

Original Article

PHYTOCHEMICAL SCREENING AND BIOLOGICAL EVALUATION OF *DYPISIS LEPTOCHEILOS* LEAVES EXTRACT AND MOLECULAR DOCKING STUDY OF THE ISOLATED COMPOUNDS

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ABSTRACT

Objective: phytochemical investigation of the ethyl acetate fraction (EAF) of 80% aqueous methanol extract (AME) of *Dypsis leptocheilos* leaves, in addition to evaluation of the antioxidant, cytotoxic and antimicrobial activities of the AME and EAF. Docking was used to predict and understand cytotoxicity of the isolated compounds.

Methods: The ethyl acetate fraction (EAF) of *Dypsis leptocheilos* leaves was subjected to different chromatographic separation techniques. Structures of the isolated compounds were established by different spectroscopic techniques (¹H/¹³C NMR). Antioxidant activity was evaluated by DPPH assay, while cytotoxicity was evaluated by MTT cell viability assay. Antimicrobial activity was evaluated by agar diffusion method. The docking study was conducted using Auto Dock Vina; the estrogen receptor (PDB 5t92) was used as a receptor for the docking.

Results: Chromatographic separation techniques were led to the isolation of five phenolic compounds; these compounds were identified to be apigenin 8-C-β-D-glucopyranoside (Vitexin) (1), apigenin 6-C-β-D-glucopyranoside (Isovitexin) (2), luteolin 7-O-β-D-glucopyranoside (3), luteolin 8-C-β-D-glucopyranoside (Orientin) (4), luteolin 6-C-β-D-glucopyranoside (Isoorientin) (5). They were isolated and identified for the first time from this plant species. The AME and EAF showed moderate activity against Gram positive and Gram negative bacteria, while both of them showed similar and powerful antioxidant activity with SC₅₀ = 12.8±0.56 µg/ml and SC₅₀ = 17±0.77 µg/ml respectively, compared to ascorbic (reference drug) SC₅₀ = 14.2±0.35 µg/ml. The EAF showed higher cytotoxic activity on the MCF-7 cells (human breast cancer cell line), with IC₅₀ = 12.3 ± 1.82 µg/ml, compared to Vinblastine Sulfate (reference drug). All isolated compounds showed good binding affinity to the estrogen receptors existed in the MCF-7 cell.

Conclusion: Five phenolic compounds were isolated for the first time from the EAF of *Dypsis leptocheilos* leaves. The AME and EAF extracts showed variable antioxidant, antimicrobial and cytotoxic activities.

Keywords: Antimicrobial, Antioxidant, Cytotoxicity, Docking, *Dypsis leptocheilos*, Polyphenolic

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INTRODUCTION

Medicinal plants are the richest bio-resource of drugs related to the traditional medical system, modern medicines, nutraceuticals, food supplements, folk medicines, pharmaceutical intermediates, and chemical entities for synthetic drugs [1]. The medicinal plants are beneficial for curing human diseases because of the presence of phytochemical constituents. Several important medicinal components are derived from plants like alkaloids, flavonoids tannins, terpenoids, steroids etc [2].

Dypsis is one of the largest single genera in the palm family, with over 170 species ranging from tiny understory plants to massive emergent canopy palms and just about every size in between. *Dypsis* are an extraordinary phenomenon of evolutionary diversity; this genus is composed of numerous former separate palm genera, including *Vonitra*, *Chrysalidocarpus*, *Phloga*, *Neophloga*, *Phogella*, *Trichodypsis*, *Haplodypsis*, *Adelodypsis*, *Antogilia*, and *Neodypsis*. *Dypsis* were catalogued and described in great detail in "The Palms of Madagascar" a seminal work by Dr. John Dransfield and Henk Beentje, first published in 1995 by Kew Botanical Gardens [3]. By reviewing the current literature, it was found that no reports dealt with the phytochemical study of *D. leptocheilos* and this encouraged our team to work for isolation and identification of its Phyto-constituents in addition to evaluation of its therapeutic activities.

On the basis of the fact, that said the development of antibiotic resistance in microorganisms is a global challenge for the clinicians,

pharmacist, and research scientists leading to the development of new medicinal formulations that are effective and easily consumable [1] we decided to study antimicrobial activities of *D. leptocheilos*. As Anti-oxidants play an important role in protecting and safeguarding health problem especially in the disease such as cancer [4] that encouraged our team to measure antioxidant activity of *D. leptocheilos*. Molecular docking of isolated compounds was carried out as it is a method which predicts the preferred orientation of one molecule to a second when bound to each other to form a stable complex [5]. Molecular docking research focuses on computationally simulating the molecular recognition process. It aims to achieve an optimized confirmation for both the protein and ligand and relative orientation between protein and ligand such that the free energy of the overall system is minimized [6].

In this study, the isolation of five phenolic compounds from *D. leptocheilos* leaves extract was reported and their molecular docking study was performed, in addition to the evaluation of the antioxidant, cytotoxic and antimicrobial activities of the AME and EAF extracts.

MATERIALS AND METHODS

Plant material

The leaves of *Dypsis leptocheilos* (Arecaceae) were collected from Al-Abed garden, Alexandria-Egypt, desert road in August 2017. The plant material was identified by Dr. Terasse Labib, Department of Flora and Taxonomy, Al-Orman Botanical Garden, Giza, Egypt. A

voucher specimen No.01Dle/2017 was kept in the herbarium in the Department of Pharmacognosy, Faculty of Pharmacy, Helwan University, Egypt.

Materials for biological studies

Mammalian cell lines: MCF-7 cells (human breast cancer cell line), HepG-2 (human hepatocellular carcinoma) and HeLa cells (human cervical cancer cell line), were obtained from the American Type Culture Collection (ATCC, Rockville, MD). Fungi: *Aspergillus fumigatus* (RCMB 002008) and *Candida albicans* (RCMB 005003 (1) ATCC 10231), Gram-positive bacteria: *Staphylococcus aureus* (RCMB010010), *Bacillus subtilis* (RCMB 015 (1) NRRL B-543), Gram-negative bacteria: *Salmonella typhimurium* (RCMB 006 (1) ATCC 14028), *Escherichia coli* (RCMB 010052) were supplied from the Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University, Cairo, Egypt.

Instruments and chemicals for biological studies

All cell lines were cultured in DMEM media supplemented with 10% (v/v) foetal bovine serum, 2 mmol glutamine and 50 µg/ml penicillin/streptomycin solution (all from Invitrogen, Paisley, UK). Dimethyl sulfoxide (DMSO), MTT and trypan blue dyes were purchased from (Sigma-Aldrich) (St. Louis, Mo., USA). Chemicals for evaluating the antimicrobial activity; ascorbic acid, Ketoconazole, Gentamycin and Vinblastine Sulfate were supplied by RCMB: Regional Center for Mycology and Biotechnology, Cairo, Egypt. Chemicals used for evaluating the antioxidant activity, DPPH (2, 2-Diphenyl-1-picrylhydrazyl) and Ascorbic acid were purchased from Sigma-Aldrich Co., UK. Authentic reference flavonoid compounds were supplied by Pharmacognosy department, Faculty of Pharmacy, Helwan University. Authentic sugars were purchased from Sigma-Aldrich Co., UK. A microplate reader (SunRise, TECAN, Inc, USA), the 96-well plate used for cytotoxicity evaluation using cell viability assay. UV-visible spectrophotometer (Milton Roy, Spectronic 1201), used for measuring the absorbance in the antioxidant assay.

Instruments and materials for chromatographic techniques

The NMR spectra were recorded using Bruker a 400 MHz for ¹H NMR and 100.40 MHz for ¹³C NMR. The spectra were run in DMSO, and chemical shifts were given as δ ppm relative to tetramethylsilane (TMS) as an internal standard. Negative ESI-MS were run on LCQ deca MS and LTQ-FT-MS spectrometers for MS analysis (Thermo Electron, Finnigan, Germany). For column chromatography, silica gel G60 for column chromatography (70-230 mesh, Merk), silica gel G60 for thin layer chromatography (E. Merk, Germany), silica gel GF254 pre-coated TLC plates (E. Merk, Germany), sheets of Whatman filter paper (1 mm) for paper chromatography (WhatmanLtd, Maid stone, Kent, England), micro crystalline cellulose (E. Merk-Darmstadt, Germany) and polyamide 6S (Riedel-De-Haen AG, SeelzeHaen AG, SeelzeHanver, Germany). Solvent systems for paper chromatography: S₁ (n-BuOH-HOAc-H₂O 4: 1: 5, top layer), S₂ (15% aqueous HOAc) were used [7].

Extraction and isolation

Air-dried leaves of *D. leptocheilos* (1 kg) were coarsely ground and extracted exhaustively with 80% methanol/H₂O (5 L x 4, 60 °C, 4 h). Then, the total extract was evaporated to dryness under reduced pressure to yield 115 g of the dark brown residue. This residue was reconstituted with 300 ml H₂O then fractionated with 3 x 300 ml of petroleum ether and ethyl acetate by liquid-liquid phase separation yielding four fractions weighing (15 g petroleum ether fraction, 17 g ethyl acetate fraction and 65 g aqueous fraction). 2D-PC of the fractions revealed the presence of a pronounced number of flavonoid spots in the ethyl acetate fraction, which were detected under UV-light and spray reagents. The EAF (10 g) was fractionated on silica gel column (5 x 90 cm), the column was eluted using 25:75 petroleum ether/CH₂Cl₂ mixtures with increasing polarity till 100% CH₂Cl₂ then increasing polarity using methanol up to 25:75 (MeOH/CH₂Cl₂), yielding 46 individual fractions, collected into 6 fractions (I-VI). Fraction I was found to be polyphenolic free. Fraction II (450 mg) was rechromatographed on polyamide column (2.5 x 50 cm), the column was eluted using water then H₂O/MeOH mixtures (0-40% MeOH) to yield 6 subfractions (i-vi), the promising

subfraction was subfraction iii, (89 mg), showed one dark purple spot which was further purified on cellulose column using H₂O, to afford pure compound 1 (14 mg). Fraction III (650 mg) was fractionated on polyamide column (3 x 65 cm), the column was eluted using water then H₂O/MeOH mixtures with decreasing polarity to yield 5 subfractions (i-v), the promising fractions were subfractions iii and v, which were further individually purified on cellulose column using H₂O, to afford pure compound 2 (15 mg), compound 3 (20 mg). Fraction IV (320 mg) was fractionated on a cellulose column using H₂O-MeOH (10-90%) mixtures, giving two subfractions, i and ii, which were further individually purified on sephadex LH-20 columns, using EtOH, to afford pure samples of 4 (12 mg) and 5 (10 mg), respectively.

Antioxidant activity (DPPH radical scavenging activity)

Antioxidant activity of the AME and EAF was determined at the Regional Center for Mycology and Biotechnology (RCMB) at Al-Azhar University by the DPPH free radical scavenging assay in triplicate and mean values were considered [8]. Freshly prepared (0.004% w/v) methanol solution of 2, 2-diphenyl-1-picrylhydrazyl (DPPH) radical was prepared and stored at 10°C in dark. A methanol solution of the tested extracts were prepared with sample concentrations (0, 10, 20, 40, 80, 160, 320, 640, 1280 µg/ml). A 40 µl aliquot of the methanol solution was added to 3 ml of DPPH solution. Absorbance measurements were recorded immediately with a UV-visible spectrophotometer. The decrease in absorbance at 515 nm was determined continuously, with data being recorded at 1 min intervals until the absorbance stabilized (16 min). The absorbance of the DPPH radical without antioxidant (control) and the reference compound ascorbic acid were also measured. All the determinations were performed in three replicates and averaged. The percentage of DPPH radical scavenging was calculated according to the formula: (scavenging activity) = [(AC-AT)/AC] × 100] (1). Where AC = Absorbance of the control at t = 0 min and AT = absorbance of the sample+DPPH at t = 16 min³.

Antimicrobial activity

The AME and EAF of *D. leptocheilos* leaves were assayed for antimicrobial activity using the susceptibility tests. Screening tests regarding the inhibition zone were carried out by the well diffusion method [9]. The inoculum suspension was prepared from colonies grown overnight on an agar plate and inoculated into Mueller-Hinton broth (fungi using malt broth). A sterile swab was immersed in the suspension and used to inoculate Mueller-Hinton agar plates (fungi using malt agar plates). The extracts were dissolved in dimethyl sulfoxide (DMSO) with different concentrations (10, 5, 2.5 mg/ml). The inhibition zone was measured around each well after 24h at 37 °C. Controls using DMSO were adequately done.

Evaluation of cytotoxicity

The cytotoxic activity of the AME and EAF of *D. leptocheilos* leaves were evaluated against (HepG2), (MCF-7) and HeLa cell lines using the MTT cell viability assay.

Cell line propagation

The cells were grown on RPMI-1640 medium supplemented with 10% inactivated fetal calf serum and 50 µg/ml gentamycin +ampicillin. The cells were maintained at 37 °C in a humidified atmosphere with 5% CO₂ and were subcultured two to three times a week.

Cytotoxicity evaluation using viability assay

For cytotoxicity assays, the cancer cell lines were suspended in media at concentration 5×10⁴cell/well in Corning® 96-well tissue culture plates, and then incubated for 24 h. The tested AME and EAF were then added into 96-well plates (three replicates) to achieve twelve concentrations for each extract. Six vehicle controls with media or 0.5 % DMSO were run for each 96-well plate as a control. After incubating for 24 h, the numbers of viable cells were determined by the MTT test. Briefly, the media was removed from the 96-well plates and replaced with 100 µl of fresh culture RPMI 1640 medium without phenol red then 10 µl of the 12 mmol MTT stock solution (5 mg of MTT in 1 ml of PBS) to each well, including the untreated controls. The 96-well plates

were then incubated at 37 °C and 5% CO₂ for 4 h. An 85 µl aliquot of the media was removed from the wells, and 50 µl of DMSO was added to each well and mixed thoroughly with the pipette and incubated at 37 °C for 10 min. Then, the optical density was measured at 590 nm with the microplate reader (SunRise, TECAN, Inc, USA) to determine the number of viable cells and the percentage of viability was calculated as [(ODt/ODc)]×100% where ODt is the mean optical density of wells treated with the tested sample and ODc is the mean optical density of untreated cells. The relation between surviving cells and drug plotted to get the survival curve of each cell line. The 50% inhibitory concentration (IC₅₀), required to cause toxic effects in 50% of intact cells, was estimated from graphic plots of the dose-response curve for each concentration. Using Graph pad Prism software (San Diego, CA, USA) [10, 11].

Statistical analysis of data

All the experimental results were expressed as mean±SD. Analysis of variance was performed by ANOVA procedures. Correlation coefficient (R²) was used to determine two variables. SPSS software was used for statistical calculations. The results with *p*<0.05 were regarded to be statistically significant.

Docking studies

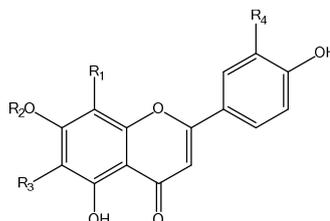
The docking study was conducted using Auto Dock Vina, M. G. L tools 1.5.7 and Discovery Studio 4.5 as a visualizer. The estrogen receptor (PDB 5t92) was used as a receptor for the docking of the five compounds. To ensure a valid docking and to evaluate the effect of the water molecules, the co-crystallized ligand in the estrogen

receptor was docked to its corresponding protein (in absence and in the presence of water) and the RMSD values between the co-crystallized ligand and the docked pose were calculated. The active site of the estrogen receptor has been determined from the binding of co-crystallized ligand. The energy minimized estrogen receptor, the co-crystallized ligand and the three isolated compounds were finally prepared in the right format using MGL tools 1.5.7 for conducting docking study by Auto Dock Vina that requires both the receptor and the ligands in pdbqt format [12]. Grid was generated for the receptor using MGL tools 1.5.7. Auto Dock Vina achieves approximately two orders of magnitude speed-up compared to the molecular docking software Auto Dock 4, while also significantly improving the accuracy of the binding mode predictions. Further speed-up is achieved from parallelism, using multithreading on multi-core machines. Auto Dock Vina uses Auto Dock score that calculates free binding energies and iterated local search global optimization algorithm [13-15]. The result of docking was visually inspected by discovery studio 4.5 visualizer. The evaluation of candidates was based on binding affinity and interaction with the receptor.

RESULTS AND DISCUSSION

Characterization and identification of isolated compounds

Chromatographic separation of EAF of *D. leptocheilos* leaves resulted in five compounds. Structures of the isolated compounds (fig. 1) were identified by different spectral techniques including ¹H NMR, ¹³C NMR, ESI-MS and CoPC against standard authentic after complete acid hydrolysis.



Compound	R ₁	R ₂	R ₃	R ₄
1	Glu	H	H	H
2	H	H	Glu	H
3	H	Glu	H	OH
4	H	H	Glu	OH
5	Glu	H	H	OH

Fig. 1: Chemical structures of the isolated compounds from *D. leptocheilos* leaves

Table 1: ¹H NMR (400 MHz, DMSO-*d*₆), ¹³C NMR (100.40 MHz, DMSO-*d*₆) of compound 1, 2, 4 and 5

C-No	Compound 1		Compound 2		C-No	Compound 4		Compound 5	
	δ C	Hδ	δ C	Hδ		δ C	Hδ	δ C	Hδ
2	164.31		163.23		2	164.17		164.57	
3	102.85	6.72, s	102.96	6.69, s	3	103.11	6.68, s	102.75	6.65, s
4	182.48		181.81		4	182.40		182.54	
5	161.72		161.76		5	163.73		160.99	
6	98.75	6.19, s	109.59		6	109.19		98.48	6.28, s
7	161.68		161.13		7	160.97		162.98	
8	106.08		94.64	6.39, s	8	93.91	6.49, s	104.91	
9	156.46		156.97		9	156.68		156.33	
10	104.32		102.96		10	104.09		104.31	
1'	122.02		121.57		1'	121.75		122.33	
2'/6'	128.89	8.00, d, (8.22)	128.73	7.90, d, (8.51)	2'	113.91	7.41, d, (2.4)	114.48	7.49, d, (2.4)
3'/5'	116.29	6.90, d, (8.61)	116.49	6.92, d, (8.58)	3'	146.14		146.51	
4'	160.85		159.20		4'	150.07		150.04	
1''	73.89	4.72, d, (9.93)	73.81	4.59, d, (9.88)	5'	116.54	6.90, d, (8.25)	116.06	6.86, d, (8.97)
2''	71.35	3.2-3.7, m,	71.01	3.2-3.7, m,	6'	119.23	7.44, dd, (2.4 and 8.25)	119.97	7.54, dd, (2.4 and 8.97)
3''	79.13	remaining of	79.54	remaining of	1''	73.50	4.59, d, (9.81)	73.48	4.69, d, (9.83)
4''	70.65	sugar protons	70.65	sugar protons	2''	71.14	4.05, t-like	71.17	4.99, s
5''	82.28		81.86		3''	79.18	3.15-3.89, m,	79.03	3.15-3.89, m,
6''	61.75		61.82		4''	70.41	remaining of sugar	62.10	remaining of sugar

5''	81.92	protons	82.37	protons
6''	61.94		56.87	

Values between parentheses represent the *J*-values in Hz

Compound 1 and 2 were isolated as yellow amorphous powder each, chromatographic properties; *R_f*-values (0.43 *S₁*, 0.49 *S₂*) and (0.57 *S₁*, 0.55 *S₂*) respectively, they gave dark purple spot under UV-light, turned to yellowish-green on exposure to NH₃ vapors, grayish-yellow fluorescence on exposure to Naturstoff and green color with FeCl₃ spray reagents. Based on its chromatographic properties, they were expected to be an apigenin structure [7]. This expectation was supported by ¹H NMR spectrum of 1 and 2 (table 1) which showed an *AzX₂* spin coupling system of two ortho-doublets, each integrated for two Protons at δ ppm 8.00 (2H, *d, J* = 8.22 Hz H-2'/6') and 6.90 (2H, *d, J* = 8.61 Hz H-3'/5') for 1, 7.90 (2H, *d, J* = 8.51 Hz H-2'/6') and 6.92 (2H, *d, J* = 8.58 Hz H-3'/5') for 2 indicated 4'-hydroxy B-ring. The glycoside moiety in both compounds was identified as β-C-glucoside from doublets at δ 4.72 and 4.59 with large *J* values (≥ 9Hz) for compounds 1 and 2, respectively. Absence of H-8 and H-6 signals from ¹H NMR spectrum of 1 and 2 led us to conclude that the C-glycosylation must be on C-8 in case of 1 and C-6 in case of 2. This evidence was confirmed from the downfield shift of C-8 to 106.08 and C-6 to 109.59 (~+10 ppm) in ¹³C NMR spectra (table 1) of compound 1 and 2 respectively; moreover, the C-glycoside moiety in the structures was confirmed as β-C-glucopyranoside depending on the characteristic up field location of C-1'' at δ 73.89 ppm for compound 1 and at δ 73.81 ppm for compound 2 and downfield location of both C-5'' and C-3'' at δ 82.28 and 79.13 ppm respectively, for compound 1 and at 81.86 and 79.54 ppm for compound 2 with respect to those of *O*-glycosides. The assignment of all other ¹³C resonances of compounds 1 and 2 was achieved by comparison with the corresponding data of structural related compounds [16-21]. Negative ESI-MS spectra of compounds 1 and 2 exhibited the molecular ion peak at *m/z* 431 {M-H} corresponding to the Mwt of 432 and molecular formula of C₂₁H₂₀O₁₀ to support evidence of apigenin mono-hexoside structure of both compounds. According to the above-discussed data as well as comparison with authentic samples, compound 1 was confirmed as apigenin-8-C-β-D-glucopyranoside (Vitexin) and compound 2 as apigenin-6-C-β-D-glucopyranoside (Isovitexin), which was isolated for the first time from *D. leptocheilos*.

Compounds 4 and 5 were obtained as yellow amorphous powder, chromatographic properties, *R_f*-values for 4 (0.43 *S₁*, 0.39 *S₂*) and (0.29 *S₁*, 0.21 *S₂*) for 5, they gave dark purple spot under UV-light, turned to yellow on exposure to NH₃ vapors, orange fluorescence on exposure to Naturstoff and green color with FeCl₃ spray reagents, Compound 4 and 5 were expected to be a luteolin structure [3]. ¹H NMR spectra (table 1) showed an ABX-spin coupling system of three proton resonances at δ 7.44 (1H, *d, J* = 2.4 and 8.25 Hz, H-6'), 7.41 (1H, *d, J* = 2.4 Hz, H-2') and ortho-doublet at δ 6.90 (1H, *d, J* = 8.25 Hz, H-5') for 4 and δ 7.54 (1H, *d, J* = 2.4 and 8.97 Hz, H-6'), 7.49 (1H, *d, J* = 2.4 Hz, H-2') and ortho-doublet at δ 6.86 (1H, *d, J* = 8.97 Hz, H-5') for 5, indicating that both 4 and 5 are luteolin derivatives. In the aliphatic region, the doublets at δ 4.59 and 4.69 with large *J* values (>9 Hz) for the anomeric protons of 4 and 5 respectively, were intrinsic for a β-C-glycoside moiety in both compounds. Absence of H-6 and H-8 signals from ¹H NMR of 4 and 5, respectively, was confirmative for C-glycosylation at C-6 in case of 4 and C-8 in case of 5. As further confirmation, ¹³C NMR spectrum for each compound (table 1) showed

well-resolved typical 15 signals of a luteolin aglycone moiety, including the three key signals of C-3', C-4' and C-3 at δ ppm 146.14, 150.07 and 103.11 for compound 4 and at δ ppm 146.51, 150.04 and 102.75 for 5. Additionally, the C-glycoside moiety in both structures was confirmed as β-glucopyranoside depending on the characteristic up-field location of C-1'' at 73.50 and 73.48 ppm for compound 3 and compound 4, respectively, and downfield location of both C-5'' and C-3'' to δ 81.92 and 79.18 ppm for compound 3 and to δ 82.37 and 79.03 ppm for compound 4, with respect to those of *O*-glycosides [18]. The C-glycosidation at C-6 in 4 and at C-8 in 5 was concluded from the downfield shift of ¹³C-signals of C-6 to 109.19 and of C-8 to δ 104.91 (~+10 ppm) for 4 and 5 respectively. The assignment of all other ¹³C NMR resonances was achieved by comparison with the corresponding data of structural related compounds [22-24]. Negative ESI-MS spectra of 4 and 5 exhibited the molecular ion peak at *m/z* 447 {M-H} corresponding to the Mwt of 448 and molecular formula of C₂₁H₂₀O₁₁. Thus according to the above-discussed data, compound 4 was confirmed as luteolin-6-C-β-D-glucopyranoside (Isoorientin), while compound 5 was confirmed as luteolin-8-C-β-D-glucopyranoside (Orientin). However, both compounds were isolated for the first time from *D. leptocheilos*.

Compound 3, Isolated as yellow amorphous powder, ¹H NMR: δ ppm 7.46 (1H, *dd, J* = 7.94, 1.8 Hz, H-6'), 7.43 (1H, *d, J* = 1.8 Hz, H-2'), 6.92 (1H, *d, J* = 7.94 Hz, H-5'), 6.80 (1H, *d, J* = 1.8 Hz, H-8), 6.75 (1H, *s, H-3*), 6.45 (1H, *d, J* = 1.8 Hz, H-6), 5.09 (1H, *d, J* = 6.83 Hz, H-1''), 3.73-3.17 (5 H, *m, H-2'', 3'', 4'', 5'' and 6''*). ¹³C NMR (100.40 MHz, DMSO-d₆): δ 182.36 (C-4), 164.95 (C-2), 163.40 (C-7), 161.59 (C-5), 157.41 (C-9), 150.48 (C-4'), 146.27 (C-3'), 121.77 (C-1'), 119.64 (C-6'), 116.47 (C-5'), 113.28 (C-2'), 105.80 (C-10), 103.58 (C-3), 100.36 (C-1''), 100.00 (C-6), 95.21 (C-8), 77.61 (C-5''), 76.83 (C-3''), 73.57 (C-2''), 70.02 (C-4''), 61.08 (C-6''). Negative ESI-MS spectra of compound 3 exhibited the molecular ion peak at *m/z* 447 {M-H} corresponding to the Mwt of 448 and a molecular formula of C₂₁H₂₀O₁₁. The ¹H NMR spectrum showed two meta-coupled doublets (*J* = 1.9 Hz) at δ 6.80 and 6.45, each integrating for one proton, and were assigned to H-8 and H-6, respectively of ring A of 5, 7-dihydroxyflavonoids. The presence of ABX system at δ 7.46 (*dd, J* = 7.94, 1.8 Hz), 7.43 (*d, J* = 1.8 Hz) and 6.92 (*d, J* = 7.94 Hz), characteristic of 1, 2, 4-trisubstituted phenyl unit. The only singlet at δ 6.75, integrating for one proton, was attributed C-3 to proton of flavonoids. These spectral data revealed the presence of luteolin skeleton. In addition, the ¹H NMR spectrum showed a series of signals between δ 3.73-3.15, attributable to the sugar moiety. The coupling constant (*J* = 6.83 Hz) of the anomeric proton located at δ 5.09 and the ¹³C NMR chemical shifts of the sugar carbons (δ 100.36, 77.61, 76.83, 73.57, 70.02 and 61.08) revealed the presence of β-*O*-glucoside unit in luteolin-7-*O*-glucoside. The ¹³C NMR data showed the presence of a ketone carbonyl (δ 182.36), two olefinic carbons (δ 164.95 C-2 and 103.58 C-3), and four hydroxyl carbons (δ 163.40 C-7, 161.59 C-5, 150.48 C-4' and 146.27 C-3'). The assignment of all other ¹³C resonances of 3 was achieved by comparison with the corresponding data of structurally related compounds [25]. According to the above-discussed data as well as comparison with authentic samples, compound 3 was confirmed as luteolin-7-*O*-glucoside, which was isolated for the first time from *D. leptocheilos*.

Table 2: Antioxidant activity (scavenging activity) of the AME and EAF of *D. leptocheilos* leaves

Sample conc. (µg/ml)	Mean of DPPH scavenging %					
	AME		EAE		Ascorbic	
	Mean	SD (±)	Mean	SD (±)	Mean	SD (±)
1280	90.11	1.83	89.04	1.26	93.45	1.08
640	86.64	0.68	87.03	0.25	88.64	0.99
320	84.28	0.46	85.48	0.51	80.92	1.64
160	81.86	0.59	83.28	1.39	75.68	1.59
80	80.46	0.74	81.66	1.04	66.13	4.45
40	76.40	1.02	78.67	3.03	53.21	1.81
20	74.38	4.31	58.87	4.02	20.31	3.11
10	40.27	1.57	29.28	1.42	11.80	2.09

Results are means ± SD (n=3) *P*<0.05

Antioxidant activity

Free radicals are known as Reactive Oxygen Species (ROS) are produced by the human body. Plants are potential sources of natural antioxidants that protect the cells against the damaging effects of reactive oxygen species (ROS) [26]. The scavenging ability of DPPH radical is widely used for antioxidant evaluation of natural products besides other several *in vitro* complementary assays based on inactivation of O₂ and NO radicals [27]. DPPH assay was selected to evaluate the antioxidant power of the extracts. The AME and EAF of *D. leptocheilos* showed similar and powerful antioxidant activity with SC₅₀ = 12.8±0.56 µg/ml and SC₅₀ = 17±0.77 µg/ml respectively when compared to ascorbic acid SC₅₀ = 14.2±0.35 µg/ml and (table 2).

Antimicrobial activity

Results which compiled in table 3, showed that The AME and EAF of *D. leptocheilos* leaves displayed moderate activity against Gram-positive and Gram-negative bacteria; however, both of them possess no anti-fungal activity.

Evaluation of cytotoxicity

Results of the cytotoxic activity of the AME and EAF of *D. leptocheilos* leaves against MCF-7, HepG2 and HeLa cell lines are represented in tables (4, 5, 6 and 7).

Table 3: Antimicrobial activity of the AME and EAF of *D. leptocheilos* leaves

Sample tested microorganisms	AME		EAF		Control	
	Mean IZ	SD (±)	Mean IZ	SD (±)	Mean IZ	SD (±)
Fungi					Ketoconazole	
<i>Aspergillus fumigatus</i>	NA	NA	NA	NA	21.01	0.02
<i>Candida albicans</i>	NA	NA	NA	NA	23.03	0.04
Gram Positive Bacteria					Gentamycin	
<i>Staphylococcus aureus</i>	10.00	1.00	12.00	1.00	30.01	0.01
<i>Bacillus subtilis</i>	8.50	0.50	7.50	0.50	26.02	0.03
Gram Negative Bacteria					Gentamycin	
<i>Salmonella typhimurium</i>	NA	NA	10.50	0.50	33.02	0.03
<i>Escherichia coli</i>	9.50	0.50	11.50	0.50	27.09	0.01

*NA: No activity. The sample was tested at concentration 10 mg/ml. (n=3)

Table 4: Cytotoxic activity (viability %) of the AME and EAF of *D. leptocheilos* leaves against MCF-7 cell line compared to vinblastine sulfate

Sample conc. (µg/ml)	Mean of viability % MCF-7 cell line					
	AME		EAF		Vinblastine	
	Mean	SD (±)	Mean	SD (±)	Mean	SD (±)
500	4.57	0.24	3.40	0.22	4.12	0.17
250	6.36	0.38	6.93	0.49	7.24	0.38
125	12.54	0.79	12.67	0.72	13.78	0.87
62.5	23.44	1.39	22.30	1.92	19.00	2.22
31.25	33.15	2.64	34.11	2.87	26.40	3.01
15.6	44.79	2.78	44.49	2.73	37.05	1.47
7.8	59.50	3.02	57.91	4.70	43.96	2.28
3.9	74.41	3.52	73.93	4.54	53.96	2.06
2.00	100	----	100	----	61.38	2.52
1.00	---	---	---	---	69.21	3.73

Results are means±SD (n=3) P<0.05

Table 5: Cytotoxic activity (viability %) of the AME and EAF of *D. leptocheilos* leaves against HepG-2 cell line compared to vinblastine sulfate

Sample conc. (µg/ml)	Mean of viability % HepG-2 cell line					
	AME		EAF		Vinblastine	
	Mean	SD (±)	Mean	SD (±)	Mean	SD (±)
500	4.43	0.19	4.02	0.22	3.91	0.16
250	7.67	0.34	7.15	0.27	6.77	0.35
125	11.82	0.32	13.64	0.99	10.38	0.44
62.5	22.38	1.95	24.52	0.93	14.89	1.68
31.25	33.22	1.61	38.20	1.30	24.33	2.33
15.6	41.87	0.92	47.09	1.55	33.00	1.09
7.8	54.86	1.62	63.97	1.89	41.94	1.24
3.9	69.29	1.33	79.62	3.08	47.62	1.83
2.00	100	----	100	----	56.84	2.48
1.00	---	---	---	---	65.76	1.19

Results are means±SD (n=3) P<0.05

Table 6: Cytotoxic activity (viability %) of the AME and EAF of *D. leptocheilos* leaves against hela cell line compared to vinblastine sulfate

Sample conc. (µg/ml)	Mean of viability % HeLa cell line					
	AME		EAF		Vinblastine	
	Mean	SD (±)	Mean	SD (±)	Mean	SD (±)
500	6.35	0.46	5.48	0.53	5.12	0.89
250	11.62	0.83	10.62	0.75	9.08	1.33
125	23.11	1.96	19.26	1.69	15.31	1.29
62.5	34.29	2.31	31.53	3.06	23.79	2.79
31.25	45.13	3.38	39.24	2.35	29.96	3.16

15.6	60.79	3.58	52.46	3.62	38.47	2.57
7.8	81.28	3.41	70.00	3.75	46.42	0.70
3.9	89.82	4.86	85.62	3.41	57.55	3.05
2.00	100	-----	100	-----	66.50	2.35
1.00	---	----	----	----	73.66	2.37

Results are means \pm SD (n=3) $P < 0.05$

Table 7: IC₅₀ values of the standard drug, the AME and EAF of *D. leptocheilos* leaves

Sample	MCF-7	HeLa	HepG2
AME	IC ₅₀ = 12.8 \pm 1.04 μ g/ml	IC ₅₀ = 26.3 \pm 2.83 μ g/ml	IC ₅₀ = 10.7 \pm 0.37 μ g/ml
EAF	IC ₅₀ = 12.3 \pm 1.82 μ g/ml	IC ₅₀ = 18.5 \pm 3.29 μ g/ml	IC ₅₀ = 14.2 \pm 0.60 μ g/ml
Vinblastine Sulfate	IC ₅₀ = 5.44 \pm 0.57 μ g/ml	IC ₅₀ = 6.54 \pm 0.39 μ g/ml	IC ₅₀ = 3.48 \pm 0.22 μ g/ml

Results are means \pm SD (n=3) $P < 0.05$

The cytotoxic activity of the plant extracts was evaluated using MTT assay. The EAF showed higher cytotoxic activity against the three cell lines, with IC₅₀ = 12.3 \pm 1.82 μ g/ml, IC₅₀ = 14.2 \pm 0.60 μ g/ml and IC₅₀ = 18.5 \pm 3.29 μ g/ml respectively, while the AME showed higher cytotoxic activity on the MCF-7 and HepG2 cells, whereas it showed moderate activity against HeLa cell with IC₅₀ = 12.8 \pm 1.04 μ g/ml, IC₅₀ = 10.7 \pm 0.37 μ g/ml and IC₅₀ = 26.3 \pm 2.83 μ g/ml respectively, as compared to Vinblastine Sulfate with IC₅₀ = 5.44 \pm 0.57 μ g/ml, IC₅₀ = 3.48 \pm 0.22 μ g/ml and IC₅₀ = 6.54 \pm 0.39 μ g/ml respectively.

Docking study

Docking is the most widely used drug design tool in which the exact binding mode between the receptor and the ligand could be predicted. But docking procedure should always be validated and compared to a

reference crystal structure. Thus the co-crystallized ligand in Estrogen receptor (PDB ID: 5t92) was docked to its corresponding active site. The calculated RMSD between the co-crystallized ligand and the docked pose was 1.55 Å in the NW-docking (docking without water) and less than 0.2 Å in the W-docking (docking with water), indicating a valid docking protocol. Because of the different results of the NW-docking and the W-docking, the water molecules were kept in the docking procedure [28]. Finally, the pure isolated compounds were docked in the estrogen receptor. The docking result was analyzed using Discovery Studio 4.5.

The binding of the five compounds with the receptor was strong and involved many types of interactions. Table 8 summarizes the binding affinity scores of the three compounds and their most significant interactions with the receptor. The good binding mode of the three compounds is shown in (fig. 2).

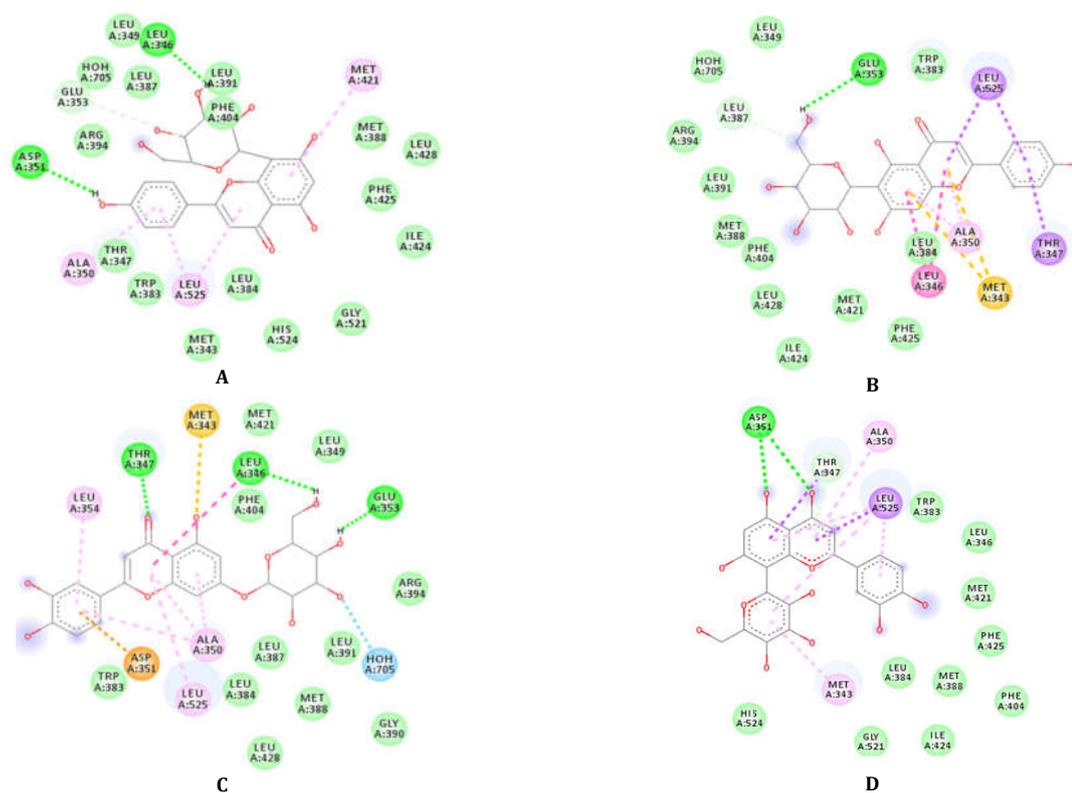




Fig. 2: The 2D interaction diagram of the three compounds with the estrogen receptor. (A) Vitexin, (B) Isovitexin, (C) Luteolin-7-O-glucoside (D) Orientin and (E) isoorientin

Table 8: The binding affinity scores of isolated compounds and their most significant interactions with the receptor

Compound name	Vina score (Kcal/MOI)	Residues involved in hydrogen bonds interactions	Residues involved in hydrophobic interactions
Vitexin	-6.8	GLU353, LEU346 and ASP351	MET421, ALA350 and LEU525
Isovitexin	-5.8	GLU353	MET343, LEU346, THR347, ALA350 and LEU525
Luteolin-7-O-glucoside	-7.5	GLU353, LEU346, THR347 and H ₂ O705	MET343, LEU346, ALA350, ASP351, LEU354 and LEU525
Orientin	-4.6	ASP351 and THR347	MET343, THE347, ALA350, HIS524 and LEU525
Isoorientin	-4.6	THR347, TRP383, HIS524 and GLU339	MET343, TRP 383 and LEU525

Results of the molecular docking of vitexin, isovitexin, luteolin-7-O-glucoside, orientin and Isoorientin in estrogen receptor revealed a favorable bind mode with high docking score, which may confirm the higher cytotoxic activity of the EAF.

CONCLUSION

Chromatographic separation of the EAF of *D. leptocheilos* leaves led to isolation and identification of five phenolic compounds for the first time from this plant species, namely apigenin 8-C- β -D-glucopyranoside (Vitexin) (1), apigenin 6-C- β -D-glucopyranoside (Isovitexin) (2), luteolin 7-O- β -D-glucopyranoside (3), luteolin 8-C- β -D-glucopyranoside (Orientin) (4), luteolin 6-C- β -D-glucopyranoside (Isoorientin) (5). The AME and EAF were evaluated for their antioxidant activity using DPPH assay; both showed similar and powerful antioxidant activity which attributed to the polyphenolic content. The antimicrobial activity of two extracts was found to be moderate against bacteria and very weak against fungi. Cytotoxic activity also has been evaluated for both extracts.

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Nil

AUTHORS CONTRIBUTIONS

Eman G. Haggag revised the manuscript; Haitham A. Ibrahim, Fatehia S. Elsharawy and Samah S. Shabana conducted the chromatographic separation, performed the structure elucidation of the pure isolated compounds and were responsible for drafting and writing the final version of the manuscript. Mahmmoud A. Elhassab performed the molecular docking study. All authors performed the antimicrobial, antioxidant and cytotoxicity assays and their data analysis in addition, they read and approved the final manuscript.

CONFLICTS OF INTERESTS

Authors declared no conflict of interest.

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