

## SYNTHESIS, CHARACTERIZATION AND EVALUATION OF NOVEL 6- SUBSTITUTED BENZIMIDAZOLE-2-CARBAMATES AS POTENTIAL ANTIMICROBIAL AGENTS

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### ABSTRACT

**Objective:** The aim of the present work was to synthesize and characterize a series of 6- substituted benzimidazole-2-carbamic acid derivatives and evaluate them for antimicrobial activity.

**Methods:** A novel series of 6-substituted benzimidazole-2-carbamic acid derivatives (6a-f) were synthesized by reacting substituted benzyl chlorides (2a-c) with 4-hydroxy-2-nitro aniline (3) in the presence of anhydrous  $K_2CO_3$  in acetone. The series of 4-benzyl oxy-2-nitro anilines (4a-c) thus obtained were reduced with Raney nickel as catalyst. The 4-substituted-1, 2-phenylene diamines were then reacted with 1, 3-bis (alkoxy carbonyl)-S-methyl isothiourea (5a-b) to yield the final compounds (6a-f). The structures of synthesized compounds were established by  $^1H$  NMR and mass spectroscopy. The synthesized compounds were screened for antibacterial and antifungal activity against five pathogenic bacterial strains and two pathogenic fungal strains.

**Results:** All compounds showed moderate to good antibacterial and antifungal activity and were nearly effective as the standard antibacterial and antifungal drugs used for comparison. The exhibited good activity against gram negative bacterial strains like *K. pneumoniae* and *P. aeruginosa* as well as against fungal strain *C. albicans*.

**Conclusion:** The 6-substituted benzimidazole-2-carbamates. compounds synthesized and evaluated for pharmacological activity are promising as antimicrobial agents and are amenable for further optimization.

**Keywords:** Benzimidazole-2-carbamic acid,  $\beta$ -tubulin, Antibacterial, Antifungal.

### INTRODUCTION

Molecules containing benzimidazole ring system have been extensively researched and reviewed. Compounds containing benzimidazole moiety have been reported to possess varied pharmacological activities such as antiulcer, anti-hypertensive, antimicrobial, antihistaminic, and anti-psychotic [1-4]. A vast number of benzimidazole derivatives bearing the substituent either on the benzene ring or imidazole nucleus have been synthesized leading to novel, biologically active molecules. Noteworthy amongst these are the 2-substituted benzimidazoles possessing antifungal activity [5, 6]. Substitution at 2-position on benzimidazole nucleus enhances antifungal activity manifold. Some of these derivatives contain an alkyl carbamate function at 2-position with the alkyl group containing 1-7 carbon atoms. Some examples of molecules belonging to this category are carbendazim, benomyl, cypendazole and have been put in to commercial use for controlling plant fungal diseases and are of economic importance [7-9]. These fungicides interact with  $\beta$ -tubulin protein and inhibit spindle formation thus accounting for their anti-mitotic activity resulting in cell death [10]. The widespread use of these agents has resulted in development of resistant strains of fungi.

Opportunistic fungal infections in humans are a cause of serious concern and none of the benzimidazoles presently available in the market are effective in humans. The 2-substituted benzimidazoles bearing a suitable substituent in the benzene ring have been reported in literature to possess significant antibacterial activity [11-14]. Benzimidazoles compete with purines due to their structural similarity, thus resulting in inhibition of synthesis of bacterial nucleic acids and proteins. The incidence of microbial infections has assumed alarming proportions worldwide because of the indiscriminate use of existing antimicrobial agents which resulted in drug resistance. Lack of newer antimicrobial agents which are efficacious has increased morbidity, mortality and health care costs. In order to overcome these challenges, there is a greater need to develop novel and effective molecules possessing

antibacterial and antifungal activity. The present work is aimed towards developing novel benzimidazole-2-carbamic acid derivatives bearing suitable substituents at 5(6) position and with improved potential for treating bacterial and fungal infections. A critical analysis of the structures of known antibacterial agents with benzimidazole as the central moiety indicated the importance of substitution at 2 and 5(6) positions for retaining antibacterial/antifungal activity. Hence in the designing of novel molecules emphasis was on varying the nature of substituents at 2 and 5(6) positions on benzimidazole ring. This led to designing novel 6-substituted benzyl oxy benzimidazole-2-carbamates with simultaneously changing the carbamate alkyl group to study its contribution to activity. The nature of substituents on the benzyl group was chosen carefully again based on the literature data available on existing derivatives. The newly designed molecules were then synthesized by efficient synthetic process with emphasis on reducing side products. The synthesized compounds were purified, characterized by  $^1H$  NMR and mass spectroscopy and evaluated for antibacterial and antifungal activity.

### MATERIALS AND METHODS

All synthesized compounds were purified by a simple recrystallization process. The melting points were recorded in open capillary tube and are uncorrected.  $^1H$  NMR spectra were recorded on 500 MHz Bruker NMR spectrometer using  $CDCl_3$  or  $DMSO-D_6$  as solvent and tetra methyl silane was used as reference. Mass spectra were obtained using ESI LC-MS Shimadzu mass spectrometer. All reactions were monitored by thin layer chromatography using precoated silica gel GF<sub>254</sub> aluminium plates purchased from E. Merck. Other chemicals were procured from commercial sources and used without further purification.

### Experimental Procedure

The substituted benzimidazole-2-carbamates were synthesized by reacting substituted 1, 2-phenylene diamine with thiourea synthon generated *in situ* as shown in Table 1 [15-17].

**Table 1: It shows thiourea synthon for introduction of carbamate group at 2-position in benzimidazole**

S. No.	Synthon	Structure	Reference
1.	1-methoxy carbonyl-s-methyl iso thiourea		[15]
2.	1,3-Bis-(alkoxy carbonyl)-S-methyl isothiurea		[16,17]

Appropriately substituted benzyl alcohols (1a-c) were used as starting materials. The benzyl alcohols were reacted with thionyl chloride in diethyl ether at 0° C and gradually stirred at room temperature for 2-24 hours to yield the corresponding substituted benzyl chlorides (2a-c). The reaction of the benzyl chlorides (2a-c) with 4-hydroxy -2- nitro aniline (3) in refluxing acetone in presence of anhydrous K<sub>2</sub>CO<sub>3</sub> gave 4-benzyl oxy -2-nitro anilines (4a-c). The nitro groups in 4a-c were then reduced using hydrazine hydrate and Raney nickel as catalyst. The 4-substituted 1, 2-phenylene diamines were obtained as a solution in ethanol by filtering the reaction mixture. 1, 3-bis (alkoxy carbonyl)-S-methyl isothiurea derivatives (5a-b) were generated *in situ* using S-methyl isothiuronium sulfate and alkyl chloroformate in the presence of KOH/NaOH. Finally the reaction of 4-substituted 1, 2-phenylene diamines with compounds 5a-b in the presence of refluxing glacial acetic acid gave 6-substituted benzimidazole derivatives (6a-f) having an alkyl carbamate group at the 2-position (scheme 1).

#### General procedure for synthesis of 4-benzyl oxy-2-nitro anilines (4a-c)

A mixture of 4-hydroxy -2- nitro aniline (3) (2.31 g, 0.015 mol), anhydrous potassium carbonate (3.5 g, 0.025 mol) and substituted benzyl chlorides (2a-c) (3.24 g, 0.015 mol) was stirred in 40 ml of dry acetone and heated at reflux for 24-48 hrs. The mixture was cooled and poured into ice. The separated solid was filtered, washed with water and dried. The solid were redissolved in acetone and 0.1N NaOH was added till precipitation was complete, filtered and dried. This gave 4a-c as yellow to orange solids in sufficiently pure form. 4a- M. P. 130° (80.4%) R<sub>f</sub> value 0.62, 4b- M. P. 88° (82.3%) R<sub>f</sub> value 0.48, 4c- M. P. 85° (78.52%) R<sub>f</sub> value 0.43.

#### General procedure for synthesis of 4-substituted-1, 2-phenylene diamines

A stirred suspension of appropriate benzyl oxy derivative (4a-c) (0.44 g, 1.32 mmol) in 10 ml of ethanol was treated with Raney nickel slurry in water (3.0 g). Hydrazine hydrate (3.0 ml, 85 % solution in water) was added drop wise. The mixture was stirred for 45-60 mins. The solution was filtered and the solvent was evaporated. The obtained oil was immediately cyclo-condensed without further purification.

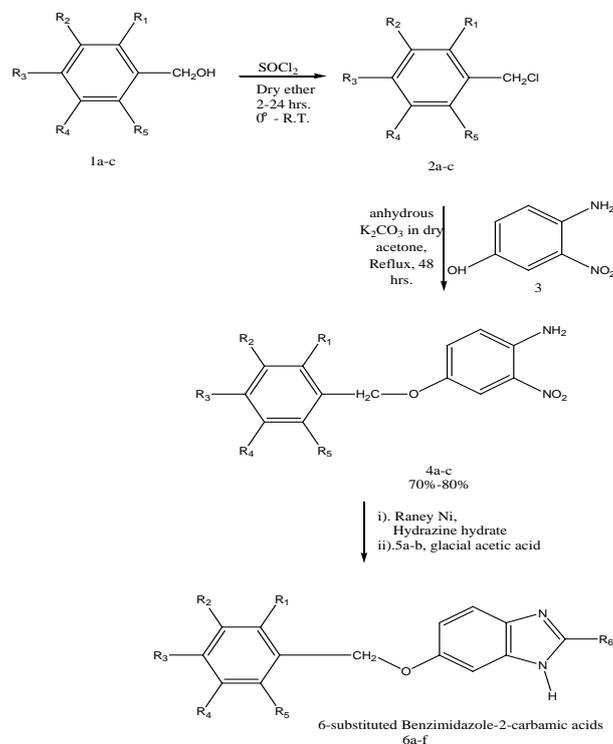
#### General procedure for synthesis of 6-substituted benzyl oxy benzimidazoles (6a-f)

To a solution of S-methyl isothiuronium sulphate in water (0.8 g, 6.06 mmol), the alkyl chloroformate (1.14 g, 12.12 mmol) was added all at once after cooling to 0°C using ice-salt mixture. A solution of 25 % aqueous NaOH/KOH was added drop wise and pH was adjusted to 7-8. The pH and temperature of the reaction was maintained at 7-8 and 0-5°C respectively for about 1 hr 15 mins. The obtained 1,3-bis (alkoxy carbonyl)- S-methyl isothiurea (5a-b) formed *in situ* was treated with glacial acetic acid added drop wise over a period of 15 mins to adjust the pH to 3-4 followed by addition of 4-substituted 1, 2-phenylene diamine (1.8 g, 6.06 mmol) in one portion. The resulting mixture was warmed to 100°C and refluxed for a period of 2-3 hrs. Solid benzimidazole derivatives were isolated by filtration and washed well with water and air-dried. They were then purified by dissolving in chloroform-methanol (9:1). Evaporation of the solvent gave 6a-f in pure form.

#### Anti microbial Evaluation

The synthesized compounds (6a-f) were screened for antibacterial and antifungal activity by cup plate (well diffusion) method [18-20].

The strains of bacteria used were *Staphylococcus aureus* ATCC 9144, *Bacillus subtilis* ATCC 6633, *Escherichia coli* ATCC 25922, *Klebsiella pneumoniae* ATCC 29665 and *Pseudomonas aeruginosa* ATCC 15442.



- 6a: R<sub>1</sub>=R<sub>5</sub>=H, R<sub>2</sub>=R<sub>3</sub>=R<sub>4</sub>=OCH<sub>3</sub>, R<sub>6</sub>=NHCOOCH<sub>3</sub>  
 6b: R<sub>1</sub>=R<sub>5</sub>=H, R<sub>2</sub>=R<sub>3</sub>=R<sub>4</sub>=OCH<sub>3</sub>, R<sub>6</sub>=NHCOOCH<sub>2</sub>CH<sub>3</sub>  
 6c: R<sub>1</sub>=R<sub>2</sub>=OCH<sub>3</sub>, R<sub>3</sub>=R<sub>4</sub>=R<sub>5</sub>=H, R<sub>6</sub>=NHCOOCH<sub>3</sub>  
 6d: R<sub>1</sub>=R<sub>2</sub>=OCH<sub>3</sub>, R<sub>3</sub>=R<sub>4</sub>=R<sub>5</sub>=H, R<sub>6</sub>=NHCOOCH<sub>2</sub>CH<sub>3</sub>  
 6e: R<sub>1</sub>=R<sub>4</sub>=OCH<sub>3</sub>, R<sub>2</sub>=R<sub>3</sub>=R<sub>5</sub>=H, R<sub>6</sub>=NHCOOCH<sub>3</sub>  
 6f: R<sub>1</sub>=R<sub>4</sub>=OCH<sub>3</sub>, R<sub>2</sub>=R<sub>3</sub>=R<sub>5</sub>=H, R<sub>6</sub>=NHCOOCH<sub>2</sub>CH<sub>3</sub>

#### Scheme 1: It shows synthesis of 6-substituted benzimidazole-2-carbamates (6a-f).

The fungal strains used were *Candida albicans* ATCC 2091 and *Aspergillus niger* 16404 and obtained from Food and Drugs Administration, Govt. of Maharashtra, Mumbai. The microbial cultures were characterized using morphology and staining technique prior to use in the screening. The media used for antibacterial testing was nutrient agar. The media used for antifungal testing was Sabouraud's Chloramphenicol Agar. These were procured from HiMedia Pvt. Ltd. Mumbai. The stock culture was maintained on agar slants and stored at 4°C. All the strains were sub cultured on the appropriate agar plate 24 h prior to antimicrobial evaluation of test compounds. The cultures were prepared by inoculating loopful of suspension from freshly prepared slants to nutrient broth (for bacteria) and Sabouraud's Chloramphenicol broth (for fungi) and incubated at 37°C for 24 hrs.

A particular volume of enriched broth was diluted in sterile saline and the turbidity of resulting suspension of organism was matched with 0.5 McFarland standard to contain approximately  $1.5 \times 10^8$  CFU/ml. Antibacterial activity was determined against a standard antibiotic such as Ceftriaxone disodium 0.25mg/ml and antifungal activity was determined using standard antifungal agent fluconazole 0.25 mg/ml. Stock solutions of the test compounds and standards with concentrations 1mg/ml were made in DMSO.

Appropriate dilutions of the test and standard stock solutions were also made in DMSO to yield a final concentration of 0.1 mg/ml for the test compounds and 0.25 mg/ml for the standards used. DMSO was used as control against all the organisms. The minimum inhibitory concentrations were also determined by agar plate method and it was taken to be as the lowest concentration that produces no visible turbidity after incubation. Molten sterile nutrient agar (20 ml) was poured in presterilized Petri plate and allowed to solidify.

The bacterial/fungal suspension (0.1 ml) was spread uniformly using a glass spreader. Wells of standard diameters was made in the nutrient agar medium using a sterile cork borer of 8 mm. The test and standard compounds were introduced in to the wells using micropipette with sterile tips.

The agar plates were incubated at 37°C for 24 hours. The diameter of the zone of inhibition was measured in mm by subtracting the diameter of borer from each reading. The test was carried out in triplicates and the mean value of these readings was recorded.

## RESULTS AND DISCUSSION

The physical constants and spectral characteristics of the key intermediate 4- benzyl oxy-2-nitro anilines (4a-c) are shown in Table 2. Mass spectra and  $^1\text{H}$  NMR confirmed structures of nitro anilines synthesized. Table 3 and Table 4 give the physical constants and the spectral characteristics of 6-substituted benzimidazole-2-carbamates (6a-f). The final compounds corresponded with their m/z mass values in the mass spectrum and  $^1\text{H}$  NMR spectra obtained were also in accordance with the structures envisaged. The results of the antimicrobial activity are shown in Table 5. The compounds exhibited moderate to good activity against the bacterial strains *S. aureus*, *K. pneumoniae* and *P. aeruginosa* and fungal strains *C. albicans* and *A. niger*. They were found to be as effective as the standards Ceftriaxone and fluconazole respectively. The activity of the compounds was found to be poor against *B. subtilis*. Compounds 6b, 6d and 6f showed good activity against *E. coli* whereas compounds 6a, 6c and 6e showed poor activity against this organism. The activity of all the compounds against fungal strain *C. albicans* was found to be slightly better than *A. niger*. The overall antibacterial activities for the gram negative bacterial strains such as *K. pneumoniae* and *P. aeruginosa* were better than *E. coli* and gram positive bacteria. The compounds exhibited moderate activity against *S. aureus* in comparison to *B. subtilis*. The minimum inhibitory concentration for the compounds as shown in table 6 ranged from 50-100 mcg/ml for *S. aureus*, *K. pneumoniae* and *P. aeruginosa*, for *E. coli* it ranged from 75- 125 mcg/ml, for *B. subtilis* the values were somewhat higher ranging from 100-150 mcg/ml whereas for *C. albicans* and *A. niger* it ranged from 25-50 mcg/ml.

**Table 2: It shows the spectral data of 4- benzyl oxy-2-nitro anilines (4a-c).**

Compound No.	$^1\text{H}$ NMR ( $\delta$ in ppm)	Mass spectrum m/z values (% rel. intensity)
4a	$\delta$ 3.6-4.0 (m, 9H, -OCH <sub>3</sub> ), $\delta$ 4.9-5.0 (s, 2H, -CH <sub>2</sub> ), $\delta$ 5.9-6.0 (br. s, 2H, -NH <sub>2</sub> ), $\delta$ 6.8-7.6 (m, 5H, PhH).	333.2 (MH,100)
4b	$\delta$ 3.90 (s, 6H, -OCH <sub>3</sub> ), $\delta$ 5.20(s, 2H, -CH <sub>2</sub> ), $\delta$ 5.80 (br. s, 2H, -NH <sub>2</sub> ), $\delta$ 6.80-7.80 (m, 6H, PhH).	302.8 (MH,100)
4c	$\delta$ 3.70-3.90 (s, 6H, -OCH <sub>3</sub> ), $\delta$ 5.08(s, 2H, -CH <sub>2</sub> Ar), $\delta$ 6.02 (br. s, 2H, -NH <sub>2</sub> ), $\delta$ 6.90-7.00 (m, 6H, PhH).	302.7 (MH,100)

**Table 3: It shows physical constants of substituted Benzimidazoles (6a-f).**

Compound No.	Molecular formula	Molecular Mass	Melting point °C	Percent yield*
6a	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>6</sub>	387	205-207	42
6b	C <sub>20</sub> H <sub>23</sub> N <sub>3</sub> O <sub>6</sub>	401	145-150	40
6c	C <sub>18</sub> H <sub>19</sub> N <sub>3</sub> O <sub>5</sub>	357	198-202	35
6d	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>5</sub>	371	136-138	37
6e	C <sub>18</sub> H <sub>19</sub> N <sub>3</sub> O <sub>5</sub>	357	193-196	42
6f	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>5</sub>	371	134-136	45

\* Percent yield calculated on basis of benzyl oxy derivative (4a-c) used as starting material for obtaining substituted benzimidazoles (6a-f).

**Table 4: It shows spectral data of substituted Benzimidazoles (6a-f).**

Compound No.	$^1\text{H}$ NMR ( $\delta$ in ppm)	Mass spectra m/z values (% rel. intensity)
6a	$\delta$ 3.4 (s,3H, -OCH <sub>3</sub> ), $\delta$ 3.66,3.74 and 3.82 (s, 9H, -OCH <sub>3</sub> ), $\delta$ 4.98(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.78(s,2H,PhH), $\delta$ 7.02-7.03(d,1H,5H), $\delta$ 7.25(s, 1H, 7H), $\delta$ 7.28(s,1H,4H), $\delta$ 11.5 (br. s,1H,-NH)	388.2 (MH <sup>+</sup> ,100)
6b	$\delta$ 1.25 (t,3H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 3.40,3.69 and 3.78 (s, 9H, -OCH <sub>3</sub> ), $\delta$ 4.15(q,2H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 5.00(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.78(m,2H,PhH), $\delta$ 7.02-7.03(d,1H,5H), $\delta$ 7.25(s, 1H, 7H), $\delta$ 7.28(s,1H,4H), $\delta$ 11.4 (br. s 1H,-NH)	403.4 (MH <sup>+</sup> ,100)
6c	$\delta$ 3.72,3.77 and 3.82 (s, 9H, -OCH <sub>3</sub> ), $\delta$ 5.04(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.73-6.77(d,1H,5H), $\delta$ 7.05-7.07(s,3H,PhH), $\delta$ 7.25(s, 1H, 7H), $\delta$ 7.28(s,1H,4H), $\delta$ 11.5 (br. s,1H,-NH)	358.0 (MH <sup>+</sup> ,100)
6d	$\delta$ 1.43-1.48 (t,3H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 3.87 (s, 6H, -OCH <sub>3</sub> ), $\delta$ 4.36-4.43(q,2H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 5.16(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.91(s,1H,5H), $\delta$ 6.93-6.93(s,3H,PhH), $\delta$ 7.05-7.09(s, 1H, 7H), $\delta$ 7.11-7.13(s,1H,4H), $\delta$ 10.56 (br. s,1H,-NH)	372.2 (MH <sup>+</sup> ,100)
6e	$\delta$ 3.69, 3.72 and 3.79 (s, 9H, -OCH <sub>3</sub> ), $\delta$ 5.18(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.77-7.10(m,4H,5H and PhH), $\delta$ 7.24(s,1H,4H), $\delta$ 7.25(s, 1H, 7H), $\delta$ 10.58 (br. s,1H,-NH)	358.0 (MH <sup>+</sup> ,100)
6f	$\delta$ 1.27 (t,3H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 3.78 and 3.73 (s, 6H, -OCH <sub>3</sub> ), $\delta$ 4.17 (q,2H,-CH <sub>2</sub> CH <sub>3</sub> ), $\delta$ 5.27(s,2H,-CH <sub>2</sub> Ar), $\delta$ 6.80-6.93(s,4H,5H and PhH), $\delta$ 7.08(s,1H,7H), $\delta$ 7.14(s, 1H, 74H), $\delta$ 10.62 (br. s,1H,-NH)	372.2 (MH <sup>+</sup> ,100)

Table 5: It shows *in vitro* data for antimicrobial screening of Benzimidazoles (6a-f).

Compd	Zone of Inhibition (mm)						
	<i>S. aureus</i>	<i>B. subtilis</i>	<i>E. coli</i>	<i>K. pneumoniae</i>	<i>P. aeruginosa</i>	<i>C. albicans</i>	<i>A. niger</i>
6a	8±0.50	trace	trace	11±0.35	9±0.20	13±0.30	10±0.12
6b	9±0.20	trace	14±0.21	12±0.35	9±0.50	15±0.35	8±0.50
6c	10±0.30	trace	trace	13±0.50	10±0.35	11±0.50	9±0.20
6d	11±0.32	trace	12±0.21	12±0.12	11±0.50	12±0.50	8±0.50
6e	9±0.20	trace	trace	12±0.12	9±0.20	13±0.21	10±0.35
6f	10±0.32	trace	13±0.50	11±0.35	10±0.20	14±0.21	12±0.20
CEFT	10±0.12	8±0.5	7±0.22	12±0.50	10±0.22	-	-
FLU	-	-	-	-	-	12±0.20	11±0.20

CEFT-Ceftriaxone, FLU-Fluconazole. Zone diameter values are expressed as mean ± s. d.

Table 6: It shows MIC (Minimum Inhibitory Concentration) for antibacterial and antifungal activity of Benzimidazoles (6a-f).

Micro-organisms	Minimum Inhibitory Concentration (mcg/ml)						
	6a	6b	6c	6d	6e	6f	
<i>S. aureus</i> ATCC 9144	50	100	50	75	50	100	
<i>B. subtilis</i> ATCC 6633	150	125	100	125	100	150	
<i>E. coli</i> ATCC 25922	125	75	100	75	100	75	
<i>K. pneumoniae</i> ATCC 29665	50	50	50	50	50	50	
<i>P. aeruginosa</i> ATCC 15442	75	50	50	50	50	50	
<i>C. albicans</i> ATCC 2091	35	25	25	25	25	35	
<i>A. niger</i> ATCC 16404	25	25	50	50	25	50	

CEFT-Ceftriaxone, FLU-Fluconazole

## CONCLUSION

A novel series of 6-substituted Benzimidazole-2-carbamates were synthesized and characterized by spectral analysis. They exhibited moderate to good antibacterial and antifungal activity. These results are promising indeed and the findings of this study imply that these compounds have potential for further development and optimization study.

## CONFLICT OF INTERESTS

Declared None

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