANJUSHA R K*, SHAHEEN BEGUM, ARIFA BEGUM, BHARATHI K

Department of Pharmaceutical Chemistry, Institute of Pharmaceutical Technology, Sri Padmavati Mahila Visvavidyalayam (Women's University), Tirupati 517502, Andhra Pradesh, India. Email: manjurk9393@gmail.com

Received: 09 April 2018, Revised and Accepted: 14 May 2018

ABSTRACT

Piperidine is a saturated heterocyclic ring, considered as a privileged scaffold in view of its role in wide range of biological activities. Piperidine is good candidate molecule for obtaining potent antioxidant agents. The planar nature of this heterocyclic nucleus allows the introduction of substituent groups at different positions on the ring. In the present review, the antioxidant profile of piperidine containing compounds has been focused. The compounds were classified into naturally occurring piperidines, unsaturated piperidines, N-substituted piperidines, piperamides, piperanols, piperidine oximes, and hydrazides.

Keywords: Piperidine nitroxides, Substituted piperidines, Unsaturated piperidines, N-acyl substituted piperidines, Diaryl substituted piperidinones, Piperidinone oximes, Piperidine hydrazides.

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INTRODUCTION

Piperidine is a saturated heterocyclic secondary amine which is associated with a diverse set of biological activities such as antimicrobial, anti-inflammatory, antiviral, antimalarial, general anesthetic, antidepressant, antioxidant, antiepileptic, antitumor, anticonvulsant, and antihyperlipidemic activities [1-8]. Several clinically available drug candidates also possess this moiety in their structure (Fig. 1) [9]. Antioxidants are the compounds, which in low concentrations, are capable of either delay or inhibit the oxidative process, which occurs under the influence of atmospheric or reactive oxygen species (ROS). Antioxidants help to prevent the deleterious effects of free radicals and have been proven to be effective in oxidative stress-related diseases such as tumors, ocular related diseases, inflammation, and atherosclerosis [10-12].

Assays used for evaluating antioxidant compounds can be classified into the one which is associated with lipid peroxidation's (thiobarbituric acid assay [TBA], beta-carotene bleaching, and conjugated diene assays) and assays which are associated with electron or radical scavenging (2,2-diphenyl-1-picrylhydrazyl (DPPH), 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS), hydrogen peroxide (H₂O₂) ferric reducing/antioxidant power (FRAP), ferric thiocyanate (FTC), and aldehyde/carboxylic acid assay) [13]. In this review article, the antioxidant activity of different piperidine containing compounds was discussed.

REVIEW ON ANTIOXIDANT AND RELATED BIOLOGICAL ACTIVITIES OF PIPERIDINE CONTAINING COMPOUNDS

Naturally occurring piperidine-based compounds

Piperine is piperidine containing an alkaloid, present in pepper extracts, i.e., *Piper nigrum* L. (Family Piperaceae), display strong antioxidant activity due to its ability to inhibit or quench free radicals (hydroxy and ROS). The medicinal value of piperine is very huge due to its antioxidant, antiplatelet, anti-inflammatory, antihypertensive, hepatoprotective, antithyroid, antitumor, and antiasthmatic activities. Nakatani *et al.* extracted various phenolic amides (Fig. 2) from pepper, which exhibited promising antioxidant properties in FTC and TBA assays. All the phenolic amides showed good antioxidant activity than α -tocopherol at 0.01% concentration. Significant antioxidant activity was observed for amide possessing buta-1, 3-dienyl-2-methoxyphenol [14].

Synthetic piperidines

The compounds were classified into piperidine nitroxides, substituted piperidines, unsaturated piperidines, N-acyl substituted piperidines, diaryl-substituted piperidinones, piperidinone oximes, and piperidine hydrazides.

Piperidine nitroxides

2, 2, 6, 6-Tetramethylpiperidin-1-yl) oxyl (TEMPO) and 2, 2, 6, 6-tetramethyl-4-piperidinol-N-oxyl (TEMPOL) (Fig. 3), piperidine nitroxides are potent antioxidant agents due to their ability to scavenge reactive free radicals. Trnka *et al.* synthesized and evaluated the antioxidant activity of TEMPOL derivatives; 2, 2, 6, 6-tetramethyl-4-(5-(triphenylphosphonio) pentoxy) piperidin-1-oxy bromide (Mito TEMPOL) (Fig. 3) and hydroxylamine of Mito TEMPOL (Mito TEMPOL-H). Mito TEMPOL is readily converted to Mito TEMPOL-H by ubiquinol within mitochondria. Authors mentioned that antioxidant activity of TEMPOL might be due to its hydroxylamine and they combinedly protect mitochondria from oxidative damage [15].

In a study by Kim *et al.*, synthesis of 3, 4, 5-trisubstituted piperidines (Fig. 4) was carried out enantioselectively, and their antioxidant activity was determined. Few of them contain a TEMPO moiety in their structure. Among all the substituted piperidines, compounds containing TEMPO moiety and hydroxy group showed promising antioxidant activity than the other compounds [16].

Substituted piperidines

Frietas *et al.* evaluated the *in vitro* antioxidant activity of 12-[(2R, 5R, 6R)-5-hydroxy-6-methyl piperidin-2-one (iso-6-cassine; ISO) (Fig. 5) by determining the activity of glutathione-S-transferase, catalase, glutathione peroxidase, and lipid peroxidation. Anticonvulsant activity was demonstrated using pilocarpine-induced seizures in rats. ISO exhibited potent *in vitro* antioxidant activity and also increased the latency to the onset of seizures. Authors suggested that anticonvulsant activity of ISO was might be due to its antioxidant activity [7].

2,3-Dihydro-1H-benzo[d]imidazole moiety was introduced at the 4th position of 3-methyl-2,6-diarylpiperidine by condensing O-phenylenediamine and 3-methyl-2,6-diarylpiperidine-4-one. The obtained compound 3'-methyl-2', 6'-diphenyl-1,3-dihydrospiro[benzo[d]imidazole-2,4'-piperidine (Fig. 6) was evaluated for its antioxidant activity using DPPH assay. The compound showed significant free radical scavenging activity [17].

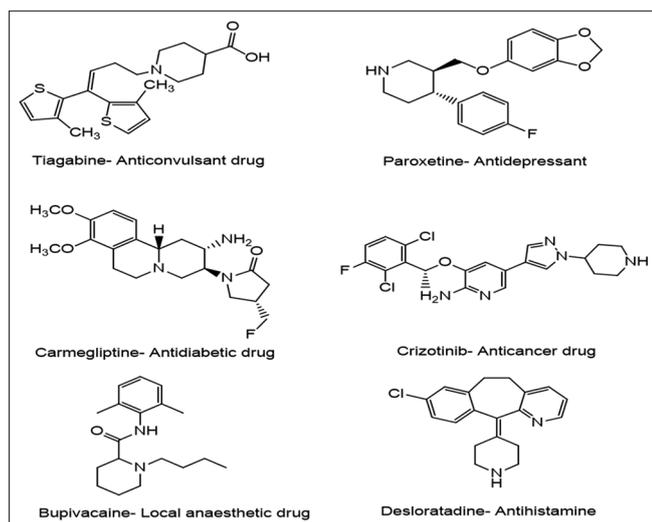


Fig. 1: Clinically available piperidine containing drugs

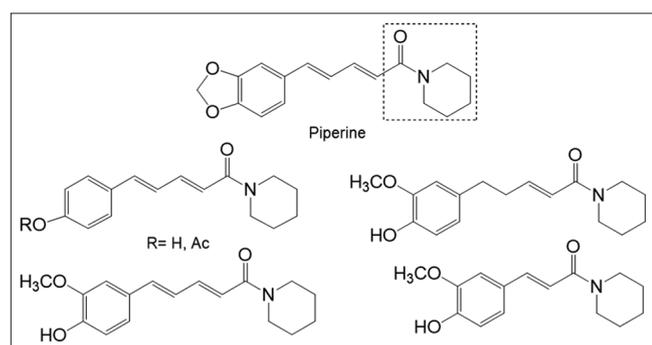


Fig. 2: Naturally occurring compounds bearing piperidine ring

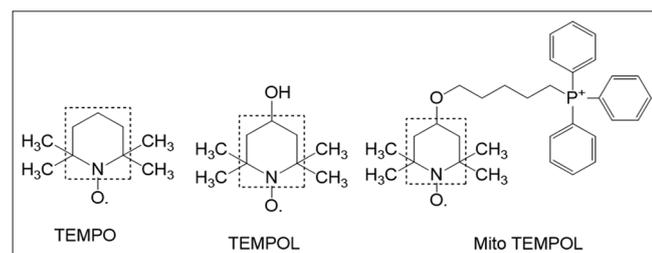


Fig. 3: Structures of 2, 2, 6, 6-Tetramethylpiperidin-1-yl oxyl, 2, 2, 6, 6-tetramethyl-4-piperidinol-N-oxyl (TEMPOL) and Mito TEMPOL

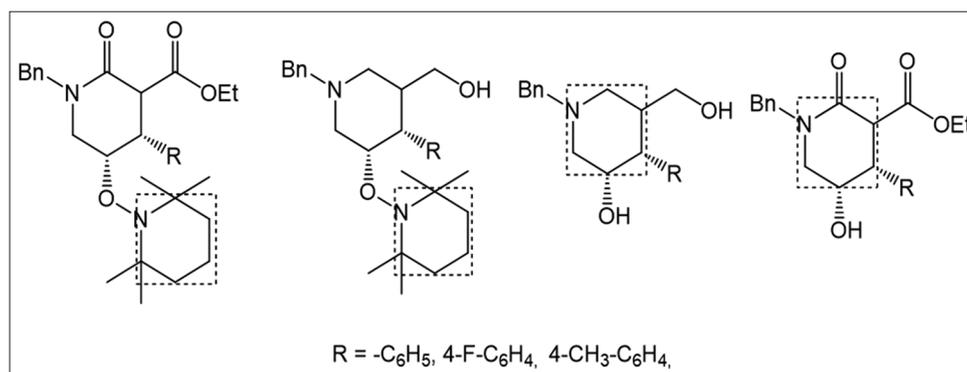


Fig. 4: 3, 4, 5-trisubstituted piperidines

Alexidis *et al.* synthesized different piperidine derivatives (Fig. 7) by introducing several substituent groups with chelating properties. Cysteamine derivatives exhibited potent antioxidant activity in the assays such as lipid peroxidation inhibitory assay, hydroxyl radical scavenging assay, and DPPH assay due to oxidizable SH group. The derivatives in which SH group was replaced by hydroxy and amine functionality, showed poor antioxidant activity. The lipid peroxidation inhibition was time and concentration dependent [18].

Kiasalari *et al.* prepared phenylcyclohexane (PCP) and its analog 1-[1-(3-methoxyphenyl) (tetralyl)] piperidine (PCP1) (Fig. 8) by introducing phenyl cyclohexane ring and 1-(2-methoxyphenyl)-1, 2, 3, 4-tetrahydronaphthalene ring at piperidine nitrogen. Antioxidant activity of PCP and PCP1 was evaluated using malondialdehyde, nitric oxide (NO), and superoxide dismutase (SOD) assessment and their anticonvulsant effect was studied using pentylenetetrazol-induced kindling model. The results showed that PCP1 exhibited marked antioxidant and anticonvulsant activity when compared to PCP [5].

Different derivatives of ethyl N-aryl-2, 6-dioxo-piperid-3-ene-4-carboxylates (Fig. 9) were synthesized and screened for their antioxidant and antimicrobial activities. Aryl groups possessing electron donating/ withdrawing groups were introduced on the nitrogen of ethyl 2,6-dioxo-1,2,3,6-tetrahydropiperidine-4-carboxylate. Among the synthesized compounds, derivatives with unsubstituted phenyl ring and 4-nitro substituted phenyl ring showed highest DPPH free radical scavenging activity. Most of the compounds showed moderate antioxidant and antimicrobial activities [19].

2E,5E)-2,5-bis (3-bromo- 5-methoxy- 4-(2-(piperidin-1-yl) ethoxy) benzylidene) cyclopentanone; (2E,5E)-2,5-bis (3-bromo- 5-methoxy- 4-(2-(2-methylpiperidin-1-yl) ethoxy) benzylidene) cyclopentanone; (2E,5E)-2,5-bis (3-bromo-5-methoxy-4-(3-(piperidin-1-yl)propoxy) benzylidene) cyclopentanone; and (2E,5E)-2,5-bis (3-bromo-5-methoxy-4-(3-(2-methylpiperidin-1-yl)propoxy)benzylidene)cyclopentanone (Fig. 10) were synthesized while preparing C5-curcuminoids. These compounds were evaluated for their *in vitro* antioxidant (DPPH free radical scavenging activity) and antibacterial activities. Derivatives possessing amino functionalities showed poor antibacterial activity [20].

A series of piperidine sulfonamide derivatives (Fig. 11) were synthesized by coupling of different sulfonyl chlorides with 4-(piperidin-1-yl) aniline. Antioxidant activity was evaluated using DPPH assay. Enzyme inhibitory activities were screened using butyrylcholinesterase, lipoxygenase, and acetylcholinesterase enzymes. N-[4-((4-piperidin-1-yl) acetamido) benzenesulfonamide demonstrated good scavenging activity in DPPH assay. This study also concluded that the unsubstituted derivative can be a potent molecule for the treatment of Alzheimer's disease, cancer, inflammation, and bronchial asthma [21].

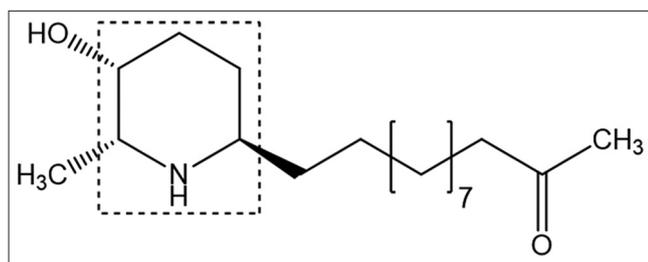


Fig. 5: Iso-6-cassine

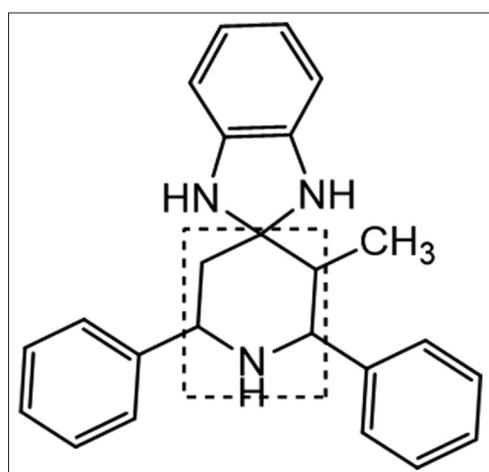


Fig. 6: Structure of 3'-methyl-2', 6'-diphenyl-1, 3-dihydrospiro[benzo[d]imidazole-2,4'-piperidine

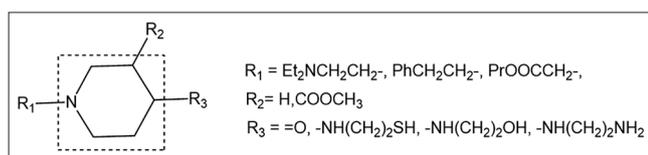


Fig. 7: N- substituted piperidines bearing cysteamine moiety

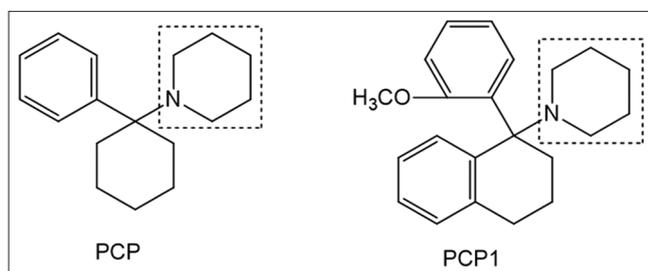


Fig. 8: Structures of phenylcyclidine (PCP) and its analog (PCP1)

A series of alkyl piperidines (Fig. 12) were synthesized using piperidine-2-methanol and piperidine-2-ethanol as parent structures and screened for their antioxidant, antibacterial, and antifungal activities. It was observed that a compound possessing fluoro group at para position demonstrated good antioxidant, antibacterial, and antifungal activities. Results concluded that good antioxidant activity of the active compounds was due to the presence of halogens fused with methanol group [22].

A novel series of nitrogen-containing (piperidine, morpholine, and N-methyl piperazine) benzophenone analogs (Fig. 13) were synthesized by Mannich reaction. The compounds elicited inhibitory activity against tumor necrosis factor- α (TNF- α), interleukin-6 (IL-6)

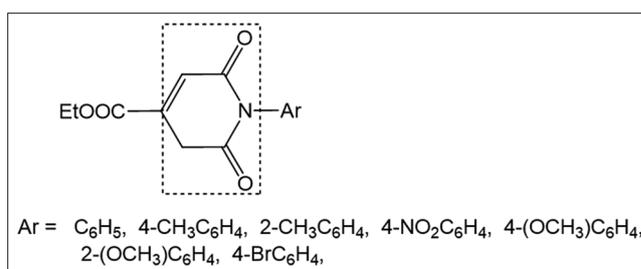


Fig. 9: Structure of ethyl N-aryl-2, 6-dioxo-piperid-3-ene-4-carboxylates

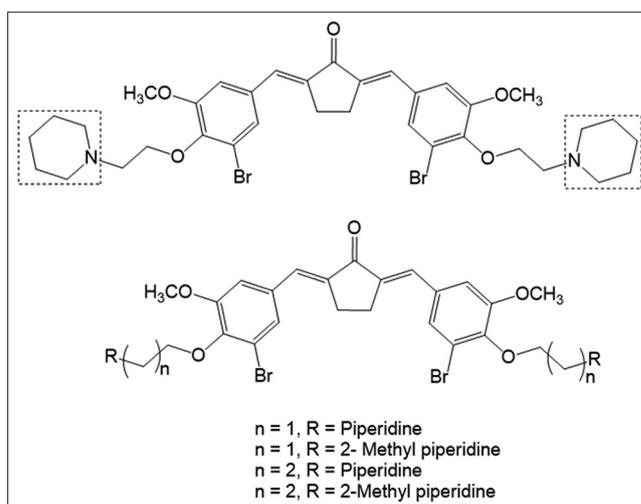


Fig. 10: N- substituted piperidines

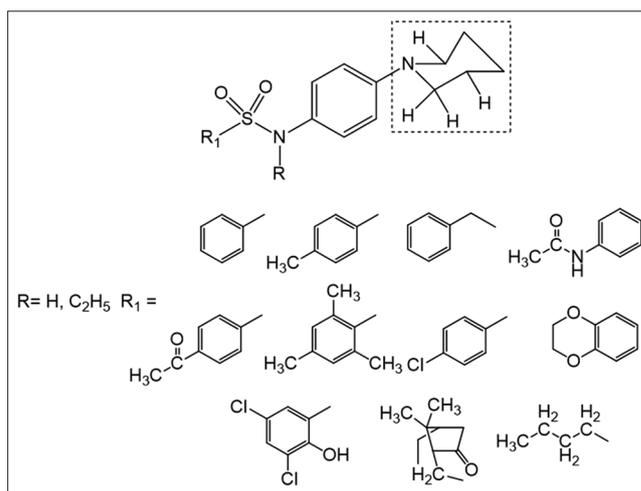


Fig. 11: Piperidine coupled sulfonamide derivatives

cytokines and found to be active in DPPH assay. Piperidine substituted compounds showed good antioxidant activity and also elicited 20–100% TNF- α inhibition at 10 μM and 83–100% IL-6 inhibition at 10 μM , respectively [23].

Unsaturated piperidine derivatives

A series of piperidine containing derivatives (Fig. 14) were synthesized, where nitrogen is substituted with an acetyl group. These derivatives were screened for antioxidant (DPPH and SOD assays) and antimicrobial activities. Among all, 1-Adamantylthio derivative bearing methyl group demonstrated potent antioxidant activity (in both the assays) and antimicrobial activity [24].

Piperidine derivatives (unsaturated at 3 and 4 positions) (Fig. 15) were synthesized, and their antioxidant and antimicrobial activities were evaluated. Among the synthesized derivatives, derivative bearing methoxy substituent at R_3 and R_8 positions showed highest DPPH free radical scavenging activity whereas compound possessing cyano group at R_2 position showed poor scavenging activity. Antibacterial activity was evaluated using agar disc diffusion method, in which compound containing trifluoromethyl group at R_3 and R_8 positions showed poor antibacterial activity and the compound which contains cyano group at R_2 position showed potent antibacterial activity [25].

N-acyl substituted piperidines

Piperamide derivatives (Fig. 16) were synthesized by treating different piperazine and piperidine compounds with (E)-3-(7-methoxybenzo[d][1,3]dioxol-5-yl)acrylic acid. The compounds were evaluated for their antibacterial, antifungal (disc diffusion method), antidepressant (forced swim test and tail suspension test), antioxidant (DPPH and superoxide radical scavenging method) activities and also for their monoaminoxidase A and B inhibitory activity. Among the synthesized piperamides, the one possessing hydroxyl group on 4th position of piperidine ring showed highest antioxidant capacity, whereas 4-phenyl substituted piperidine derivative exhibited poor antioxidant activity indicating that this substitution is unfavorable for the activity [4].

Piperamide derivatives (Fig. 17) were synthesized using different substituted cinnamic acids. The prepared compounds were evaluated for their antibacterial and antioxidant (using DPPH and H_2O_2 radical

scavenging assay) activities. Significant antioxidant and antibacterial activities were noticed for methoxy containing piperamides. Among all the prepared derivatives, derivative bearing chloro group on phenyl ring demonstrated promising antifungal activity [26].

[1-(Substituted-benzoyl)-piperidin-4-yl]-(2,4-difluoro-phenyl)-methanone oximes (Fig. 18) were synthesized and screened for their *in vitro* antioxidant (DPPH and ferrous ion chelating assays), antibacterial, and antiproliferative (MTT(3-(4,5-dimethylthiazolyl-2)-2,5-diphenyltetrazolium bromide) assay) activities. All the derivatives exhibited good antioxidant activity, in that; compound bearing trimethoxy substitution on phenyl ring demonstrated promising antioxidant activity in both the assays [27].

6-Fluoro-3-(piperidin-4-yl) benzo [d] isoxazole derivatives (Fig. 19) were synthesized and evaluated for their antioxidant and antimicrobial activities. Compounds were evaluated for their antioxidant efficacy in H_2O_2 and DPPH methods. Among all, derivatives with electron withdrawing groups (F, Cl) exerted good antioxidant and antimicrobial activities [28].

Piperidine conjugated benzisoxazole derivatives (Fig. 20) were synthesized and screened for their antioxidant, antibacterial, and anti-inflammatory activities. The results indicated that compounds bearing electron donating groups such as OCH_3 , CH_3 groups exerted good inhibitory activity in DPPH, hydroxyl, and superoxide anion radical scavenging assays, while compounds with dinitro substituents exhibited highest antibacterial activity, compounds with nitro group

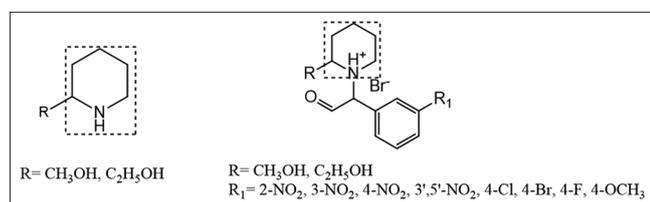


Fig. 12: Structures of alkyl piperidines

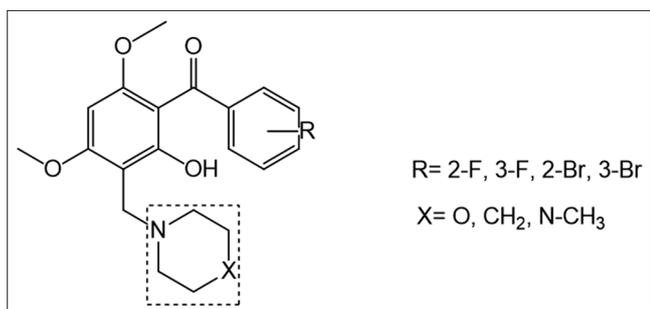


Fig. 13: Nitrogen-containing benzophenone derivatives

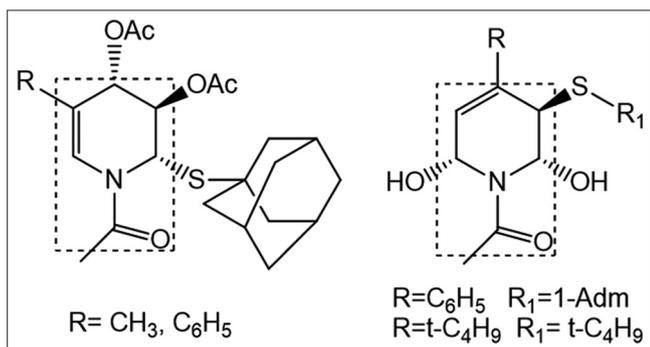


Fig. 14: N-acyl substituted piperidines (unsaturated piperidine ring)

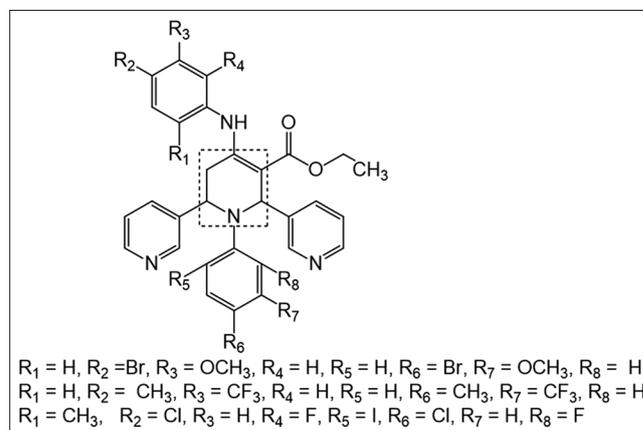


Fig. 15: Structure of piperidine derivatives which contain unsaturation at 3 and 4 positions

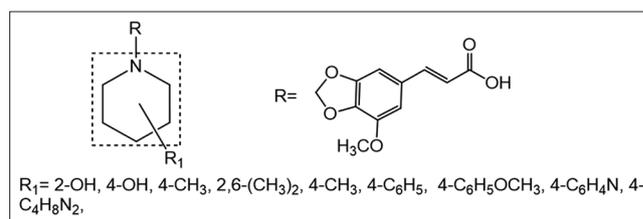


Fig. 16: N-acyl substituted piperidine derivatives

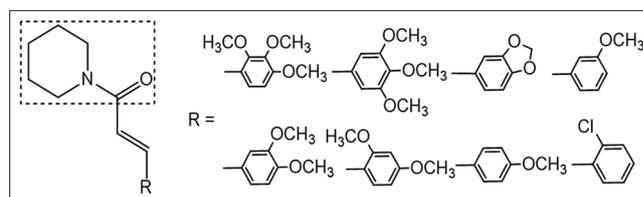


Fig. 17: Piperamide derivatives

showed highest anti-inflammatory activity in both phospholipase A2 and lipoxygenase inhibition assays [29]. By replacing sulfonyl group with carbonyl moiety, authors further synthesized several benzisoxazole derivatives (Fig. 20) and determined their antioxidant, antibacterial, and anti-inflammatory activities.

Methyl and methoxy substituted derivatives displayed good antioxidant activity against DPPH, superoxide anion, and hydroxyl radical scavenging assays. Antibacterial activity was evaluated using disc diffusion method, in which compound with unsubstituted phenyl ring exhibited highest antibacterial activity. The derivatives containing 4-methyl or 2/4-nitro group showed good anti-inflammatory activity [30].

Diaryl substituted piperidinones

Piperidinone based compounds, especially substituted with aryl rings at C₂ and C₆ positions display promising antioxidant activities. Ajay *et al.* recently reviewed various synthetic procedures and biological activities of diaryl substituted piperidinones [31]. The antioxidant potential of 3-benzylidene-5-methyl-2,6-diarylpiperidin-4-ones (Fig. 21) was studied using DPPH assay. In the tested compounds (Z)-3-(4-chlorobenzylidene)-2,6-bis(4-chlorophenyl)-5-methylpiperidin-4-one and (Z)-3-(3-nitrobenzylidene)-5-methyl-2,6-bis(3-nitrophenyl) piperidin-4-one (Fig. 21) were found to be more active when compared to ascorbic acid. Authors mentioned that electron-donating nature of the substituents such as hydroxy or methoxy groups on 3-arylidene-4-piperidones might be responsible for antioxidant activity [32].

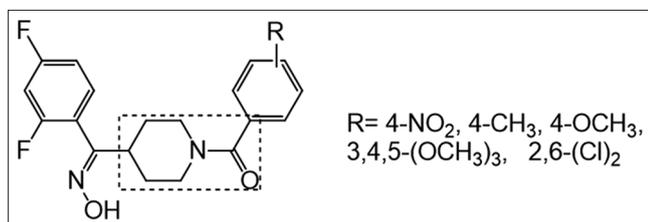


Fig. 18: N-acyl substituted piperidine derivatives

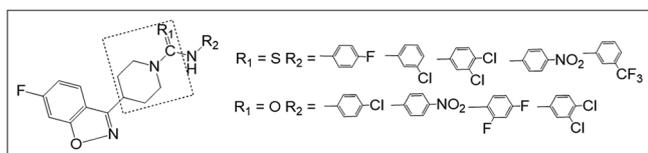


Fig. 19: 6-Fluoro-3-(piperidin-4-yl) benzo [d] isoxazole derivatives

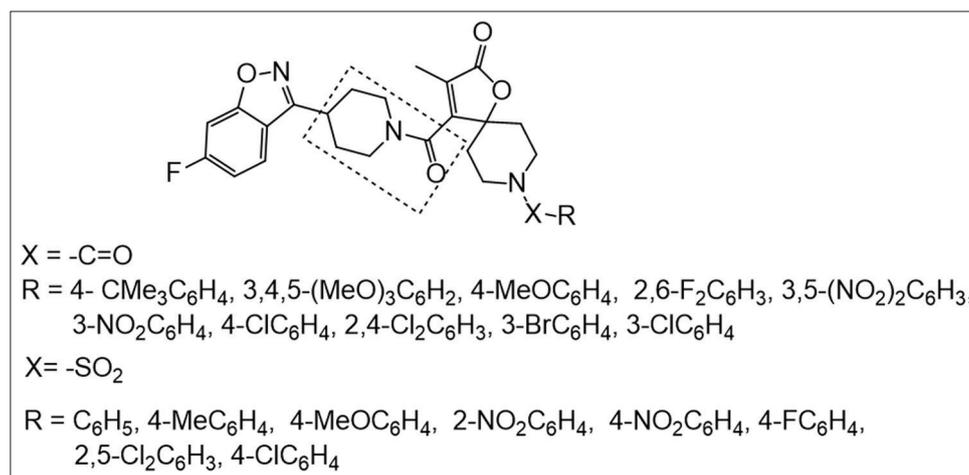


Fig. 20: Piperidine conjugated benzisoxazole derivatives

Substituted piperidines and their respective alcohols (Fig. 22) were synthesized and evaluated for their antioxidant activity. Piperidones showed a highest antioxidant effect when compared to the isomeric alcohols [33].

A series of 4-methyl-N'-(3-alkyl-2r,6C-diaryl piperidin-4-ylidene)-1,2,3-thiadiazole-5-carbohydrazides (Fig. 23) were synthesized and evaluated for their antioxidant, antimicrobial, and anticancer activities. Antioxidant activity was evaluated by DPPH, ABTS, superoxide, hydroxyl, and NO free radical scavenging assays. Compounds possessing electron donating groups (methoxy and methyl) at para position of phenyl ring exhibited significant free radical scavenging activity. Compounds possessing electron withdrawing groups such as fluoro or chloro or bromo groups at para position of phenyl ring which is attached to piperidine ring at C₂ and C₆ positions, displayed promising antitumor and antimicrobial activities [6].

Piperidinone oximes

A series of novel 2,6-diphenyl-3-alkylpiperidin-4-one-O-[2,4,6-tritertiary butyl cyclohexa-2-5-dienon-4-yl] oximes (Fig. 24) were synthesized, and their antioxidant and antimicrobial activities were screened. Most of the compounds were effective free radical scavengers in DPPH, superoxide, NO, ABTS, and hydroxyl assays. Compound with the electron releasing ethyl group exhibited promising antioxidant activity when compared to other compounds. Results showed that the compound possessing ethyl group at 3rd position on piperidine moiety showed potent antimicrobial and antifungal activities [34].

Substituted piperidine oximes were synthesized and evaluated for their antioxidant (DPPH and superoxide free radical scavenging assays) and anti-inflammatory (carrageenan-induced rat paw edema) activities. 3,3-Dimethyl 2,6-dimethyl piperidine 4-one oxime (Fig. 25) exerted potent antioxidant activity. The active compound also displayed marked anti-inflammatory activity comparable to standard drug dexamethasone [35].

Harini *et al.* synthesized various piperidinone oxime esters (Fig. 26) and screened *in vitro* antioxidant and antimicrobial activities. Results showed that antioxidant activity was (DPPH assay, ABTS radical scavenging assay, FRAP assay, and cupric ion reducing antioxidant capacity assay) enhanced when the hydroxy group was introduced on the phenyl ring. Among the synthesized compounds, compounds with electronegative fluoro or chloro or bromo group on benzoyl ester moiety showed excellent antibacterial activity whereas compound with the fluoro group exhibited significant antifungal activity [36].

Harini *et al.* synthesized piperidinone oxime esters by replacing 4-methoxy phenyl ring with vanillin moiety few (Fig. 27) and screened their antioxidant activity. Compounds bearing hydroxy groups on

phenyl ring demonstrated potent antioxidant activity. Compounds possessing fluoro or chloro group displayed good antibacterial and antifungal activities [37].

Further, thiazole-based piperidinone oximes were synthesized by the same research group, and their antioxidant and

antimicrobial activities were evaluated. Among all, 2,6-bis(4-hydroxy-3-methoxyphenyl)-1-methylpiperidin-4-one O-(2-(4-hydroxy-3-methoxybenzylidene)hydrazinyl)thiazol-4-yl)oxime

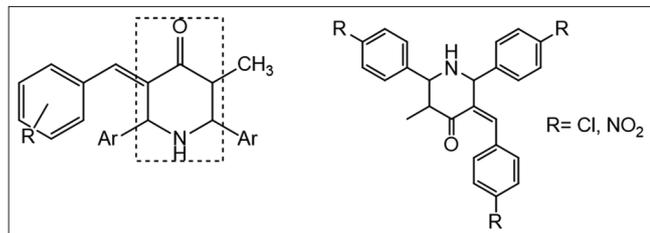


Fig. 21: Diaryl substituted piperidinones

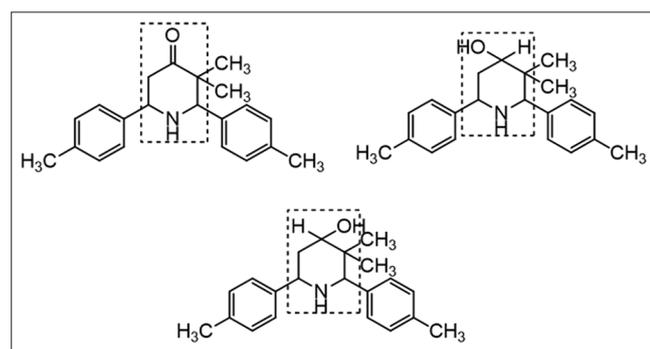


Fig. 22: Substituted piperidines

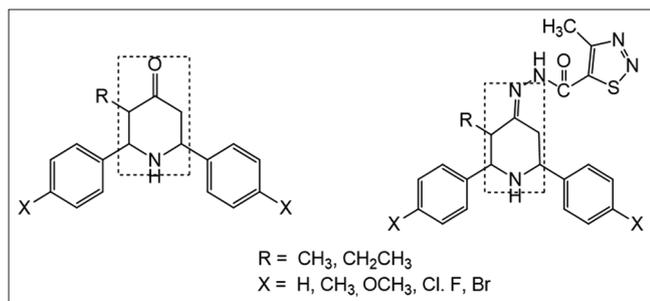


Fig. 23: Structures of 4-methyl-N'-(3-alkyl-2r, 6C-diaryl piperidin-4-ylidene)-1,2,3-thiadiazole-5-carbohydrazides

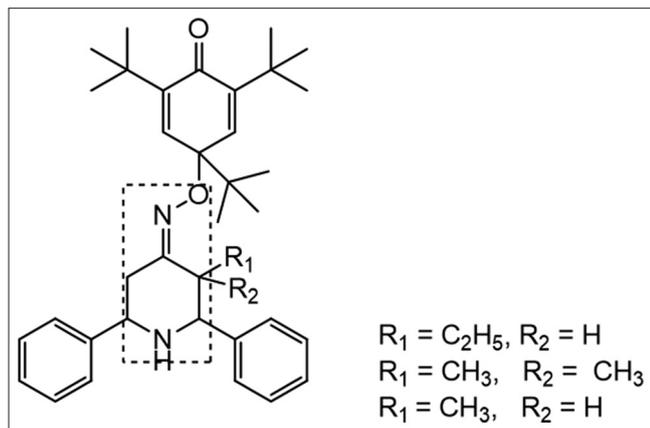


Fig. 24: Substituted 2,6-diphenyl-3-alkylpiperidin-4-one-O-[2,4,6-tritertiary butyl cyclohexa-2-5-dienon-4-yl] oximes

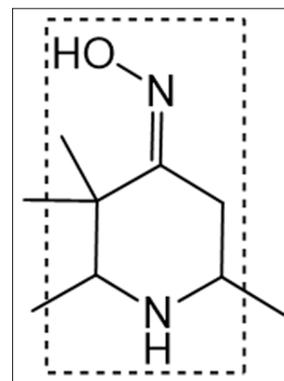


Fig. 25: Substituted piperidine oxime structures

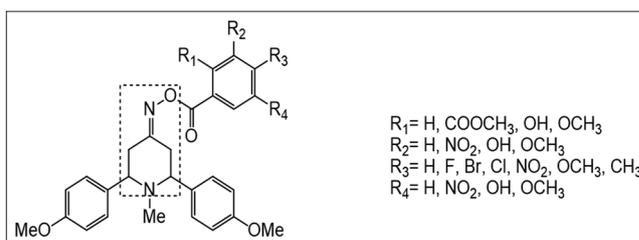


Fig. 26: Piperidinone oxime esters

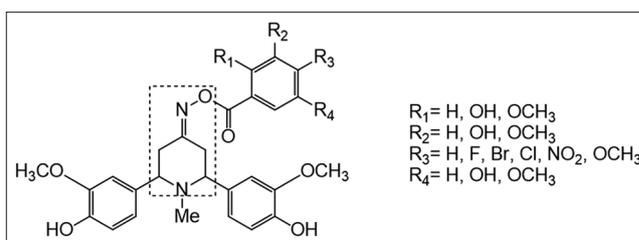


Fig. 27: Piperidinone oxime esters

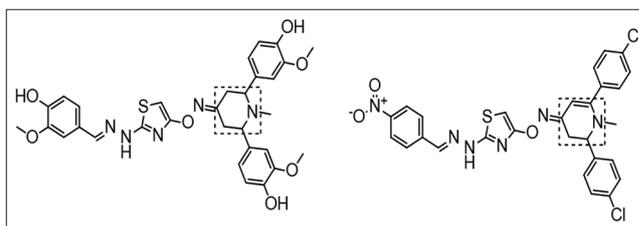


Fig. 28: Thiazole-based piperidinone oximes

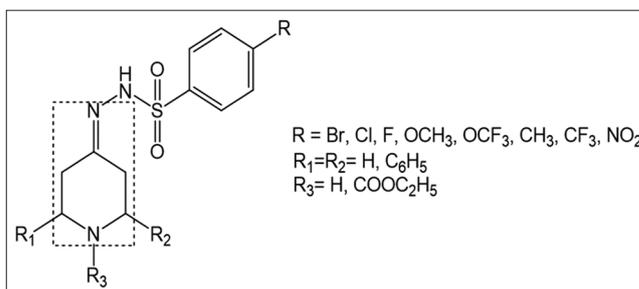


Fig. 29: Sulfonyl hydrazones bearing piperidine ring

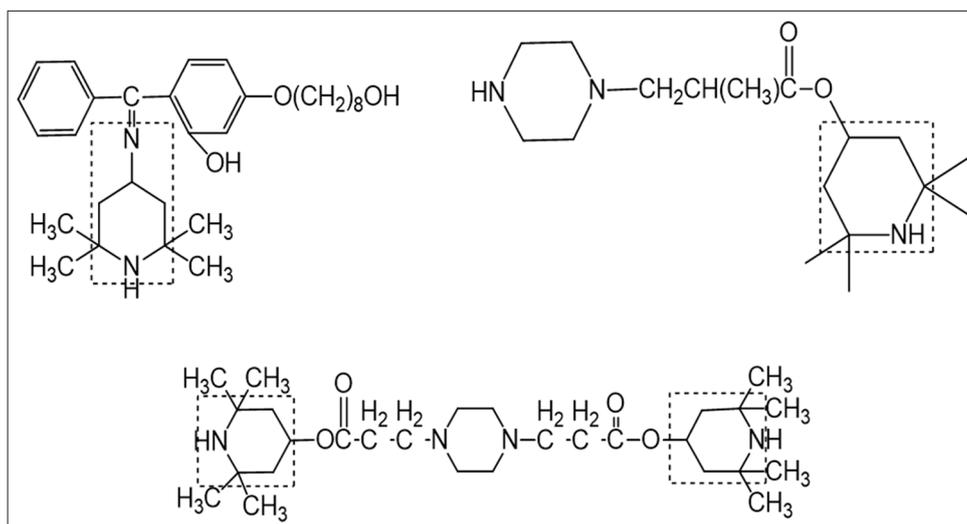


Fig. 30: Imine substituted piperidine derivatives

(Fig. 28) displayed promising antioxidant activity whereas 2,6-bis(4-chloro phenyl)-1-methylpiperidin-4-oneO-(2-(2-(4-nitrobenzylidene)hydrazinyl)thiazol-4-yl)oxime (Fig. 28) exhibited significant antimicrobial activity. Authors mentioned that enhancement in the antioxidant activity of the designed compounds might be due to the radical dissipation activity of thiazole ring and lipophilic nature of piperidinone oxime moieties. It was noticed that active compounds contain the electron donating substituents on the phenyl rings [38].

Piperidine hydrazides

Sulfonylhydrazones bearing piperidine derivatives (Fig. 29) were synthesized by condensing benzene sulfonylhydrazides with 2,6-diphenyl piperidin-4-one and ethyl 4-oxopiperidine-1-carboxylate. Antioxidant and anticholinesterase activity of the compounds was evaluated, and their structural activity relationship was investigated.

Compound possessing methoxy derivative showed good antioxidant activity in CUPRAC assay. Compound substituted with bromo group showed highest antioxidant activity in DPPH, linoleic acid assays and exhibited potent activity in anticholinesterase inhibitory activity [39].

Miscellaneous

Imine substituted hindered piperidine stabilizers (Fig. 30) were synthesized, and their thermal and light stabilizing action was compared. The imine stabilizer structure is based on 2-hydroxy benzophenone and a 4-amino-tetra methyl piperidine structure. The polymer oxidation rate was determined by carbonyl index using Fourier-transform infrared spectroscopy (Fourier-transform infrared spectroscopy), and its hydroperoxide formation was measured. Most of the compounds were highly effective thermal and light stabilizers for polyolefin films and also inhibited the formation of hydroperoxide during thermal aging in polymers [40].

Plant-derived and synthetic piperidines are extremely important for their antioxidant properties [41-43]. For example, naturally occurred piperidine alkaloids demonstrated promising bactericidal, anticancer, and antioxidant properties [44]. The marked relationship between antioxidant properties and life-threatening diseases such as cancer highlight the significance of piperidines in the current drug research.

AUTHOR'S CONTRIBUTIONS

Manjusha RK and Shaheen Begum collected the articles and drafted the manuscript. Arifa Begum reviewed and drafted the article. Bharathi K supervised the review work. All authors discussed and finalized the manuscript.

CONFLICTS OF INTEREST

None.

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