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LEAVES EXTRACT-BASED BIOGENIC SYNTHESIS OF CUPRIC OXIDE NANOPARTICLES, CHARACTERIZATIONS, AND ANTIMICROBIAL ACTIVITY

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ABSTRACT

Objectives: Tulsi (*Ocimum tenuiflorum*) leaves extract-based synthesis of cupric oxide nanoparticles (CuONPs), characterizations, and antimicrobial activity.

Methods: The small cut leaves were washed with double distilled water and boiled for 30 min. After filtration, the extract was treated with 0.2 M copper acetate solution and the initial color change of this solution indicated formation of copper nanoparticles. This solution was stirred for a specific time, heated and treated with 0.1 M NaOH solution. The formation of CuONPs was confirmed by the development of brownish-black precipitates. Then, CuONPs have been tested for their antibacterial effects by applying well diffusion method against *Escherichia coli, Streptococcus mutans, Proteus vulgaris,* and *Staphylococcus aureus*.

Results: The biologically synthesized CuONPs have been well characterized by using ultraviolet-visible, Fourier-transform infrared, X-ray powder diffraction, and field-emission scanning electron microscopy techniques and all these analytical methods indicated a successful and efficient formation of CuONPs. After the incubation period, significant zones of inhibition were observed for *E. coli, S. mutans, P. vulgaris,* and *S. aureus.*

Conclusions: The method was found highly efficient, eco-friendly, and low cost for the synthesis of biologically important CuONPs. The CuONPs have been found an excellent antibacterial agent.

Keywords: Cupric oxide nanoparticles, Ocimum tenuiflorum, Biological synthesis, Characterizations, Antimicrobial activity.

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INTRODUCTION

The major advantages of nanotechnology in chemical, biological, and engineering sciences are to improve the efficiency, durability, and effectiveness of the materials. The physical, chemical, and biological properties of nanomaterials are usually differ and improved due to the change in the ratio of surface area to volume [1]. The conventional methods used to synthesize nanoparticles are not much efficient, needed harmful chemicals, and unsafe to the environment. The applicability of nanotechnology with green processes has achieved great attention in the synthesis of metal and metal oxide nanoparticles, and these nanoparticles show many unique biochemical, physicochemical, and optoelectronic properties. The common metal and metal oxide nanoparticles are silver (Ag), gold (Au), copper (Cu), manganese (Mn), iron (Fe), palladium (Pd), platinum (Pt), cupric oxide (CuO), manganese oxide (MnO2), calcium oxide (CaO), magnesium oxide (MgO), iron oxide (Fe2O3), zinc oxide (ZnO), and titanium oxide (TiO2) [1-12]. Researchers have now focused on the developing of green synthetic methods for these nanoparticles because such nanoparticles are used as catalysts, in medical applications, disinfection, as antimicrobials, semiconductors, cosmetics, and chemical sensing devices [13-17]. Cupric oxide nanoparticles (CuONPs) have attracted great attention due to their applications in biomedical, supercapacitors, magnetic storage media, sensors, catalysis, and semiconductors [18-20]. Biologically synthesized CuONPs are safer than chemically originated CuONPs and biomedically become more effective to inhibit the growth of microorganisms [23]. The plant Tulsi (Ocimum tenuiflorum) belongs to Angiospermic family Lamiaceae and has many medicinal properties and used in traditional medicines and pharmaceuticals. In India, this plant is known for worship and grown at temples and homes. Metabolites present in the aerial part of the plant are used in the treatment of diarrhea, bronchitis, dysentery, etc [21-24]. The leaves of O. tenuiflorum have been collected from a local garden near Uttaranchal University Dehradun (India).

METHODS

Preparation of leaves extract

After several washing of collected leaves of *O. tenuiflorum* with double distilled water, leaves were cut into possible small pieces and 2 g of leaves added with 100 ml of double distilled water. The content was boiled for 30 min on a magnetic stirrer with a constant shake. After boiling, the mixture was cooled down and filtered with Whatman filter paper. The filtrate was used as an extract and preserved at 4°C for further uses.

Synthesis and characterizations of CuONPs

Take a requisite amount of copper acetate Cu (CH₂COO)₂ in 100 ml of double distilled water to make 0.2 M copper acetate solution. Add 10 ml of leaves extract in this solution and mixed properly with a shake for 25-30 min on a magnetic stirrer. The color change of this solution from blue to green indicated the formation of copper nanoparticles. Now, this solution was continuously stirred for 2.5 h and then heated at 80°C for 3 min and then added dropwise 0.1 M NaOH solutions. The formation of brownish-black precipitate indicated the formation of CuONPs. This black precipitate was isolated from the solution by centrifugation at 10,000 rpm for 10 min. Further, precipitate washed with ethanol to remove all impurities and then heated at 60-65°C in a hot air oven. The dried CuONPs were preserved in airtight bottles for characterizations and antimicrobial activities. The characterization methods included Fourier transform infrared (FTIR), ultraviolet (UV)visible, X-ray powder diffraction (XRD), and field-emission scanning electron microscopy (FESEM) [21,25].

Antimicrobial activity

The antibacterial activity of CuONPs against *Escherichia coli*, *Streptococcus mutans, Staphylococcus aureus,* and *Proteus vulgaris* was checked using well diffusion method. The liquid media was poured on sterilized Petri plates and then solidified. The nanoparticles were loaded on the wells on the agar plates and then incubated. After the incubation period, the zones of inhibition around the nanoparticles have been measured [20,25].

RESULTS

Characterizations of CuONPs

UV-visible

UV-visible spectroscopy is the very economic and common method used to analyze organic, polymeric as well inorganic compounds. Electron transition takes place from lower energy to higher energy levels in the atoms or molecules after absorption of the precise wavelength of UV and visible radiations. The UV-visible spectra of CuONPs are represented in Fig. 1 in the range of wavelength 220–400 nm.

FTIR

FTIR spectroscopy is used to observe the presence of bonds on the surface of biologically originated nanomaterials. This technique is very useful in the study of a variety of quantitative analysis and bonding mechanisms on the surface of solids. It is also very economical, sensitive, and nondestructive interference-based technique [26,27]. The FTIR spectra of CuONPs is shown in Fig. 2 in the range between 4000 and 500 cm⁻¹.

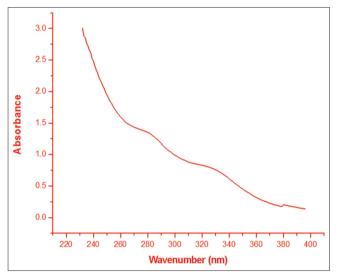


Fig. 1: Ultraviolet-visible spectra of cupric oxide nanoparticles

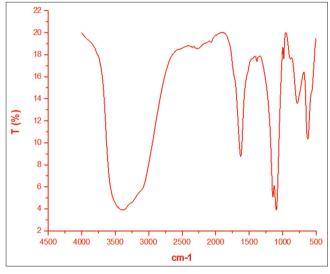


Fig. 2: Fourier transform infrared spectra of cupric oxide nanoparticles

Powder XRD

XRD is an advanced analytical technique used for the analysis of finely ground and homogenized materials. It provides phase identifications, unit cell dimensions, and determines the average bulk compositions. The scattered X-rays from particles produce a diffraction pattern and which finally provides an atomic arrangement in crystals [28-30]. The XRD pattern of CuONPs is represented in Fig. 3.

FESEM

FESEM is based on the scanning of nanoparticles with high-energy electrons under higher vacuum conditions. It provides the morphological characteristics of nanoparticles, and the FESEM images of CuONPs are shown in Fig. 4; the agglomerated CuONPs show spherical as well as thick-needle morphologies.

Antimicrobial activity

Most of metal and metal oxide nanoparticles have now been identified as good antimicrobial agents. Due to their small size and effectiveness, they show efficient inhibition mechanisms inside the cell of microorganisms. Biologically synthesized nanomaterials are more efficient and unique than chemically originated nanomaterials. Biogenic CuONPs are one of them and very effective to destroy bacterial cells; the antibacterial activity of CuONPs has been checked by using well diffusion method against *S. mutans, E. coli, S. aureus,* and *P. vulgaris.* The CuONPs were loaded on the wells on the agar plate and after the incubation period, significant zones of inhibition found around the loaded nanoparticles.

DISCUSSION

In UV-visible spectra, the elevated peaks are obtained at 285 and 330 nm indicate the formation of CuONPs using Tulsi leaves extract at such

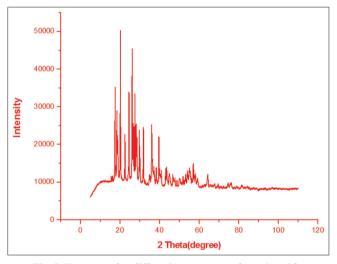


Fig. 3: X-ray powder diffraction patterns of cupric oxide nanoparticles

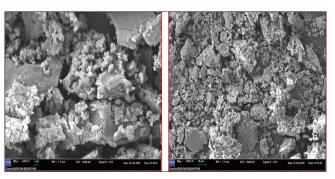


Fig. 4: FESEM images of cupric oxide nanoparticles

wavelengths. Broad peaks of FTIR are found at 3392 cm⁻¹, 1625 cm⁻¹, 1384 cm⁻¹, 1144 cm⁻¹, 1094 cm⁻¹, 982 cm⁻¹, 780 cm⁻¹, and 621 cm⁻¹. These peaks indicate the presence of O-H, N-H, C=O, C-O, C-C, Cu-O, Cu-C, etc., bonds on the surface of biologically synthesized CuONPs. The XRD peaks are assigned at 32.6°0, 38.6°, 48.9°, 53.4°, 58.2°, 61.8°, 66.2°, 68.2°, and 76.2°. These peaks are corresponding to 110, 111, 202, 020, 202, 113, 022, 220, and 311 planes, respectively [18,21]. The CuONPs enter into the bacterial cells which cause distortions and destroy the cell membranes and that finally cause death a bacterial cell [1,9,15,18,31,32]. At initial dosage 5 mg/ml of CuONPs, the zones of inhibitions have been observed 18 mm, 12 mm, 14 mm, and 16 mm for S. mutans, S. aureus, E. coli, and P. vulgaris. The zones of inhibition were increased with increase in the amount of CuONPs on the agar plates and maximum zones of inhibition recorded 38 mm. 34 mm. 32 mm. and 28 mm for S. mutans, S. aureus, E. coli, and P. vulgaris 15 mg/ml of CuONPs (Table 1, Fig. 5).

CONCLUSIONS

The biological-based nanomaterials are specific and have more potential and with some advanced properties than chemically synthesized nanobased materials. In this study, we have observed the applied green synthetic method which is more efficient and suitable to synthesize CuONPs. These nanoparticles have been well analyzed by FTIR,

Bacterial species	Amount of CuONPs (mg/ml)	Zones of inhibition (mm)
S. mutans	5	18
S. aureus	5	12
E. coli	5	14
P. vulgaris	5	16
S. mutans	7.5	26
S. aureus	7.5	18
E. coli	7.5	16
P. vulgaris	7.5	18
S. mutans	10	30
S. aureus	10	24
E. coli	10	22
P. vulgaris	10	18
S. mutans	12.5	36
S. aureus	12.5	28
E. coli	12.5	26
P. vulgaris	12.5	24
S. mutans	15	38
S. aureus	15	34
E. coli	15	32
P. vulgaris	15	28

E. coli: Escherichia coli, S. mutans: Streptococcus mutans, P. vulgaris: Proteus vulgaris, S. aureus: Staphylococcus aureus

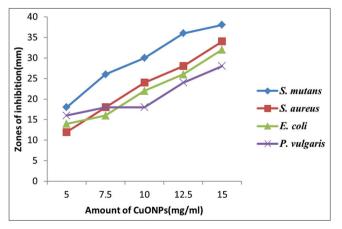


Fig. 5: Zones of inhibition with amount of cupric oxide nanoparticles

UV-visible, FESEM, and XRD and found a good antibacterial agent for *S. mutans, S. aureus, E. coli*, and *P. vulgaris* under our laboratory scales.

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AUTHORS' CONTRIBUTIONS

The experimental work was carried out by Mrs. Yashwani Prakash under the supervision of Dr. Naveen Chandra Joshi in Advanced chemistry laboratory, Uttaranchal University, Dehradun (India). The characterizations of CuONPs and writing work have been completed by Dr. Naveen Chandra Joshi.

CONFLICT OF INTEREST

The authors declare that they have no conflict of interest.

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