

THE EFFECT OF ANTICOAGULANT TYPES ON THE *IN VITRO* ANALYSIS OF CLOPIDOGREL IN HUMAN PLASMA USING LIQUID CHROMATOGRAPHY TANDEM-MASS SPECTROMETRY

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ABSTRACT

Objective: The aim of this study was to optimize and validate a plasma clopidogrel analysis method using liquid chromatography tandem-mass spectrometry.

Methods: Plasma samples were analyzed using a BEH C18 column (1.7 μ m; 100 mm \times 2.1 mm), the mobile phase was 0.1% formic acid in acetonitrile (30:70, v/v). The flow rate was 0.2 mL/min, with a column temperature set to 35°C, an injection volume of 5 μ L, an analysis time of 4 min, and irbesartan as the internal standard. Aliquots were obtained by liquid-liquid extraction using ammonium acetate and diethyl ether. The stability and peak area ratio of the respective plasma area responses were evaluated using ANOVA.

Results: No significant differences ($p > 0.05$) were observed between anticoagulants regarding analyte stability. However, the peak area ratio showed significant differences ($p < 0.05$) between the anticoagulants. The accuracy and precision of the analysis with citrate, heparin, and ethylenediaminetetraacetic acid (EDTA) plasma met the quality requirements, and a linear calibration curve was created with concentrations ranging from 0.02 to 5.0 ng/mL.

Conclusion: The results showed that improved analysis of clopidogrel was achieved using citrate or heparin plasma compared with EDTA plasma.

Keywords: Citrate, Clopidogrel, Ethylenediaminetetraacetic acid, Heparin, Liquid chromatography tandem-mass spectrometry.

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INTRODUCTION

Clopidogrel is one of the slowest-onset action prodrugs (Brunton, Lazo, and Parker, 2006) and is the drug of choice for antiplatelet agents [1]. Clopidogrel is also used as secondary prevention of cerebral and cerebrovascular infarction in patients who do not tolerate acetylsalicylic acid or experience new attacks while using acetylsalicylic acid [2]. Clopidogrel is adenosine diphosphate receptor antagonist and irreversibly inhibits platelet function [3].

The maximum concentration of clopidogrel in plasma is reportedly very low, that is, 7921.49 \pm 3921.39 pg/mL [4]. Thus, developing accurate analytical methods for clopidogrel are important, particularly for the examination and monitoring of clopidogrel levels in plasma, which is the most common biological matrix for the analysis of drugs in the body. Plasma is obtained from blood samples with the use of anticoagulants. Thus, an important factor when obtaining plasma is the type of anticoagulant used and studies often use different anticoagulants, making comparisons between analytical methods difficult [5,6]. The ethylenediaminetetraacetic acid (EDTA), heparin, and citrate are anticoagulants often used in drug [5].

The anticoagulants themselves are generally present in the blood collection tube at a concentration sufficient to inhibit blood clotting, as the use of excess concentrations may lead to undesirable effects during bioanalysis [7]. However, the use of anticoagulants can lead to errors during the analysis of certain drugs [8]. Thus, selecting the appropriate anticoagulant during drug analysis serves to provide many benefits, including minimizing interferences and improving the stability of the drug and/or its metabolites [7]. Indeed, differences between anticoagulants have been known to affect the measurement of small molecules [9], metabolic profiles [6,10], and clinical parameters of the drug being analyzed [11-13]. In addition, some of the ions (e.g., Na⁺ or K⁺) present in the plasma anticoagulant such as Na-citrate and

K-EDTA may cause ion suppression or ion enhancement of the drug and its metabolites, which may affect analysis [6]. Moreover, for plasma drug analysis, a selective and sensitive bioanalysis method is required because the plasma level of clopidogrel is very low [14]. Ultra-high-performance liquid chromatography tandem-mass spectrometry (UHPLC-MS/MS) with electrospray ionization (ESI) is a specific and sensitive method of analysis that has become the standard for the measurement of drugs, metabolites and endogenous compounds in biological matrices [15].

This study was conducted to evaluate the commonly used anticoagulant types, namely, citrate, heparin, and EDTA, on parameters such as stability, recovery, matrix effect, and area response of the analyte.

MATERIALS AND METHODS

Materials

Standard clopidogrel (MSN Laboratories, India) and standard irbesartan (Zhejiang Huahai Pharmaceutical, China) were purchased. Blood was taken directly from healthy subjects, and plasma was obtained using the citrate anticoagulant citrate phosphate dextrose adenine (CPD-A) (Indonesian Red Cross). Reagents were high PLC (HPLC)-grade and included acetonitrile, formic acid, diethyl ether, and ammonium acetate were purchased from Merck.

Tools

A ultra PLC-MS/MS (UPLC-MS/MS) system (Waters Xevo TQD Triple Quadrupole) consisting of a quaternary solvent manager (Acquity UPLC H-Class), Sample ManagerFTN (Acquity UPLC), a nitrogen generator compressor (PEAK Scientific), a BEH C18 column (Waters Acquity™ UPLC 100.0 mm \times 2.1 mm, 1.7 μ m), a triple quadrupole (Xevo TQD) mass analyzer with an ionization source (Zspray™), data processing software (MassLynx Software), and a computer was used. Other equipment used included vacuum tubes or Vacutainers with K3EDTA anticoagulant

(Vacuette®), vacuum tubes with the anticoagulant Li-Heparin (Vacuette®), blood collection tubes (BD Vacutainer® Beckton, Dickson and Company), analytical scales (Acculab), degasser (Elmasonic S40H), evaporator (Turbovap), centrifuge (Digital Lab Centrifuge), vortex (Maxi Mix II), pH meter (pH Eutech 510), refrigerator (4–8°C), freezers (–20°C and –80°C) (Biomedical Labtech DeepFreeze), Eppendorf tubes (Soccorex), centrifugation tubes, blue tips, yellow tips, and various glassware.

Sample preparation

The quantitative determination of an analyte and its metabolites in a biological matrix takes place in several stages, including sample storage, sample preparation, separation, identification, and analyte quantification. Sample preparation has three principles: Dissolution of the analyte in a suitable solvent, the removal of any interfering compounds, and the pre-concentration of the analyte. Some of the techniques employed include protein precipitation, liquid-liquid extraction, solid phase extraction, and ultrafiltration. In this research, the sample preparation used liquid-liquid extraction using ammonium acetate and diethyl ether.

Chromatography condition

Chromatographic separation is based on the transfer of a liquid (mobile phase) carrying the analytical mixture through a stationary medium. The interaction difference of the analyte with the surface of the medium results in a difference in retention time between the components of the mixture [16-20]. UPLC differs from other chromatographic methods in the size of the column particles is substantially smaller. With UPLC, the speed of analysis and peak capacity is maximized. In addition, solvents are used less due to the high analysis speeds; thus, UPLC improves efficiency and resolution [21]. Small diameter UPLC diameter particles (<2.0 µm) can improve efficacy (Waters, 2016). The instrumentation of a UPLC system is almost the same as LC and generally consists of a solvent reservoir, pump, injector, column, detector, and a data processing system. The bridged ethylene hybrid column (BEH) has the advantage of high retentivity for basic compounds, good peak shape at high pH, universal columns for various compounds, stability at various pHs and can be used for separation at high temperatures (80 °C) [17].

Analysis conditions

The analytical method used in this study was UPLC. For analysis, was used C18 columns (100 mm×2.1 mm, 1.7 µm), a 0.1% formic acid filtration phase in 0.1% formic acid solution in acetonitrile (30:70, v/v), as mobile phase isocratic elution at 0.2 mL/min flow rate, and a 35°C column temperature. MS detection using the ESI positive ionization method and MRM was performed with capillary tubing of 3.5 kV, a desiccation gas temperature of 400°C, a dissolved gas flow rate of 800 L/h. The obtained m/z values were the molecular weight of the parent ion and the molecular weight of the productions. The molecular weight value of clopidogrel was 321.8 g/mol, while that of irbesartan was 428.5 g/mol, with clopidogrel at m/z 322.086>212.097 and irbesartan at m/z 429.233>207.131.

System suitability test

Clopidogrel and irbesartan standards were carefully weighed and placed into a measuring flask and dissolved with acetonitrile to the measurement limit to obtain a solution of 1 mg/mL (1000 µg/mL). The two dilutes were diluted again to the concentration of the system suitability test solution and the concentration solution of the calibration curve. A 100 ng/mL clopidogrel mixture solution was injected (as much as 10.0 µL) into the chromatography system. Injection results were recorded and counted for the return value (%KV) over five different injections. The calculated parameters were the areas and times of peak analytical and internal retention peaks.

Blood sampling and plasma preparation

Plasma with the citrate anticoagulant CPD-A was obtained from the Indonesian Red Cross (Palang Merah Indonesia), while blood plasma with the anticoagulants heparin and EDTA was obtained from the separation of blood samples taken from six healthy subjects that gave

informed consent. Blood samples were divided into two vacuum tubes, each containing anticoagulants in the form of Li-Heparin and K3EDTA. Blood in each vacuum tube was inverted 10 times to mix the blood with the anticoagulant in the tube. The vacuum tube was centrifuged at 3000 rpm for 10 min at room temperature. The plasma was then transferred to another clean tube. Plasma was stored in a freezer at –80°C until analysis.

Sample preparation

A total of 250 µL plasma containing clopidogrel of a certain concentration plus 20 µL irbesartan (100 ng/mL) was vortexed for 10 s. Plasma with ammonium acetate (0.05 M; pH 6.8) was vortexed for 10 s. The extraction was performed with the addition of diethyl ether, vortexed, and centrifuged at 13,000 rpm for 10 min. A total of 750 µL of the organic phase was separated into clean tubes to evaporate and dry at 60°C for 10 min. The obtained residue was reconstituted in 150 µL acetonitrile, vortexed for 10 s, sonicated for 10 s, and vortexed again for 10 s. The final solution was transferred to a vial insert and centrifuged at 3000 rpm for 5 min. A total of 5.0 µL of the final solution was injected into the chromatographic system.

RESULTS AND DISCUSSION

The internal standard used irbesartan because it has similar physicochemical properties to clopidogrel, particularly its acidity and solubility; thus, it can be detected and eluted using the same analytical method. The analytical conditions resulted in a good separation between the analyte and the standard with clopidogrel retention of 2.68 min and an irbesartan retention of 1.16 min. A relatively short run time of 4 min was also considered a good achievement of this method.

Liquid-liquid extraction methods can produce larger analytical areas [12]. In this study, the liquid-liquid extraction method was optimized. Liquid-liquid extraction is best used for analysis using HPLC-MS/MS because it does not damage the column when compared with the protein precipitation method. In addition, the resulting area response is greater due to the drying process, and fewer additional solvents (concentration process) are used. The optimization of the liquid-liquid extraction process conducted in this study was aimed to obtain an efficient and rapid extraction step so as to minimize contamination. However, during liquid-liquid extraction, an emulsion can sometimes form during processing, so the process needs to be repeated. Thus, a full validation was performed on plasma citrate [22].

Comparison of clopidogrel analysis in three types of anticoagulants

The different characteristics of each anticoagulant, including their different physicochemistry, plasma pH, and ion type differences, it was necessary to analyze three types of anticoagulants by observing the chromatograms and parameters produced by each plasma, including the recovery value, peak area ratio, the stability of the analytes, and matrix effects.

Comparison of the chromatograms and spectrums for all three plasmas

Observations were made by comparing the plasma chromatograms, which showed no significant differences. Other observations were made by comparing the spectrum of the blanks for each plasma. The matrix effect of the three plasmas showed no significant differences.

Comparison of clopidogrel area responses in all three plasmas

Based on the data analysis on Peak Area Ratio (PAR), the obtained $p < 0.05$ (Kruskal-Willis) for all concentrations. This showed that there was a significant difference between the PAR values generated from the three types of plasma. Furthermore, further statistical analysis using the Mann-Whitney method for two types of plasma indicated that the ratio of the citrate-heparin plasma at all concentrations has a $p > 0.05$, indicating no significant difference. In contrast, plasma citrate-EDTA and heparin-EDTA analysis had a value of $p < 0.05$, meaning that there was a significant difference.

Table 1: Comparison of clopidogrel analysis in three plasma samples

Parameter analysis	Types of plasma			p	Explanation
	Citrate	Heparin	EDTA		
Peak area ratio(±SD)					
1 LLOQ	0.1931±0.02	0.1833±0.03	0.1230±0.01	<0.05	Citrate-EDTA; Heparin-EDTA
2 QCL	0.2509±0.02	0.2448±0.04	0.1635±0.01	<0.05	Citrate-EDTA; Heparin-EDTA
3 QCM	2.0008±0.19	2.1025±0.35	1.3953±0.13	<0.05	Citrate-EDTA; Heparin-EDTA
4 QCH	3.0671±0.30	3.2186±0.55	2.1979±0.15	<0.05	Citrate-EDTA; Heparin-EDTA
Recovery (%±SD)					
1 QCL	68.27±8.31	70.49±5.64	45.20±1.61	<0.05	There is significant difference
2 QCH	64.43±6.17	69.45±9.51	49.17±5.23	<0.05	There is significant difference
Stability in±25°C (short-term stability)					
1 QCL	Minimum 24 h	Minimum 24 h	Minimum 24 h	-	-
2 QCH	Minimum 24 h	Minimum 24 h	Minimum 24 h	-	-
Stability post-preparation (autosampler)					
1 QCL	Minimum 24 h	Minimum 24 h	Minimum 24 h	-	-
2 QCH	Minimum 24 h	Minimum 24 h	Minimum 24 h	-	-
Freeze-thaw stability					
1 QCL	Minimum 3 cycles	Minimum 3 cycles	Minimum 3 cycles	-	-
2 QCH	Minimum 3 cycles	Minimum 3 cycles	Minimum 3 cycles	-	-
Stability in temperature-20°C (Long-term stability)					
1 QCL	Minimum 28 days	Minimum 21 days	Minimum 21 days	-	-
2 QCH	Minimum 28 days	Minimum 21 days	Minimum 21 days	-	-

LLOQ: Lower limit of quantitation, EDTA: Ethylenediaminetetraacetic acid, SD: Standard deviation, QCL: Quality control low, QCM: Quality control medium, QCH: Quality control high

Comparison of the recovery values of all three plasmas

Observations were made statistically on the concentrations of quality control low (QCL) and Quality control high (QCH). At the concentration of QCL and QCH for the recovery for the three types of plasma, obtained a $p < 0.05$, indicating a significant difference. In the comparative analysis of the recovery values obtained statistically for the concentration of QCL and QCH, $p < 0.05$ also indicated significant differences. This may be due to the very small concentrations of the analysis, which were in the picogram per milliliter range, thus the recovery of each plasma varied.

Comparison of clopidogrel stability for all three plasma samples

Based on the results of the study (Table 1), clopidogrel was stable in plasma citrate, plasma heparin, and plasma EDTA for short-term storage and autosampler. For long-term stability stored at -20°C , clopidogrel was stable for at least 21 days in all three types of plasma. Based on the stability data, there were no significant differences in the stability of clopidogrel in plasma citrate, heparin, and EDTA.

CONCLUSION

The analytical method met the requirements based on EMEA guidelines from 2011 for all three types of plasma (citrate, heparin, and EDTA). In addition, clopidogrel analysis in plasma citrate or heparin provides better results than in plasma EDTA.

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CONFLICTS OF INTEREST

All authors have none to declare.

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