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Research Article

DEVELOPMENT OF AQUEOUS POLYMERIC DISPERSION (APD) OF EUDRAGIT E PO-PERFORMANCE CHARACTERIZATION FOR AQUEOUS BASED TASTE MASKING COATING SYSTEM

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ABSTRACT

The present research work was performed to develop an Aqueous Polymeric Dispersions (APD) of Eudragit E PO using emulsification solvent evaporation technology including optimization of solvent, emulsifier & stabilizer. The basis rationale for the present research is to develop water based coating system over organic solvent based coating systems to achieve certain advantages over organic solvents with respect to ecological, toxicological and manufacturing safety concerns, including Pollution, Explosion hazards, Risk of operators, High cost of organic solvents, solvent toxicity and many more. The APD of Eudragit E PO was developed by emulsification solvent evaporation technology including optimization of Ethyl Acetate as solvent, 1%w/w SLS as emulsifier & 4% w/v Polaxomer F127 as stabilizer. The Concentration of Eudragit E PO was optimized as 15%w/v among three APD formulations F1-5%, F2-10% & F3-15% by optimizing Particle size, pH & Viscosity parameters. The optimized APD (F3-15%w/v APD) was characterized for different parameters viz. Average Particle Size by TEM & found 732 nm, pH by electrode pH meter & found 6.8 & viscocity by brook's field viscometer & found 0.6 Pascle. The Optimized APD (F3-15%w/v APD) was characterized for any possible interactions by DSC & FT-IR spectra. Three different free films were prepared with three different plasticizer viz. di-butyl phthalate, propylene glycol and tri-ethylcitrate to plasticize optimized aqueous polymeric dispersion of Eudragit EPO (F3-15%w/v APD). Tri-ethyl-citrate 30%w/v of dry polymer was optimized for the development of the free films from an optimized APD(F3). The optimized aqueous polymeric dispersion of Eudragit EPO (F3-15%w/v APD) was used for the performance characterization for taste masking coating properties of ofloxacin tablets. APD (F3-15%w/v APD) coated Ofloxacin tablets were comparatively evaluated with Eudragit E PO Organic solvent coated Ofloxacin tablets for different parameters viz. (a.) Physical- Hardness, Friability, Wt. Variation etc (b.) in-vitro dissolution profile- In Distilled water & Simulated Gastric Fluid (SGF) (c.) Biological-Taste masking. The accelerated stability study of optimized APD (F3-15%w/v APD) was also performed for 30 days evaluated for Particle size, Viscosity & Redispersibility. It was concluded that the APD was stable at room temperature & unstable at 4-8°C.

Keywords: APD, Eudragit E PO, In vitro release study, TEM, FT-IR, DSC, Polaxomer F127.

INTRODUCTION

Water soluble polymers are always being in good demand because of their certain advantages over organic solvents based coating systems with respect to ecological, toxicological and manufacturing safety concerns, including Pollution, Explosion hazards, Risk of operators, High cost of organic solvents, solvent toxicity and many more.

Aqueous polymeric dispersions, latexes, or pseudolatexs are all colloidal systems in which high molecular weight polymers are homogeneously dispersed in submicron sizes with the aid of surfactant(s) and other stabilizing agents ^[1].

There are a wide range of water insoluble polymers includes Cellulose: Ethyl cellulose, Vinylics: Polyvinyl acetate phthalate, Cellulosic esters: cellulose acetate phalate (CAP), Acryates: Eudragit, & many more which provide a great research platform to develop an aqueous polymeric dispersions (APD) with the intention of conferring benefits and properties to the dosage form over the uncoated variety or organic coating.

To comply with government restrictions on the emission of solvents to the atmosphere, the coating industry has developed replacements for organic solvent based coating to decrease or eliminate solvent emissions, including water based coatings, high solid coatings and powder coatings. Presently, water based coatings are among the most promising candidate systems for functionality, effectiveness, convenience and economics ^[2].

However, these aqueous polymeric dispersions have their obvious disadvantages which are typical of colloids like they can be irreversibly affected by various factors such as electrolytes, heat or frost, pH changes and high shear forces ^[3-6].

The present research work was performed to develop an Aqueous Polymeric Dispersions (APD) of Eudragit E PO using emulsification solvent evaporation technology including optimization of solvent, emulsifier & stabilizer. The basis rationale for the present research is to develop water based coating system over organic solvent based coating systems to achieve certain advantages over organic solvents.

Eudragit EPO polymer is a type of Poly (meth) acrylates.It is a cationic polymer with Dimethylaminoethyl methacrylate as a functional group helps you to seal sensitive activities and increase patient compliance by masking taste and odors. Even thin layers of Eudragit provide the desired effect, making it an extremely economical application. It is a versatile polymer for protective Oral Dosage Formulation. This polymer used for Insulating coating, taste masking, odour masking, moisture protection and light protection ^[7].

MATERIALS AND METHODS

Eudragit E PO (B. No- G071231197) was gifted by M/s Evonik, Mumbai. Ofloxacin was obtained from Kasliwal brothers, Indore and other chemicals was from Loba chemie Pvt. Ltd., Mumbai, India. All chemicals used were of analytical grade.

1. Preparation of Aqueous Polymeric Dispersion of Eudragit E PO by Emulsification Solvent Evaporation Method

Aqueous polymeric dispersion was prepared by emulsification solvent evaporation method in two steps using Eudragit E PO, organic phase, stabilizer and emulsifier [9].

Step 1: Preparation of oil-in-water crude emulsion

For the preparation of oil-in-waster crude emulsion, organic phase, emulsifier and stabilizer were selected and there quantities were optimized.

1.1 Selection of organic phase

Different solvent can be used for the preparation of organic phase. Polymer may solubilized or swelled in the solvent. Polymer solution should be clear for the emulsion preparation. The polymer should not be more viscous. 0.5 gm of Eudragit E PO was taken and dissolved in 10 ml of solvent/solvent system and kept for half an hour. Different solvents including methanol, ethanol, acetone, and toluene were used. The solution was observed for the clarity of solution.Toluene has revealed least solubility for polymer (100mg in 1ml). Ethyl acetate showed the solubility in 150mg in 1ml. In Methanol the solubility of polymer was 350smg/ml. In Acetone the solubility of polymer was 550smg/ml. In Ethanol solubility of polymer was more than other solvents (600smg/ml). The multiple solvent Acetone and Ethanol showed more solubility than other solvents (700mg/ml). Finally it was concluded that for single solvent system Ethanol was optimized & for multiple solvent system

Acetone & Ethanol (1:1) were optimized to prepare a cost effective technique.

1.2 Selection of emulsifier

The emulsifiers are required to prepare a stable aqueous polymeric dispersion. Emulsifiers play an important role in the preparation of colloidal polymer dispersions. They facilitate the formation of the emulsion and prevent agglomeration and coalescence of the dispersed polymer particles during solvent evaporation and storage^[6,7]. The emulsifying agent operative in the aqueous medium in relatively low concentrations, generally included in proportions of about 0.1 to 3% by weight of the water in aqueous phase (%w/v). Two different formulations were prepared with two different types of emulsifiers; SLS and Tween80 were used. Formulation F1 containing Tween 80 & F2 containing SLS were formed to make an emulsion with milky white appearances. The selection of the emulsifier was done on the basis of physical evaluation Viz. Appearance, pH, Viscosity & Particle size.

Table 1:- Selection of emulsifier

S.No.	Parameter	F1 (Tween 80)	F2 (SLS)
1	Appearance	Milky white	Milky white
2	рН	6.2	6.5
3	Viscosity	20 cps	14 cps
4	Particle size distribution	0.45 μm	0.756 µm

The Optimized Emulsifier was SLS because of better stability & easy redispersibility.

1.3 Optimization of emulsifier concentration

Three different formulations were prepared with various concentrations of emulsifier in process to optimize the concentration of SLS.

Table 2:- Optimization of emulsifier concentration

		Formulation code		
S.No	Ingredient	F1 (0.5%)	F2 (1%)	F3 (1.5%)
1	Eudragit E PO	1gm	1gm	1gm
2	Ethyl Acetate	30ml	30ml	30ml
3	Cetyl-alcohal (w/w)	50mg	50mg	50mg
4	SLS(w/v)	1gm	2gm	3gm
5	Distilled	100ml	100ml	100ml

Out of these three formulations, formulation F1 was neutralized pH, lesser viscosity and lesser particle size.

1.4 Optimization of stabilizer concentration

Five different formulations were prepared for the optimization of stabilizer concentration with different concentration of Polaxomer F127. Optimized Formulation F4 was given in table no.3. Different

parameters were preformed like pH, viscosity and particle size distribution for the best formulation.

Table 3:- Oblimization of stabilizer's concentratio	on of stabilizer's concentrat	tion
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S.No.	Parameter	F1 (0.5%)	
1	Appearance	Milky white	
2	pH	6.9	
3	Viscosity	0.54Ps	
4	Particle size distribution	0.86 µm	

Out of these three formulations, F4 formulation was neutralized pH, lesser viscosity and lesser particle size determination.

Step 2: Preparation of the aqueous dispersion from optimized o/w emulsion

To prepare oil-in-water crude emulsion polymer was dissolved in organic solvent. Emulsifier and stabilizer dissolved in aqueous phase at 75°C temperature and then drop wise addition of polymer solution in aqueous phase with mechanical stirrer operated at 1000 rotation per minute (rpm) and crude emulsion was obtained. Then prepared emulsion was placed in a rotating flask of the rotary evaporator and slowly rotated at 50°C and 500 mm Hg vacuum to remove the organic solvent and polymer solution was concentrated. This process was continued for three hours to obtained aqueous polymeric dispersion¹⁶.

2.1: Optimization of Eudragit EPO concentration for development of aqueous polymeric dispersion of Eudragit EPO

Three different formulations F1, F2 & F3 of aqueous polymeric dispersion of Eudragit E PO were prepared by emulsification solvent evaporation method and characterized for different parameters like color, odor, particle size distribution, pH and viscosity were observed.

	Table 4:- 0	ptimization	of Eudragit	EPO	concentration
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S.no	Ingredients	Formulation (F3-15%APD)
1	Polymer	Eudragit EPO 12.5gm
2	Organic solvent	Ethyl acetate 20ml
4	Surfactant	SLS
		(1.0%w/v)
5.	Stabilizer	Polaxomer F127
		(4%w/v)
6.	Plasticizer	Stearic Acid
		(10%w/w)
7.	Distilled water	50ml

2.2: Characterization of aqueous polymeric dispersion of Eudragit EPO

The different formulations of Aqueous Polymeric Dispersion of Eudragit EPO were characterized for the following parameters.

2.2.1: Color, odor and appearance of aqueous dispersion

The aqueous polymeric dispersion colour was milky white and has no characteristic odor that indicates the complete removal of the solvent form the formulations. Appearance of the aqueous polymeric dispersion was also good.

2.2.2: Particle Size Determination

The aqueous polymeric dispersion of Eudragit EPO formulations F1, F2, F3, F4 and F5 were diluted with distilled water and poured in to glass plates. The particle size of dispersion was measure by optical microscope ^[7].

Least count = (N2/N1) 0.01mm

Where, N1=Division of the eye piece

N2=Division of the stage micrometer

(0.01mm is 1 division of stage micrometer)

The diameter of the particles was calculated by multiplying the division of the eyepiece to the least count. The average particle size of the formulations 5%, 10%, and 15% and found 0.838 μ m, 0.24 μ m, 0.27 μ m and 0.28 μ m respectively All the formulations formed dispersion. Due to large particle size of the dispersed phase, sedimentation was observed immediate half an hour, which although redispersed after shaking ^[9, 10].

2.2.3: pH of aqueous polymeric dispersion of Eudragit EPO

The pH plays an important role on the aqueous polymeric dispersion suability ^[11]. Formulations subjected for the pH determination using the digital pH meter.

Table 5:-	pH of	different	formulation
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S.No	Formulation	Ph	
1	F1	6.9	
2	F2	6.5	
3	F3 (Optimized)	6.8	

2.2.4: Viscosity of Aqueous Polymeric Dispersion

Viscosity is an expression of the resistance of a fluid of flow, the higher the viscosity the greater the resistance. Viscosity measurement is very important for the dispersion system also ^[11]. Viscosities of the Aqueous Polymeric Dispersion were measure by the digital viscometer using spindle no. three at room temperature

Table 6:- Viscosity of different formulations

S.No	Formulation	Viscosity (Ps)	
1	F1	0.54 Ps	
2	F2	0.58 Ps	
3	F3 (Optimized)	0.60 Ps	

2.2.5: Characterization of developed Aqueous Polymeric Dispersion of Eudragit E PO- After evaluating all the above said three formulations for different parameter, F3 formulation (with 15% w/v Eudragit EPO) was found as the best optimized formula for the development of aqueous polymeric dispersion of Eudragit E PO.

Table 7: Characterization results of the formulation F3

S. No.	Parameter	Formulation F3
1	Odor	Odorless
2	Appearance	White dispersion
3	Particle shape	Spherical
4	Average particle size (nm)	730nm
	As determined by TEM	
5	рН	6.8
6	Viscosity (Ps)	0.6Ps

3. Spray drying of the Optimized APD Formulation

The optimized formulation of the polymer was spray dried to convert its physical state from milky white liquid dispersion to spray dried white powder in order to extend the stability & reconstitution purpose for coating ^[12].

4. INTERACTION STUDIES [13, 14]

Chemical/Molecular Interaction study of the prepared spray dried APD was performed using following techniques:-

4.1 FT-IR Spectroscopy- Infrared spectroscopy of the Polymers & its spray dried APD were performed for the interaction purpose. The FT-IR spectra of pure polymer & its spray dried APD form was compared with the help of Overlay FT-IR Spectra ^[13, 14].

Conclusion- Overlay FT-IR spectra confirms that there is no molecular interaction in the basic structure of the polymer- Eudragit EPO.

4.2 DSC Analysis- DSc Analysis were done of the Polymers & its spray dried APD by recording their DSc thermogarm & Glass transition temperatures.

Conclusion- DSC thermogram were recorded for the poymer as well as its spray dried APD form. Overlay DSC thermograms clears that there is a decrease in glass transition tempreture (Tg) of spray dried APD form as compared to pure polymer. It means APD formation decreases glass transition tempreture (Tg) of the polymer & increases its elasticity.

OVERLAY FT-IR SPECTRA



Fig No. 1 - Eudragit EPO V/s Spray dried APD of Eudragit EPO



Fig No. 2- DSC Thermogram of Eudragit E PO

5. Development of film of Aqueous Polymeric Dispersion of Eudragit EPO

Three different films were prepared with three different plasticizer di-butyl phthalate; propylene glycol and tri-ethyl-citrate were used to plasticize aqueous polymeric dispersion of Eudragit EPO & one film was prepared without any plasticizer (Free Films). Among all the three formulation 15% w/v aqueous polymeric dispersion was formed the film with tri-ethyl-citrate (30%w/v of polymer) & 15% w/v aqueous polymeric dispersion free films were also formed without addition of any plasticizers ^[15].

5.1 Preparation of film of aqueous polymeric dispersion of Eudragit EPO

A clear film was obtained with tri-ethyl-citrate plasticizer. The thickness of the film was found approx $173 \,\mu\text{m}$.

5.2 Optimized Formulation- 15% APD with 30%w/v of polymer of tri-ethyl-citrate.

Thickness- 173 μm

Folding Endurance- 250+

- Moisture Content- 8%
- % Moisture Uptake- 7%

6. Performance Characterization of the Developed Aqueous Polymeric Dispersion Eudragit EPO

Ofloxacin uncoated tablets were prepared and these Ofloxacin uncoated tablets were coated with the aqueous polymeric dispersion of the Eudragit EPO. Eudragit EPO with organic solution was taken and drug release study was determined and compared the aqueous based coating, organic based coating and marketed based coating formulation ^[16].

Fable 8: Forn	ulation of	Ofloxacin	Tablets
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S.No.	Ingredients	Quantity (in mg)	Uses
1	Ofloxacin	100	Active ingredients
2	Microcrystalline cellulose	50	Filler
3	Lactose	35	Diluent/Filler
4	Magnesium state	2	Lubricant
5	Talc	3	Glidant

6.1 Evaluation of tablets

 Table 9: Evaluation parameter of tablets ^[17]

S.No.	Parameters	Result
1	Wt variation test	Within the limits (±7.5%)
2	Friability test	Within the limits (0.6%)
3	Hardness test	5kg/cm ²
4	Disintegration test	Pass

6.2 Drug release study from the uncoated and aqueous polymeric dispersion of Eudragit E PO coated tablets and comparative study with aqueous polymeric dispersion of Eudragit E PO coated tablets and organic solution coated tablets ^[18, 19].

A. Drug release study in distilled water

100 mg Ofloxacin tablet was weighed accurately and added to 900 ml deionized water and maintained at 37°C. Drug release was performed at 100 rpm for 30 minutes. A 5-ml sample removed from mixtures each kept at 5 to 30 minutes was filtered, and the amount of drug was estimated spectrophotometrically by using UV-spectrophotometer at 292nm against deionized water as a blank. ^[18], ^{19]}. Dissolution study of selected formulations were carried out for 30min. by using USP 24 type II dissolution apparatus. It was found that only 0.8% & 0.6% (Absorbance=0.036, *y=0.0864*0.6-0.0156*) of the drug was released in distilled water from APD coated tablets & Organic base coated tablets respectively indicating the stability of the optimized coating system and suitability for selecting distilled water as a vehicle for suspension.

B. Drug release study in simulated gastric pH

100 mg Ofloxacin tablet was weighed accurately and added to 900 ml simulated gastric pH i.e.1.2 (SGF) water and maintained at 37°C. Drug release was performed at 100 rpm for 30 minutes. A 5-ml sample was removed after specific time intervals from the mixtures and each kept at 5 to 30 minutes, filtered, and the amount of drug estimated spectrophotometrically by was using UVspectrophotometer at 292nm against SGF as a blank [18,19]. Dissolution study of selected formulations were carried out for 2 hrs by using USP 24 type II dissolution apparatus. It was found that the drug released more than 50% drug within 30min at average gastric pH- & more than 95% (95.88%) drug was released within 2hrs 1.2 in case of both i.e. APD coated tablets & Organic base coated tablets hence the optimized coating system not effected the basic pharmacokinetics of the drug & confirm the drug release at gastric media.

Table No. 9: Observation table for drug release profile of

optimized formulation of Ofloxacin				
Time	% CPDR	% CPDR		
(min.)	Drug Release From	Drug Release From		
	APD Coated Tablets	Organic Coated		
		Tablets		
120	95.88	96.78		



Fig No. 3 In-vitro drug release profile in simulated gastric pH of APD base coated Ofloxacin Tablet

6.3 Biological evaluation

Evaluation of bitterness was done by time sensitivity method on 10 healthy human volunteers for 15 minutes to confirm the palatability of the APD coated tablets. It was found that none of the volunteers felt considerable bitter taste of the optimized formulation after 15 min^[20].

Table 10: Observation table for Bitterness Evaluation test h	ŊУ
time sensitivity method- APD Coated Tablets	

Volunteers	Bitterness levels after					
	10	1	2	5	10	15
	sec.	min.	min.	min.	min.	min.
1.	Х	0	0	0	0	0
2.	0	0	0	0	0	0
3.	0	0	0	0	0	0
4.	Х	Х	1	0	0	0
5.	1	0	0	0	0	0
6.	Х	Х	Х	0	0	0
7.	1	Х	0	0	0	0
8.	Х	0	0	0	0	0
9.	Х	0	0	0	0	0
10.	0	0	0	0	0	0

7. Stability Study of Prepared Aqueous Polymeric Dispersion of Eudragit EPO

The Aqueous Polymeric Dispersion comes under the category of functional excipients. It was designed to use for coating of pellets and tablets to prepare sustained release or modified release dosage form. It should maintain its physical form during the storage up to use. The stability study was designed to find out the preferred storage conditions for the aqueous dispersion. The prepared Aqueous Polymeric Dispersion was kept in different storage conditions like freezing, room temperature and accelerated conditions ($40\pm2^{\circ}C$ and $70\pm5\%$ RH) [^{21-22]}.

Table No. 11: Stability testing

S.N o	Evaluation Parameters	Freshly prepar	At 4-8°C		At room temperature	
		ed APD	15 days	30 days	15 days	30 days
1.	Particle size	0.898µ m	Faile d	Faile d	0. 924μ m	0.960µ m
2.	рН	6.8	Faile d	Faile d	6.4	6.0
3.	Viscosity (cps)	0.6Ps	Faile d	Faile d	0.73P s	0.82Ps
4.	Redispersibi liy	2 Hrs	Faile d	Faile d	1.7 Hrs	1.5 Hrs

RESULTS & DISCUSSIONS

In the present research work, The development of 15%w/v Aqueous Polymeric Dispersions (APD) of Eudragit E PO were developed by using emulsification solvent evaporation technology in three following steps:-

A.) Development of Aqueous Polymeric Dispersion

B.) Development of Polymeric films of optimized Aqueous Polymeric Dispersions

C.) Performance Characterization of developed Freeze-Dried Aqueous Polymeric Dispersions

The purpose is to replace an organic solvent by aqueous solvent in order to achieve conferring benefits over Organic solvents for different aqueous insoluble polymers in pharmaceutical industries including:

- Cost reduction
- Non-toxicity
- Ecological benefits
- Manufacturing safety
- Ease of availability

Table 12: Optimized Formulations

Polym er	Organi c solven t Phase	Emulsifi er (%w/v of Aq. Phase)	Surfacta nt (%w/w of dry polymer wt.)	Plasticiz er	Aqueo us Phase
Eudragi t E PO	Ethyl Acetat e	1% SLS	4% Polaxom er F127	Stearic Acid	q.s.

Table 13: Characterization Of Optimized Formulations

S. No.	Parameter	Eudragit		
		E PO APD		
		15% w/v		
1	Odor	Odorless		
2	Appearance	Milky White dispersion		
3	Particle shape	Spherical		
4	Average particle size			
	(nm)	730nm		
	As determined by TEM			
5	рН	6.8		
6	Viscosity	0.60Ps		

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REFERENCES

- Wang J and Ghebra S, Aqueous Polymeric Dispersions as a film former in pharmaceutical dosage forms: Dispersed systems, edited by Liberman HA, Rieger Martin and Banker G S, Vol I, 2nd edition, Marcel Dekker Inc., 1996.
- Wheatley T A, Steuernagel C R,Latex emulsion for controlled drug delivery, in: J.M. McGinity (Ed.), Aqueous Polymeric Coating For Pharmaceutical Dosage Forms, Dekker, New York, 1997, pp. 1-54.
- Vanderhoff J W, theory of colloids in pharmaceutical dosage forms : Dispersed systems, edited by Liberman HA, Rieger Martin and Banker G S, Vol I, Second edition, Marcel dekker Inc., 1996
- Quintanar- Guerrero D, Allemann E, Doelker E, Fessi H, pseudolatex prepration using a novel emulsion-diffusion process involving direct displacement of partially watermiscible solvents by distillation, International Journal of Pharmaceutics 188 (1999), 155-164.
- Bindschaedler C, Gumy R and Doelker E, Theoretical Concepts Regarding the Formation of Films from Aqueous Microdispersions and Application to Coating, Lab. Pharm. Probl. Tech., 31, 1987.
- Bodmeier R, Siepmann J, Nondegradable Polymers for Drug Delivery, in Encyclopedia of controlled drug delivery, vol. II, A wiley Interscience Publication, 1997.
- 7. www.roehm.com, Rohm Gabh & Co. KG,Pharma Polymers, Evonik Degussa.
- J.W Vanderhoff., 1984. Inversion Emulsion Polymerization of Acrylamide: Polymerization Kinetics And Process Development. Journal Of Dispersion Science And Technology, Vol. 5, Issue 3-4, pp. 323-363.
- 9. Subramanyam CVS, A Textbook of physical pharmacy, Vallabh Prakashan, 2000.
- Lehmann KOR, Chemistry and application properties of polymethacrylate coating systems, In: McGinity, J.W. (Ed.), Aqueous Polymeric 9.Coatings for Pharmaceutical Dosage Forms. Marcel Dekker, Inc., New York, 1997, pp. 101–176.
- V. Gallardo^a, M.E. Morales^a, M.A. Ruiz^a, A.V. Delgado; European Journal of Pharmaceutical Sciences 26 (2006) 170–175.
- Pawar. A, Theory and Practical of Pharmaceutics-1, Second Edition, Career Publication, July 2012, "Spray Drying", 150-151 & 264-265.
- 13. Chatwal G.K. Instrumental method of analysis, Meeruth publication 2005, Page no 4.2-26
- 14. Silverstene, "Spectrophotometric Estimation of chemicals" 4th edition, 2005
- 15. Nicholson JW., 1990. Wasson EA. Film spreading and film formation by waterborne coatings. Surface Coatings. Elsevier Applied Science, Vol. 3, pp. 91–123.
- 16. Lachman. L, Lieberman. A.H, Industrial Pharmacy, Special Indian Edition, 2009,"Manufacturing of tablets", 317-324.
- 17. Lachman. L, Lieberman. A.H, Industrial Pharmacy, Special Indian Edition, 2009, "Evaluation of tablets", 296-303.
- United States Pharmacopoeia-25, Asian Edition, 2002, "Dissolution test", 2011-2022.
- 19. Lachman. L, Lieberman. A.H, Industrial Pharmacy, Special Indian Edition, 2009,"Dissolution test", 301-303.
- Borodkin, S.; Yonker, M.H.; Journal of Pharmaceutical Sciences; 1970, 59(40), 481.
- Sanjay Bajaj, Dinesh Singla and Neha Sakhuja: Stability Testing of Pharmaceutical Products, Journal of Applied Pharmaceutical Sciences, 02 (03); 2012: 129-138.
- 22. Handbook of Stability Testing In Pharmaceutical Development, Part II, Stability Methodologies And Best Practices, Springer Sciences & Business Media, https://books.google.co.in/books.