

THIOLATION OF FENUGREEK SEED POLYSACCHARIDE; UTILIZATION AS A NOVEL BIOMUCOADHESIVE AGENT IN DRUG DELIVERY

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ABSTRACT

Objective: The objectives of the present work were to carry out thiol-modification (i.e., thiolation) of fenugreek polysaccharide (FP) and to assess the synthesized thiolated product (TFP) as a mucoadhesive excipient in the designing of mucoadhesive formulations (metronidazole gels and metronidazole buccal discs).

Methods: Extracted CG was thiol-modified via an esterification reaction by utilizing thioglycolic acid with an acidic milieu (using hydrochloric acid). Metronidazole mucoadhesive gels and buccal discs made of extracted FP and TFP (as mucoadhesive excipients) were prepared and evaluated to assess their biomucoadhesivity. Mucoadhesive gels containing 1%w/v metronidazole were prepared using both FP and TFP (1% w/v), separately. Mucoadhesive buccal discs containing metronidazole were prepared by the compression method, where FP and TFP (100 mg) were used as mucoadhesive excipients separately, along with 50 mg lactose and 25 mg PEG 4000.

Results: The yield of TFP was 53.46% and the content of the thiol group in TFP was found to be 5.18 mmol of thiol group/g of FP. FTIR analysis results indicated the thiolation of FP in the synthesized TFP. Both types of formulations (mucoadhesive gels and buccal discs) made of TFP exhibited excellent improved *ex vivo* biomucoadhesion and a sustained pattern of metronidazole release over a prolonged period.

Conclusion: The synthesized TFP can be used as improved mucoadhesive agent in the designing of biomucoadhesive systems for drug delivery.

Keywords: Mucoadhesive polymer, Plant polysaccharide, Fenugreek polysaccharide, Drug delivery, Buccal disc

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INTRODUCTION

Plant-derived polysaccharides and their chemical derivatives have widely been employed in many industrial applications, including foods, pharmaceuticals, biomedical, water purification, etc. [1-3]. The main causes of the gaining popularity of the use of plant polysaccharides in pharmaceutical applications include easy availability from plant resources, economic extraction methodologies, hydrophilicity, nontoxicity, and biodegradability [4, 5]. In addition, their diverse molecular structures and the availability of functional chemical moieties in their molecular structures facilitate chemical modifications or functionalization like cross-linking, carboxymethylation, esterification, sulfation, phosphorylation, thiolation, grafting, etc., in order to introduce or improve derived functional properties [6-9]. Amongst these chemical modifications, a thiolation is a well-recognized approach to polysaccharide modification that imparts improvement of mucoadhesion of polysaccharides [10, 11]. Thus, the above-mentioned chemical modifications can improve the applicability of different plant polysaccharides.

The term 'Biomucoadhesion' describes the adhesion of any system to the biological mucous membrane [10]. Mucoadhesive polymers are capable of prolonging contact as well as residence time when they are incorporated as mucoadhesive excipients in drug delivery systems [12]. On hydration, the mucoadhesive drug delivery system becomes adhesive and, therefore, can be utilized to release drugs over a prolonged period [12, 13]. In general, mucoadhesivity of the majority of polymers is associated with the formation of different kinds of non-covalent bonds like ionic interaction, hydrogen bonds, and Van der's Wall, which are weak bonds in nature [14, 15]. Thus, these non-covalent bonds are not capable of providing high mucoadhesive contact and prolonging residence time [10, 11]. As a result, the drug delivery system is unable to adhere to the mucous membrane of the target site over a prolonged period.

To improve the mucoadhesivity of polymers, the chemical modification of polymers via the introduction of a thiol group (i.e., thiolation) has already been investigated and reported [16]. Thiol groups are capable of forming stronger covalent bonds (di-sulfide

linkage) when they come in contact with glycoproteins of the mucus [17]. The formation of stronger covalent bonds in-between thiolated mucoadhesive polymers and mucus glycoproteins guarantees the improvement of biomucoadhesivity, and this facilitates the localization of drug delivery systems at target sites. In the last few years, different plant polysaccharides, such as tamarind seed polysaccharides [14], tamarind xyloglucan [18], moringa exudate gum [15], karaya gum [19], etc., have been chemically modified via thiolation to improve their biomucoadhesivity.

Fenugreek seed polysaccharide (FP) extracted from methi (*Trigonella foenum-graecum* L.; family-Leguminosae) seeds is a heterogenous polysaccharide with a β (1 \rightarrow 4)-D mannan backbone and a D-galactose branch connected [α -(1 \rightarrow 6)] at the C-6 position with galactose to mannose ratio of 1: 1 [20]. It has already been reported as a suspending agent [21], tablet-binder [22], disintegrating agent [23], gelling agent [24], sustained release agent [25, 26], and so on. Our research group already reported the usefulness of FP as a plant-derived mucoadhesive agent in the designing of various biomucoadhesive beads [25-27]. The molecular structure of FP contains-OH groups to side-chain and these-OH groups are capable of interacting with the mucosal glycoprotein contributing to its biomucoadhesivity. Amidation of FP has already been done to improve biomucoadhesivity and is reported [20]. In the current research, the introduction of thiol groups within the molecular structure of FP (i.e., thiolation of FP) was planned to improve biomucoadhesion of FP, which has not been reported till date. Therefore, the current research is novel of its kind. The current research attempted to synthesize thiolated FP (TFP) using thioglycolic acid and to evaluate the use of synthesized TFP as an improved biomucoadhesive excipient in two different kinds of drug delivery formulations—metronidazole biomucoadhesive gels and metronidazole biomucoadhesive buccal discs.

MATERIALS AND METHODS

Materials

Metronidazole (obtained from B. S. Traders, India) was used as a model drug. FP was extracted from ripe and matured fenugreek

seeds, which were collected from Baripadamarket in the month of March, 2019. Thioglycolic acid (99%), Carbopol 974P, Lactose, PEG 4000, L-cysteine, triethanolamine, Ellman's reagent [DTNB; 5,5'-dithiobis (2-nitrobenzoic acid)] were procured from Hi-Media Laboratories Pvt. Ltd., India. All other reagents, chemicals and solvents used in this research were of analytical grade and commercially available.

Methods

Extraction of FP from fenugreek seeds

FP was extracted from ripe and matured fenugreek seeds. Hot water extraction method was used for the extraction of polysaccharide from fenugreek seeds [28, 29]. Fenugreek seeds (500 g) were soaked in demineralized water (4 L) overnight and boiled using a temperature-controlled electric water bath to prepare a slurry. After cooling, the cooled slurry was kept in a refrigerator for 12 h to settle out the undissolved materials. The cooled solution (at room temperature) was added to thrice the volume of acetone with continuous stirring. The precipitate was formed and the formed precipitate was repeatedly rinsed using acetone and dried at 45 ± 2 °C in a tray drier for 24 h. The extracted material was ground by using a clean pestle mortar and was then passed through the sieve 80. The extracted FP was then stored in an air-tight desiccator for further experiment.

Preparation of TFP by esterification

Thiolation of extracted FP was carried out via esterification employing thioglycolic acid ($\text{HSCH}_2\text{CO}_2\text{H}$) and hydrochloric acid (HCl) [15, 30]. In distilled water (50 ml), FP (12 g) was dissolved to prepare an aqueous FP solution. To the prepared FP solution, 7.2 ml $\text{HSCH}_2\text{CO}_2\text{H}$ and 4 ml of 7N HCl were added. Then, the mixture of 12 g FP, 7.2 ml $\text{HSCH}_2\text{CO}_2\text{H}$ and 4 ml of 7N HCl was allowed for esterification at a temperature of 80 °C for 3 h. The above reaction mixture was cooled, and the cooled reaction mixture was poured into 1 L of acetone. A creamy-white precipitate of the thiolated

product was washed thrice with acetone to eradicate unreacted $\text{HSCH}_2\text{CO}_2\text{H}$ and then dried in a tray drier at a temperature of 50 °C.

Characterization of TFP

Determination of thiol group contents by Ellman's method

The number of thiol group substitutions in TFP was determined by quantifying the thiol group amount in both extracted FP and synthesized TFP by Ellman's method [14, 15]. In brief, a 0.1% w/v aqueous solution of TFP was prepared in 5 M phosphate buffer (pH 8) and the prepared solution was incubated with a 0.03% w/v solution of Ellman's reagent (DTNB, in 5 M phosphate buffer, pH 8) at room temperature for 2 h, followed by measurement of the absorbance of the reaction mixture at 450 nm. The thiol group substitution in the synthesized TFP was determined by measuring the absorbance of the reaction mixture at 450 nm. The number of thiol group substitutions (Thiol group/g) in TFP was calculated using the calibration curve prepared by the reaction of standard solution of L-cysteine with Ellman's reagent.

Confirmatory study of thiol group by fourier transform infrared (FTIR) spectroscopy

Extracted FP (extracted from ripe and matured fenugreek seeds) and synthesized TFP samples (by thiolation reaction) were subjected to FTIR spectroscopy analyses in a FTIR spectrophotometer (Perkin-Elmer Spectrum RX I, USA) as potassium bromide (KBr, IR-grade) pellets (containing samples) in a scan range of 4000–400 cm^{-1} .

Preparation of metronidazole gels made of FP and TFP by aqueous dispersion method

Gel was prepared by the aqueous dispersion method [14, 30]. In brief, 1.5% w/v Carbopol 974P was dispersed into the aqueous solution of 1% w/v metronidazole containing 1% w/v FP or TFP (separately). The mixture solutions are allowed to sufficiently hydrate overnight, followed by the addition of triethanolamine (1% w/v) to form FP-based metronidazole-containing gels. Table 1 shows the compositions of metronidazole gels made of FP and TFP.

Table 1: Compositions of metronidazole gels made of FP and TFP

Code	FP (% w/v)	TFP (% w/v)	Carbopol 974P (% w/v)	Triethanolamine (% w/v)	Metronidazole (% w/v)
G-FP	1	-	1.50	1	1
G-TFP	-	1	1.50	1	1

Evaluation of metronidazole gels made of FP and TFP

Measurement of viscosity

The viscosity of these formulated metronidazole gels containing FP and TFP was measured using a Brookfield viscometer (Brookfield DV-E Viscometer) at different speeds with spindle 6 at room temperature.

Evaluation of ex vivo biomucoadhesivity

The *ex vivo* biomucoadhesivity of these formulated metronidazole gels containing FP and TFP was evaluated using a modified physical balance comprising of a tared 2-arm balance, one side of which contained 2-glass plates: the lower plate was permanently fitted to the basement of the one-arm of the base and the upper plate was glued to the one-arm of the modified balance [31, 32]. In this study, the biological membrane used for biomucoadhesivity testing was goat intestinal mucosal membrane. The excised and fresh goat intestinal mucosal membrane was fixed with glue (cyanoacrylate adhesive) to the upper plate. Metronidazole gel (1 g) was placed on the upper portion of the lower plate. The upper plate was positioned over the lower plate and on initial pressure as preload was applied by finger-tip for 5 min. After the removal of the preload force, a gradually increasing weight was applied to the second arm (of the modified physical balance). The weight (in g) required to detach the lower plate from the upper plate was noted as mucoadhesive strength (shear stress). Furthermore, the force of adhesion (in N) and bonding strength (in N/M^2) of metronidazole gels were calculated employing the following formula [33]:

$$\text{Force of adhesion (N)} = \text{Mucoadhesivestrength} \times 9.81/1000$$

$$\text{Bonding strength} = \frac{\text{Force of adhesion}}{\text{Surface area of mucosal membrane}}$$

In vitro drug release study

Drug release was studied in a dialysis bag using a phosphate buffer solution [34]. In brief, 1 g of metronidazole gels made of FP and TFP was sampled individually for an *in vitro* drug release study using a dialysis sac (molecular weight cut off of 10 kDa) in a USP type-II dissolution apparatus (Campbel Electronics, India). Accurately weighed metronidazole gels were individually placed within the dialysis sac and then tied with the paddle of the dissolution apparatus. The paddle was then immersed in the release medium, which was phosphate buffer at pH 6.8. The whole dissolution system was maintained at 37 ± 0.5 °C temperature with a paddle rotation of 50 rpm speed. Aliquots (5 ml) were sampled at various time intervals; freshly prepared release medium (phosphate buffer, pH 6.8) of the same volume (*i.e.*, 5 ml) was added at the same time points of sample withdrawal. The released metronidazole contents from the gels were estimated by an UV-VIS spectrophotometer (Shimadzu, Japan) at a wavelength of 320 nm.

Preparation of metronidazole buccal discs made of FP and TFP

Metronidazole buccal discs using of FP and TFP were prepared by direct compression method [35, 36]. Powdered blends of FP or TFP (100 mg), metronidazole (100 mg), lactose (50 mg) and PEG 4000 (25 mg) were prepared and compressed at 75 kg/cm^2 for 1 min using an IR hydraulic press to produce discs of 13 mm diameter. Table 2 shows the compositions of metronidazole buccal discs made of FP and TFP.

Table 2: Compositions of metronidazole buccal discs made of FP and TFP

Code	FP (mg)	TFP (mg)	Lactose (mg)	PEG 4000 (mg)	Metronidazole (mg)
G-FP	75	-	75	25	10
G-TFP	-	75	75	25	10

Evaluation of metronidazole buccal discs made of FP and TFP

Measurement of weight uniformity

The weight uniformity of metronidazole buccal discs made of FP and TFP was measured by weighing the buccal discs (20 nos.) individually from each formulation using an electronic balance (Mettler Toledo). The average weight and standard deviation were calculated. The weight variation (%) was calculated as [37]:

$$\text{Weight variation (\%)} = \frac{\text{Standard deviation}}{\text{Mean weight}} \times 100$$

Measurement of thickness uniformity

The thickness uniformity of metronidazole buccal discs made of FP and TFP was measured using a digital slide caliper (Mitutoyo Corporation, Japan).

Measurement of content uniformity

The uniformity of drug contents in these metronidazole buccal discs made of FP and TFP was measured by estimating the metronidazole contents in 6 buccal discs from each formulation. The buccal discs were crushed by using a set of clean pestle and mortar. The powdered sample (100 mg) was dissolved in phosphate buffer, pH 6.8, filtered by using Whatman® filter paper (No. 40) and appropriately diluted with freshly prepared phosphate buffer, pH 6.8. The metronidazole contents in the filtrates were determined by measuring the absorbance at 320 nm in a UV-VIS spectrophotometer (Shimadzu, Japan).

Measurement of friability

The friability of metronidazole buccal discs made of FP and TFP was measured by the Friability tester USP 23 (Electro Lab, India). For each formulation, 6 buccal discs were weighed and placed in a friabilator. These discs were rotated for 4 min at 25 rpm. The discs were dedusted and reweighed. The friability (%) was computed as the weight loss (%) using the following formula [36]:

$$\% \text{ Friability} = \frac{(I - F) \times 100}{I}$$

where, I = Initial weight and, F = Weight after friability

Evaluation of biomucoadhesivity

The *ex vivo* biomucoadhesivity of metronidazole buccal discs made of FP and TFP was assessed by measuring their biomucoadhesion

time using goat buccal mucosal membrane [15, 36]. Goat buccal pouch collected within an hour of slaughter was cleaned by removing underlying fat materials and then thoroughly ringed using simulated saliva fluid (pH 6.8). The washed mucosal membrane was tied to the paddle of the USP type-II dissolution apparatus. One buccal disc was pasted on the inner portion of the mucosal membrane, employing a light force for about 30 sec. The *ex vivo* mucoadhesion time was then calculated by estimating the time required for the buccal discs to detach or erode from the mucosal surface.

In vitro drug release study

In vitro release of metronidazole from buccal discs made of FP and TFP was performed using a USP type-II dissolution apparatus (Campbel Electronics, India) [15, 36]. The dissolution medium used was 250 ml simulated saliva fluid (pH 6.8), which was maintained at 37 ± 0.5 °C temperature with a paddle rotation of 50 rpm speed. The buccal disc was attached with cyanoacrylate glue to the inner portion of the dissolution vessel. At an appropriate interval, aliquots (5 ml) were sampled and then replaced with freshly prepared 5 ml of simulated saliva fluid (pH 6.8) at the same time points of sample withdrawal. The released metronidazole contents from the buccal discs were estimated by an UV-VIS spectrophotometer (Shimadzu, Japan) at a wavelength of 320 nm.

Statistical analysis

The mean standard deviation (SD) was used to express all data. The simple statistical analysis was conducted using Microsoft Excel 2002.

RESULTS AND DISCUSSION

Extraction of FP and TFP

FP is reported as a plant-derived heterogenous polysaccharide possessing a galactomannan structure [20]. The yield of extracted FP was found to be 18.44%. In the current study, the chemical esterification reaction in between the -OH groups of the galactomannan structure of FP and a the-COOH group of $\text{HSCH}_2\text{CO}_2\text{H}$ was performed under the acidic milieu to achieve the TFP (fig. 1). The yield of TFP was found to be 53.46%. The synthesized TFP was creamy-white in colour and soluble in cold as well as hot water.

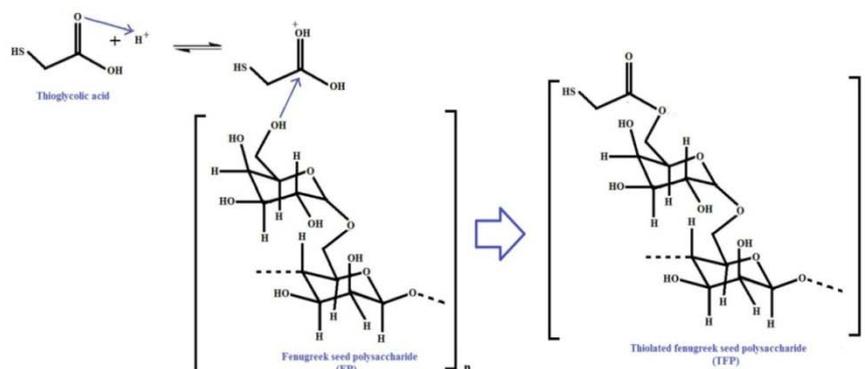


Fig. 1: Schematic representation of thiolation of FP using $\text{HSCH}_2\text{CO}_2\text{H}$

Characterization of TFP

Thiol group contents

The thiol group content in TFP was found 5.18 mmol of thiol group/g of FP as determined.

FTIR

The FTIR spectrum of unmodified FP (extracted) and thiol-modified FP (i.e., TFP) is presented in fig. 2. FTIR spectra of FP (fig. 2a) showed characteristic peaks in-between $3500\text{-}3150\text{ cm}^{-1}$ for -OH stretching

vibration, at 2923.17 cm^{-1} for -CH stretching vibration, at 1637.99 cm^{-1} for scissoring vibration of -OH bond, and at 1382.03 cm^{-1} for symmetrical deformation of -CH₂ groups. In addition, a series of characteristic peaks were noticed in-between 1150–800 cm^{-1} representing highly coupled C-C-O, C-OH, and C-O-C stretching of the polymeric backbone of FP molecular structure. This result is almost similar to the previous researches reported [20, 25-27]. In the FTIR spectrum of TFP (fig. 2b), various characteristic peaks of the unmodified FP (extracted) were found to appear with very minute/without any significant shifting. In particular, characteristic peaks at 2925.80 cm^{-1} for -CH stretching vibration, at 1637.86 cm^{-1} for

scissoring vibration of -OH bond, and at 1387.57 cm^{-1} for symmetrical deformation of -CH₂ groups were observed in the FTIR spectrum of TFP. In addition, analogous patterns of characteristic peaks in-between 3500-3150 cm^{-1} representing -OH stretching vibration along with a series of characteristic peaks in-between 1150-800 cm^{-1} representing highly coupled C-C-O, C-OH, and C-O-C stretching of the polymeric backbone were also noticed in the FTIR spectrum of TFP. However, a weak shoulder at 2558.02 cm^{-1} due to the -SH stretch of the thiol group was detected in the FTIR spectrum of TFP. Recent research has also found that the characteristic peaks representing the thiol group are only weakly detectable in FTIR spectroscopy [15, 38].

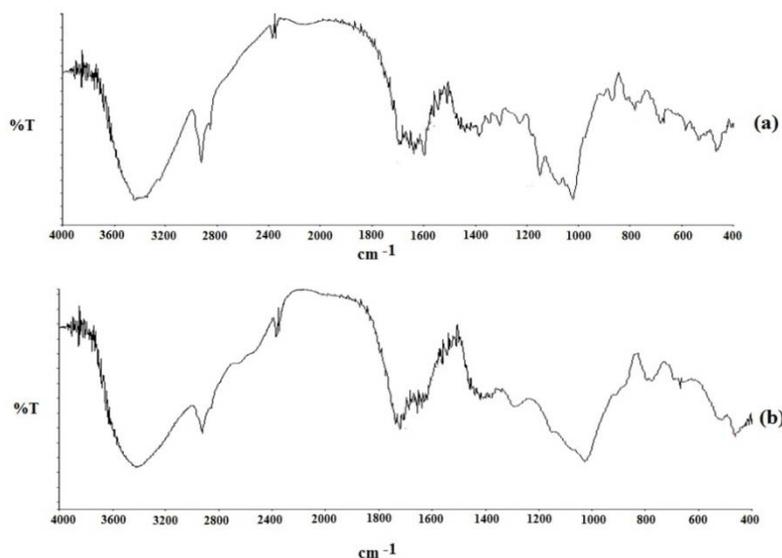


Fig. 2: FTIR spectrum of (a) unmodified FP and (b) thiol-modified FP (TFP)

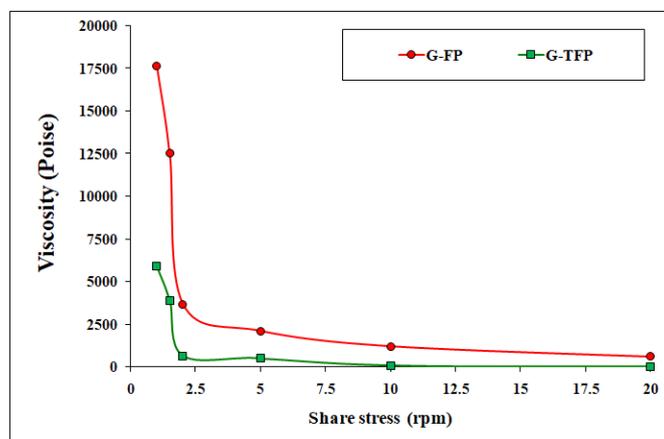


Fig. 3: The comparative viscosity results of metronidazole gels made of FP and TFP (G-FP and G-TFP)

Preparation of metronidazole gels made of FP and TFP

To evaluate the mucoadhesive potential of TFP, mucoadhesive gels were formulated, where metronidazole (1% w/w) was used as a model drug. In these gels (G-FP and G-TFP), 1.5% w/v Carbopol 974P was used as a gel-forming agent, whereas 1% w/v FP and thiolated FP were employed as biomucoadhesive agents. In addition, triethanolamine (1% w/v) was used to form gels.

Evaluation of metronidazole gels made of FP and TFP

Viscosity

The viscosity of all these metronidazole gels was measured by a Brookfield viscometer (Brookfield DV-E Viscometer). The

comparative viscosity results of metronidazole gels made of FP and TFP (G-FP and G-TFP) are presented in fig. 3. It was observed that the viscosity of G-FP metronidazole gel made of FP as a mucoadhesive agent was higher than that of G-TFP metronidazole gel made of TFP. Similar result was also noticed in a work reported where thiolated tamarind seed polysaccharide was investigated as a mucoadhesive agent by formulating metronidazole gel [14].

Biomucoadhesivity

The *ex vivo* biomucoadhesion of these metronidazole gels (1 g) made of FP and TFP (G-FP and G-TFP) onto excised goat buccal mucosa was evaluated by modified physical balance. The *ex vivo* mucoadhesivity results (mucoadhesive strength, force of adhesion,

and bonding strength) of metronidazole gels made of FP and TFP (G-FP and G-TFP) are presented in table 3. The measured *ex vivo* mucoadhesive strength, force of adhesion, and bonding strength of G-TFP metronidazole gel made of TFP were higher (more than 1.75-folds) than those of G-FP metronidazole gel made of unmodified FP. The biomucoadhesivity of G-FP metronidazole gel can be attributed to the fact that the hydroxyl groups occurred in the molecular structure of FP from weaker non-covalent bonds like hydrogen

bonds, Vander's Wall forces, and ionic interactions with the mucous glycoproteins [15]. In contrast, the enhanced biomucoadhesivity by the G-TFP metronidazole gels made of TFP can be attributed to the fact of the ability of thiol groups (occurred in the synthesized TFP) to form stronger covalent bonds (di-sulfide linkage) in contact with glycoproteins of mucous. Therefore, TFP can be used as an improved biomucoadhesive excipient in biomucoadhesive gel-based drug delivery systems.

Table 3: *Ex vivo* mucoadhesivity results (mucoadhesive strength, force of adhesion and bonding strength) of metronidazole gels made of FP and TFP (G-FP and G-TFP)

Code	Mucoadhesive strength (g) ^a	Force of adhesion (N)	Bonding strength (N/m ²)
G-FP	12.98±0.16	0.1273	448.23
G-TFP	22.81±0.24	0.2238	788.02

^aData are expressed as mean±SD; n=3

In vitro drug release

The *in vitro* drug release from metronidazole gels made of FP and TFP (G-FP and G-TFP) was evaluated using a dialysis sac in phosphate buffer, pH 6.8. The comparative *in vitro* drug release from metronidazole gels made of FP and TFP (G-FP and G-TFP) is presented in fig. 4. The G-FP metronidazole gels made of FP released almost all the total drug within the release period of 6 h. The G-TFP metronidazole gel made of TFP exhibited slower sustained drug release than that of the G-FP metronidazole gel made of FP. The

observed results of the *in vitro* drug release study were not found to be consistent with the viscosity results. Even though the G-FP metronidazole gel made of unmodified FP showed a comparatively higher viscosity than that of the G-TFP metronidazole gel made of TFP. Similar result was also noticed in a work reported where thiolated tamarind seed polysaccharide was investigated as a mucoadhesive agent by formulating metronidazole gel [14]. Comparatively slower-sustained drug release from G-TFP metronidazole gel made of TFP could be due to possible interaction of metronidazole with TFP or due to *in situ* cross-linking contributed by di-sulfide linkage of TFP.

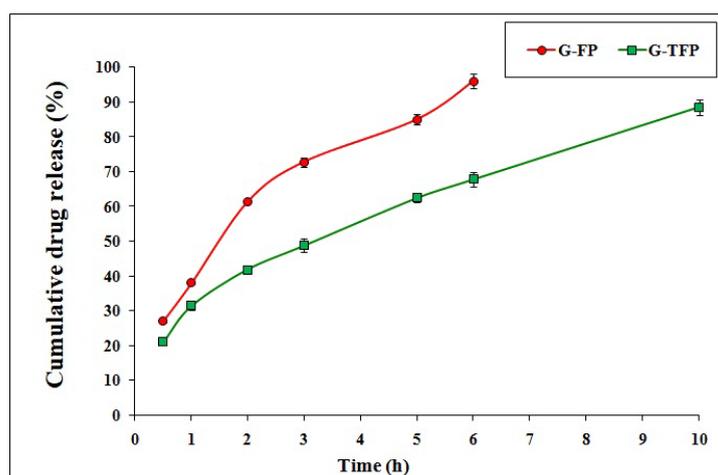


Fig. 4: Comparative *in vitro* drug release from metronidazole gels made of FP and TFP (G-FP and G-TFP) in phosphate buffer, pH 6.8 [Data are expressed as mean±SD; n=3]

To analyze the *in vitro* drug release kinetics along with the drug release mechanism, the *in vitro* drug release data of metronidazole gels made of FP and TFP (G-FP and G-TFP) in phosphate buffer, pH 6.8 was fitted into important mathematical kinetic models. Table 4 presents the model-fitting results for *in vitro* drug release from these metronidazole gels. The *in vitro* drug release from both the metronidazole gels made of FP and TFP (G-FP and G-TFP) followed the Korsmeyer-Peppas model ($R^2 = 0.9872$ for G-FP and $R^2 = 0.9968$ for G-

TFP) as the best-fitting model. However, in both the cases of metronidazole gels, the Higuchi model ($R^2 = 0.9828$ for G-FP and $R^2 = 0.9899$ for G-TFP) was also found to be closer to the best-fitting Korsmeyer-Peppas model. Further, the value of the release exponent (n) of the Korsmeyer-Peppas equation ($n \leq 0.5$) indicated that the mechanism of *in vitro* drug release from both the metronidazole buccal gels made of FP and TFP (G-FP and G-TFP) in phosphate buffer, pH 6.8 was followed by a diffusion-dependent releasing mechanism.

Table 4: The model-fitting results for *in vitro* drug release from these metronidazole gels made of FP and TFP (G-FP and G-TFP)

Code	Zero-order model	First-order model	Higuchi model	Korsmeyer-Peppas model	Release exponent (n)
G-FP	0.9337	0.8403	0.9828	0.9872	0.4910
G-TFP	0.9647	0.8435	0.9899	0.9968	0.4634

Preparation of metronidazole buccal discs made of FP and TFP

In this work, metronidazole buccal discs (13 mm dia.) were formulated using FP or TFP (as mucoadhesive agent), PEG 4000 (as

plasticizer) and lactose (as diluent) via compression employing an IR hydraulic press. Powdered blends of FP or TFP (75 mg), metronidazole (100 mg), lactose (75 mg) and PEG 4000 (25 mg) were compressed at 75 kg/cm² for 1 min.

Evaluation of metronidazole buccal discs made of FP and TFP

Weight uniformity

Weight uniformity results, namely, average weight (mg) and weight variation (%) of metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) are tabulated in table 5. The average weight (mg) of D-FP metronidazole buccal discs made of unmodified FP (191.23±2.30 mg) was almost similar as compared to that of D-TFP metronidazole buccal discs made of TFP (190.41±2.68 mg).

However, the weight variation (%) of D-FP metronidazole buccal discs made of unmodified FP (1.20%) was less as compared to that of D-TFP metronidazole buccal discs made of TFP (1.41%). The weight variation (%) values of both the metronidazole buccal discs were less than 5%. Therefore, the results of the weight uniformity of both the metronidazole buccal discs indicated that powdered FP or TFP, metronidazole, lactose, and PEG 4000 were well blended during the preparation of these metronidazole buccal discs made of FP and TFP (D-FP and D-TFP).

Table 5: The results of weight uniformity (average weight and weight variation), thickness uniformity, content uniformity and friability of metronidazole buccal discs made of FP and TFP (D-FP and D-TFP)

Code	Average weight (mg) ^a	Weight variation (%)	Thickness (mm) ^b	Drug content (%) ^b	Friability (%) ^b
D-FP	191.23±2.30	1.20	1.14±0.02	99.37±2.42	0.82±0.03
D-TFP	190.41±2.68	1.41	1.21±0.04	99.08±2.57	0.96±0.02

^aData are expressed as mean±SD; n = 20, ^bData are expressed as mean±SD; n = 6

Thickness uniformity

The thickness of both the metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) was measured (by digital slide caliper), and the results are tabulated in table 5. The thickness of D-FP metronidazole buccal discs made of unmodified FP (1.14±0.02 mm) was almost similar as compared to that of D-TFP metronidazole buccal discs made of TFP (1.21±0.04 mm). Therefore, the results of the thickness uniformity of both the metronidazole buccal discs indicated that the sizing of buccal discs (13 mm dia.) was appropriately done during preparation, in particular by applying a uniform compression force of 75 kg/cm² for 1 min by IR hydraulic press.

Content uniformity

The drug content uniformity results of both the metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) are tabulated in table 5. The drug content of D-FP metronidazole buccal discs made of unmodified FP (99.37±2.42%) was almost similar as compared to that of D-TFP metronidazole buccal discs made of TFP (99.08±2.57%). Therefore, the drug content uniformity results ascertain the presence of contained metronidazole within the compendia quality limit in both the metronidazole buccal discs made of FP and TFP (D-FP and D-TFP).

Friability

The measurement of friability is usually used to assess the resistance of the solid dosage forms (here, buccal discs) to abrasion. The friability (%) results of both the metronidazole buccal discs

made of FP and TFP (D-FP and D-TFP) are tabulated in table 5. The friability (%) of D-FP metronidazole buccal discs made of unmodified FP (0.82±0.03%) was comparatively less than that of D-TFP metronidazole buccal discs made of TFP (0.96±0.02). Overall, it was found that the friability (%) values of both the metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) were less than 1% w/w, which conformed to the compendial specifications of friability for tablets (i.e., less than 1% w/w).

Biomucoadhesivity

The *ex vivo* biomucoadhesivity of metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) was assessed by measuring their biomucoadhesion time using goat buccal mucosal membrane. The *ex vivo* biomucoadhesion time of D-TFP metronidazole buccal discs made of TFP (12.5 h) was comparatively higher (almost 1.5-fold) than that of D-FP metronidazole buccal discs made of FP (8.2 h). Similarly, enhanced *ex vivo* biomucoadhesivity of G-TFP metronidazole gels made of TFP was observed as compared to that of G-FP metronidazole gels made of unmodified FP. The enhanced *ex vivo* biomucoadhesivity of thiol-modified polysaccharide-based formulations can be attributed to the ability of thiol groups (occurred in the synthesized thiolated polysaccharide) to form stronger covalent bonds (di-sulfide linkage) in contact with glycoproteins of mucous [15, 36]. Therefore, TFP can be used as an improved mucoadhesive excipient in the buccomucoadhesive drug delivery system. Similar results were also noticed in the studies reported where thiolated thiolated moringa gum and thiolated xanthan gum were investigated as a mucoadhesive agent by formulating metronidazole buccal discs [15, 36].

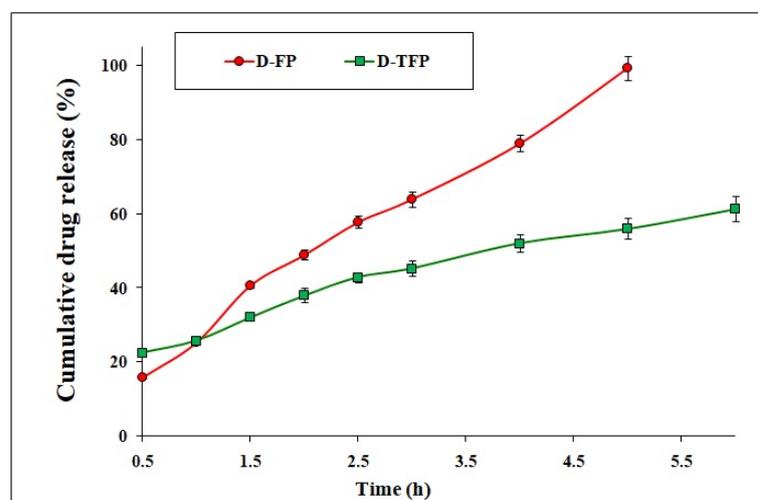


Fig. 5: Comparative *in vitro* drug release from metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) in simulated saliva fluid, pH 6.8 [Data are expressed as mean±SD; n = 3]

In vitro drug release

The *in vitro* drug release from these metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) was evaluated in simulated saliva fluid, pH 6.8. The comparative *in vitro* drug release from metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) is presented in fig. 5. The D-FP metronidazole buccal discs made of unmodified FP released almost all the drug within the release period of 5 h. In contrast, D-TFP metronidazole buccal discs made of TFP showed a sustained pattern of drug release over 24 h. The slower drug release by D-TFP metronidazole buccal discs made of TFP can be explained by the formation of stronger covalent bonds (due to *in situ* cross-linking contributed by di-sulfide linkage) in-between glycoprotein of the mucous and the thiol-group present in TFP (which was incorporated within the formula of D-TFP metronidazole buccal discs).

Furthermore, the *in vitro* drug release data of metronidazole buccal discs made of FP and TFP (D-FP and D-TFP) in simulated

saliva fluid, pH 6.8 was analyzed by fitting into important mathematical kinetic models to reveal the drug-releasing kinetics and mechanism. Table 5 presents the model fitting results for *in vitro* drug release from these metronidazole buccal discs made of FP and TFP (D-FP and D-TFP). The *in vitro* drug release from D-FP metronidazole buccal discs made of FP followed the Korsmeyer-Peppas model ($R^2 = 0.9941$) as the best-fit model. However, it was also closer to the zero-order model release kinetics ($R^2 = 0.9894$). On the other hand, *in vitro* drug release from G-TFP metronidazole buccal discs made of TFP followed the Korsmeyer-Peppas model ($R^2 = 0.9825$). The values of the release exponent (n) of the Korsmeyer-Peppas equation ($n \approx 0.43$) indicated that the *in vitro* drug release from D-FP metronidazole buccal discs followed an anomalous transport mechanism (diffusion and relaxation-dependent releasing), while the G-TFP metronidazole buccal discs made of TFP followed the diffusion-dependent (Fickian) mechanism.

Table 5: The model-fitting results for *in vitro* drug release from these metronidazole buccal discs made of FP and TFP (D-FP and D-TFP)

Code	Zero-order model	First-order model	Higuchi model	Korsmeyer-Peppas model	Release exponent (n)
D-FP	0.9894	0.8777	0.8197	0.9941	0.7953
D-TFP	0.9644	0.9001	0.8676	0.9825	0.4281

CONCLUSION

Natural polysaccharides have numerous open hydroxyl groups in their molecules, which makes them a suitable material for easy attachment of thiol groups. They are also inexpensive, and the process of thiolation is also simple. Gels prepared by FP and TFP showed biomucoadhesion property. However, the biomucoadhesion potential of thiolated products (i.e., TFP) was much higher (1.75 times). From the release study, it was observed that thiolated products showed a slower rate of drug release and, thereby, TFP can be used as a biomaterial for preparing sustained release formulations for application in gum as well as the mouth cavity in the management of oral infections. Thin and sleek buccal discs can also be prepared by quite a simple method of compression. These discs showed uniformity in weight, were non-brittle, and had uniformity in drug content. However, metronidazole buccal discs prepared with a thiolated derivative of polysaccharide showed improved biomucoadhesion potential, which was 1.5 times higher than the metronidazole buccal discs prepared by FP. Discs prepared with TFP showed a slow and uniform release of metronidazole. This quality is essential when the sustained release of a drug is required for a long period of time. Therefore, these biomucoadhesive discs can be used for gastroretentive drug delivery.

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AUTHORS CONTRIBUTIONS

All the authors have contributed equally.

CONFLICTS OF INTERESTS

The authors declare no conflict of interest

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