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Original Article

MICROBIAL SYNTHESIS OF SILVER NANOPARTICLES USING *STREPTOMYCES* SP. PG12 AND THEIR CHARACTERIZATION, ANTIMICROBIAL ACTIVITY AND CYTOTOXICITY ASSESSMENT AGAINST HUMAN LUNG (A549) AND BREAST (MCF-7) CANCER CELL LINES

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ABSTRACT

Objective: Synthesis of silver nanoparticles using *Streptomyces* sp. PG12 and their characterization, antimicrobial activity and cytotoxicity against A549 and MCF-7 cancer cell lines.

Methods: The silver nanoparticles were subjected to UV-Vis. spectroscopy, Fourier transform infrared spectroscopy (FTIR), scanning electron microscopy with energy-dispersive X-ray spectroscopy (SEM/EDS), high-resolution transmission electron microscopy (HR-TEM), zeta potential, and X-ray diffractometry (XRD) analyses. Further, the antimicrobial potential was determined by using the agar well diffusion method and cytotoxicity was determined with the help of cell viability (MTT) assay and reactive oxygen species (ROS) assay.

Results: The initial indication of silver nanoparticles synthesis was noticed by the colour change in the reaction mixture and the absorption maximum at 421 nm in UV-Vis. analysis; whereas, the FTIR analysis displayed the biological functional groups responsible for the capping and stabilization of silver nanoparticles. SEM and TEM micrographs revealed the surface morphology, spherical shape, and smallest particle size as 18.91 nm. The EDS and XRD patterns confirmed the involvement of various elements during the synthesis of silver nanoparticles and the crystalline, face-centered cubic nature, respectively. The silver nanoparticles displayed considerable antimicrobial activity against human pathogens even at low MIC and MBC concentrations and exhibited increased anticancer activity against A549 and MCF-7 cell lines, where the ability of silver nanoparticles to significantly restrict the growth of tumour cells was observed at IC₅₀ values of 69.04µg/ml and 138.30µg/ml, respectively.

Conclusion: Streptomyces sp. PG12 synthesized silver nanoparticles show significant anticancer activity against A549 and MCF-7 cell lines.

Keywords: Silver nanoparticles, Streptomyces sp., Antimicrobial activity, ROS expression, Human lung and breast cancer cell lines

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INTRODUCTION

Nanotechnology is the concept that deals with the production, manipulation, characterization, and applications of nano-scale materials whose size usually ranges between 1 to 100 nm. Nano-materials prepared from metals offer unique properties of size, shape, and larger surface area to volume ratio [1]. In the recent decade, these metal nanoparticles prepared using gold, silver, platinum, copper, zinc, lead, iron, and many more are extensively used for various applications in divergent fields such as nano-medicine, electrical, agriculture, pharmaceuticals, cosmetics, food industry, textiles, and data storage [2, 3].

Silver is more often considered for synthesizing nanoparticles due to potential antimicrobial, antioxidant, anti-inflammatory, and anticancer activities. The recent advancements in nano-science have pointed out the significance of silver nanoparticles (AgNPs) in the areas of biomedical, DNA sequencing, biological sensors, plasmonics, catalysis, energy generation, and clean water technology [4, 5]. Generally, AgNPs are produced by one of two paths, i.e., 'top-down' or 'bottom-up' strategies via physical or chemical or biological synthesis routes. The physical and chemical techniques require more energy, thermal ablation, and hazardous chemicals that are carcinogenic and induce genotoxicity. Hence, due to its low cost, non-toxicity, and eco-friendly nature, the biological synthesis method is essential [6, 7].

The greener approach of AgNPs synthesis is carried out by employing biological resources like algae, lichens, bacteria, yeast, fungi, actinobacteria, and plant extracts. Among the microorganisms, bacteria are considered the ideal source for metal nanoparticles synthesis, owing to their metal resistance mechanism, which helps them reduce ionic silver (Ag⁺) into elemental (Ag^o) NPs [8, 9]. The microbial synthesis method of AgNPs allows for safer, cheaper, and more stable metallic AgNPs as the microbes produce various secondary metabolites, which actively participate in the reduction, capping, and stabilization [10].

The increasing multi-drug resistance in human pathogenic microorganisms is of great concern, and there is an urgent need for the development of broad-spectrum antimicrobials to treat severe infections. The AgNPs have gained importance as the alternative for antibacterial and antifungal agents and help reduce toxicity, avoid the side effects in patients, and prevent pathogens [11, 12]. Cancer is ranked top on the list of deadly diseases as the second leading cause for the most deaths across the globe, and the numbers are expected to increase further [13]. From the past few decades, AgNPs are extensively used as cytotoxic agents against various human cancers. The microbial AgNPs synthesized using species of *Pseudomonas* [14], *Enterococcus* [15], and *Streptomyces* [16, 17] have successfully prevented the growth of tumour cells.

Actinomycetes belong to Gram-positive, aerobic and filamentous bacteria with high G+C content. Many of the Actinomycetales members are known for their peculiar capacity to produce bioactive secondary metabolites having properties from antimicrobial to anticancer and many more [18-20]. Because of these essential biological properties of actinobacterial secondary metabolites, the culture filtrate of *Streptomyces* sp. PG12 was to investigate for the extracellular synthesis of AgNPs and their evaluation for antimicrobial and anticancer potential.

MATERIALS AND METHODS

Collection of pathogens

The pathogenic microorganisms such as *Staphylococcus aureus* (MTCC6908), *Pseudomonas aeruginosa* (MTCC424), *Enterococcus*

faecalis (MTCC6845), Escherichia coli (MTCC40), Alternaria alternata (MTCC2060), Fusarium oxysporum (MTCC2087), Candida albicans (MTCC227), and Candida glabrata (MTCC3814) were procured from Institute of Microbial Technology, Chandigarh, India.

Isolation of actinobacterial strains

During the present study, the marine samples were collected from Maravanthe beach, Kundapura, Karnataka, India. The serially diluted samples were inoculated onto starch casein agar (SCA) media plates. The plates were subjected to incubation at 30 °C for 5 d, and 45 Actinomycete strains were isolated. From these 45 isolates, a single Actinomycete was selected for the green AgNPs synthesis and designated as strain PG12.

Genotypic characterization of the isolate

The fresh culture of strain PG12 was employed to extract genomic DNA using the instructions of Hi-PurA DNA purification kit manufacturers. The 16S rRNA gene was sequenced (Sanger Sequencing 3500 Series, Genetic Analyzer) and deposited to the NCBI database via nucleotide BLAST web portal. Similar DNA sequences were selected to construct a phylogenetic tree using MegAlign Pro 17 (DNASTAR Lasergene) software as per the standard procedures [21].

Preparation of cell-free extract

The strain PG12 was inoculated into a 250 ml Erlenmeyer flask with sterile starch casein broth. A rotary shaker at a speed of 200 rpm was used to incubate the culture flasks for 5 d at room temperature. After incubation, the centrifugation at 10000 rpm (REMI R-24) for 15 min was done to obtain the cell-free extract for AgNPs synthesis [21].

Biosynthesis of AgNPs

For the AgNPs synthesis, about 100 ml aqueous solution of 0.1 mmol AgNO₃ was mixed with 100 ml cell-free supernatant of PG12 strain. After adjusting the pH at 8.0, incubation for 5 d under dark conditions was followed, and the reduction of silver ions (Ag⁺) was examined by observing the change in colour of the reaction mixture [21].

Characterizations of AgNPs

The reduction of Ag⁺ into Ag^o was recorded using a double beam UV-Visible spectrophotometer (UV-9600A, METASH Instruments, Shanghai, China). The probable biochemical functional groups responsible for stabilizing and capping AgNPs were determined by FTIR spectrophotometer (Nicolet 6700, Thermo Fisher Scientific, Waltham, Massachusetts, USA) scanning from 4000 to 400 cm⁻¹ in transmittance mode. The morphology, shape, size, and elemental composition of AgNPs were analyzed with SEM/EDS (Jeol, JSM-IT500 LA, Japan) and HR-TEM (FEI, TECNAI G2, F30) instruments. Further, the electrostatic stability and zeta potential was calculated (HORIBA nanoparticle analyzer SZ-100, Kyoto, Japan) along with the crystalline nature of synthesized AgNPs by X-ray diffractometer (Rigaku SmartLab SE, Austin, Texas, US).

Antimicrobial activity of AgNPs

The antagonistic activity of *Streptomyces* sp. PG12 cell-free extract mediated AgNPs were determined against pathogenic fungi and bacteria using the agar well diffusion method on potato dextrose agar and nutrient agar, respectively. The different volumes of AgNPs (25, 50, 75, and 100 μ l) from the stock suspension of 50 μ g/ml concentrations were poured in appropriate wells (6 mm) along with positive (streptomycin and fluconazole) and negative (distilled water) controls. The plates were incubated at 35 °C for 24 h for bacterial pathogens and at 35 °C for 72 h for fungal pathogens. The formation of inhibition zones were measured as diameter in mm with the help of the Hi-antibiotic zone scale-C [22].

Evaluation of minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) of AgNPs

The MIC of the synthesized AgNPs on bacterial pathogens was determined with broth dilution assay in a 96-well microtitre plate with resazurin as standard, whereas; the MBC test was carried out on Mueller-Hinton agar plates. About 24 h incubation at 37 $^{\circ}$ C was used for the test plates, and the concentration where no growth of

bacteria was observed on plates was recorded as the MIC and MBC values [22].

Anticancer activity assessment of AgNPs

The cytotoxic activity of strain PG12 mediated AgNPs was investigated by employing the 96-well plate MTT cell proliferation assay. The cancer cells viz. A549 and MCF-7 cell lines were procured from NCCS, Pune, India, and sub-cultured and maintained on Dulbecco modified eagle medium (D-MEM) before incubation in a 5% CO₂ incubator at 37 °C. The cancer cells were plated in a 96-well plate (Nunc, Thermo Scientific, Cat. No. 265300) at a cell density of 20,000 cells/well for MTT assay, and ROS assay 10,000 cells/200 µl was used. For MTT assay, 3.48 µg/ml camptothecin (MCF-7), 2.58 µg/ml cisplatin (A549) were taken as standard, and only media with cells were taken as control; whereas, appropriate concentrations of AgNPs suspensions in dimethyl sulfoxide (DMSO) (12.5, 25, 50, 100 and 200 µg/ml) were added into respective wells. Finally, tumour cell viability was examined recording the absorbance at 570 nm on ELISA reader by keeping 630 nm as reference wavelength [23].

The ROS assay was conducted by treating the cells with test compounds with IC_{50} concentrations except for cell control in 200 µl of culture medium and incubated the cells for 24 h. During ROS expression study, 2', 7'-dichlorodihydrofluorescein diacetate (H2DCFDA) was used for staining the cells, which oxidizes and converts into fluorescent 2', 7'-dichlorofluorescein (DCF) by the generation of intracellular ROS. The cells were observed under a confocal microscope with 488 and 535 nm excitation and emission wavelengths for ROS study. All the images were recorded at a different time of intervals using an inverted phase-contrast microscope (Biolink) [23].

Statistical analysis

All the analysis performed was in triplicate and the data represented as mean ± standard deviation. The statistical analysis was performed by using GraphPad Prism 4.0 software (San Diego, California, USA).

RESULTS AND DISCUSSION

Genotypic characterization of the isolate

The 16S rRNA gene sequence of 1011 base pairs obtained from the strain PG12 was submitted to the NCBI database (Accession No. MN235876) and initially characterized by nucleotide BLAST to get homologous sequences. A neighbor-joining phylogenetic tree was constructed using the homologous sequences of various other species in which the strain PG12 showed the highest similarity with *Streptomyces* sp. strain Al-Dhabi-87 (fig. 1). Similarly, a strain CGG11n of actinobacteria identified based on 16S rRNA gene sequencing and phylogenetic tree analysis was reported [24].

Visual and UV-Vis analysis

The initial indication of AgNPs synthesis was noticed by the change in colour of the reaction mixture; the supernatant was orange-brown before adding 1 mmol AgNO₃ solution, the colour turned dark brown after adding AgNO₃ and incubation in the dark (fig. 2A). The further confirmation on the synthesis of AgNPs was determined by sharp absorption maximum at 421 nm in UV-Vis. spectrophotometer (fig. 2B). Similar observations about the colour change in the reaction mixture after mixing with AgNO₃ were reported and mentioned that the colour change is due to reducing silver ions in the AgNO₃ solution. Due to surface plasmon resonance (SPR) excitation, a sharp peak was obtained, confirming the AgNPs formation from the supernatant of strain PG12 [25, 26].

FTIR analysis

The FTIR analysis was carried out in transmittance mode from 4000 to 400 cm⁻¹ to recognize possible biochemical functional groups responsible for the AgNPs capping and stabilization. The FTIR spectrum of strain PG12 mediated AgNPs shown various characteristic peaks, which could be related to specific functional groups (fg. 3). The broad intensity peak at 3270 cm⁻¹ corresponds to H-bonded free alcohols/phenols, whereas the small intensity peak at 2925 cm⁻¹ is attributed to C-H stretching alkanes. The sharp, medium intensity peaks at 1725 and 1634 cm⁻¹ were correlated to C=O bonded

aldehydes and amides, respectively. The small intensity peaks at 1377 and 1229 cm⁻¹ corresponded to N=O bonded nitro compound (R-NO₂) and S=O bonded sulfates and sulfonamides. The strong intensity peak at 1026 cm⁻¹ relates to C-N bonded amines, and the weak peak at 446 cm⁻¹ is attributed to C-X bonded halides. The functional groups that

0.01

helped in the synthesis of AgNPs were identified by observing a shift in absorption spectra. These results suggest the critical role of biological molecules in the AgNPs synthesis along with stabilization by having the ability to link to metal. The presence of active functional groups indicates effective reduction and capping of AgNPs [27, 28].

AY999733.1_ Streptomyces violaceoruber strain NRRL B-3631	
HQ385919.1_Streptomyces sp. strain ERI-CPDA-1	
MG770770.1_ Streptomyces sp. strain 7K111	
KF841613.1_ Streptomyces sp. NEAU-D12	
MH185951.1_ Streptomyces caelestis strain ghp3	
MH236119.1_ Streptomyces caelestis strain ghp9	
LC425068.1_ Streptomyces enissocaesilis AL3	
r NR 0412211 Strentomyrae afrikanjancje strajn NBPC 12921	
MH201350.1. Strentomyces en strein Al-Dhabi-87	
White (Soch_ Supporting to Sp. Scale Ar Order of	MAN 225076 1 Strantomucae on strain DG1
	Invites boron _ streptomyces sp. strain Pon
MW131348.1_ Streptomyces rochei strain CMB47	
NR_112281.1_ Streptomyces mutabilis strain NBRC 12800	
C490345.1 Streptomycer op. etrain BBST1.4	





Fig. 2: Visual and UV-Vis. analysis of *Streptomyces* sp. PG12 AgNPs; A) colour change in the reaction mixture before and after adding AgNO₃, B) A sharp peak obtained in UV-Vis. absorption spectrum



Fig. 3: FTIR spectrum of Streptomyces sp. PG12 AgNPs

SEM with EDS analysis

The morphological details of synthesized AgNPs were ascertained by SEM imaging, where the shape and distribution of AgNPs were found to be spherical and polydispersed, respectively. The EDS analysis revealed Ag at 3 keV and various elements such as C, O, Na, Cl, K, and Ca, which involved the capping of synthesized AgNPs (fig. 4A and

4B). The previous SEM analysis results of green AgNPs suggested the spherical shape, poly-dispersed AgNPs from the microbial origin, and most of the particles had a smooth surface with a compact arrangement. Similarly, the strong signal peak at 3 keV in the EDS spectrum was due to the SPR of metallic silver. The analysis confirmed AgNPs synthesis with suggesting silver as the significant ingredient element [29, 30].



Fig. 4: SEM with EDS analysis of *Streptomyces* sp. PG12 AgNPs, A) SEM image showing morphology and distribution of AgNPs, B) EDS spectrum showing peaks for various elements

HR-TEM analysis

TEM analysis was performed to understand the morphological details of synthesized AgNPs, and the images confirmed the spherical shape of synthesized AgNPs (fig. 5A). The TEM data depicted the poly-dispersed nature of AgNPs with few agglomerates, and the AgNPs ranged from 20 to 32 nm in size with the smallest particle size of 18.91 nm (fig. 5B). These findings agreed with the previous studies on TEM analysis of AgNPs, where the biogenic AgNPs were spherical, poly-dispersed, and ranged from 20 to 50 nm in size. The TEM micrograph also revealed the uneven distribution of AgNPs with a negligible amount of agglomeration; the small sizes of particles have a crucial role in the penetration process inside microbial cells to induce a cascade of events that lead to cell death [31, 32].



Fig. 5: HR-TEM analysis of *Streptomyces* sp. PG12 AgNPs, A) TEM micrograph showing surface morphology and distribution of AgNPs, B) Image showing average size of AgNPs



Fig. 6: Graph showing zeta potential of Streptomyces sp. PG12 AgNPs

Zeta potential analysis

The stability of synthesized AgNPs was determined with zeta potential studies and the potential value of *Streptomyces* sp. PG12 AgNPs were determined to be -43.2 mV (fig. 6). Along with the stability, zeta analysis indicated the high dispersivity and well-defined colloidal nature of the AgNPs. This result was well supported by the previous studies on zeta potentials of AgNPs, hinting that the higher negative value attributes to the higher stability of AgNPs [33, 34].

XRD analysis

The XRD pattern obtained from PG12 AgNPs indicated four firm Bragg's peaks in the 2 θ range of 30[°] to 80[°], confirming the synthesis of AgNPs. The four Bragg's peaks were obtained at (111), (200), (220), and (311) for 2 θ angles of 38.13[°], 46.21[°], 64.48[°], and 77.42[°] determined the face-centered cubic nature of the synthesized AgNPs (fig. 7). Similar XRD patterns were observed in previous studies on AgNPs from *Streptomyces albogriseolus* [35] and *Streptomyces* sp. AS3 [36], suggesting the large crystalline domain sizes corresponding to pure silver metal. According to the previous reports, the extra peaks were due to the bio-organic phase present on the surface of AgNPs, and the more width of the peaks relates to

the smaller particle size. The obtained results of characteristic peaks were verified with the JCPDS database.

Antimicrobial activity of AgNPs

The dried AgNPs powder was washed and re-suspended in distilled water before testing antimicrobial activity on pathogenic microorganisms. Among the selected bacterial pathogens, E. faecalis was the most sensitive with inhibition zones of 19. 21. 21. and 23 mm for 25, 50, 75, and 100 μl of AgNPs suspensions, respectively. For fungal pathogens, AgNPs showed the highest activity against A. alternata with 18, 19, 20, and 21 mm inhibition zones for 25, 50, 75, and 100 µl, respectively; whereas S. aureus, P. aeruginosa, and F. oxysporum were found to be less sensitive for AgNPs (fig. 8). The previous antimicrobial activity results suggested that Streptomyces olivaceus (MSU3) mediated AgNPs effectively suppressed the growth of S. pneumoniae, S. mutant, K. pneumoniae, and E. coli at 30 µl [37]. Similarly, Streptomyces parvus (Al-Dhabi-91) showed increased antimicrobial activity even at low concentrations against pathogenic bacteria. Also, they suggested that the AgNPs may interact with microbial cell walls and lipopolysaccharides and disrupt them. Finally, the AgNPs leads to cell death when entering into the bacterial cell and binds to enzymes, proteins, and DNA by disrupting the cellular biochemical reactions and DNA replication mechanism [38].



Fig. 7: XRD pattern of Streptomyces sp. PG12 AgNPs



Fig. 8: Antimicrobial activity of Streptomyces sp. PG12 AgNPs against microbial pathogens, *values are the mean of three replicates±SD

Evaluation of MIC and MBC

The MIC can be defined as the minimum concentration of an antibiotic substance required to prevent bacterial growth, and the minimum concentration necessary to kill the pathogen is referred to as the MBC value. The biosynthesized AgNPs showed considerable MIC and MBC values against the test pathogens (table 1) and successfully inhibited the pathogens even at very low concentrations. Similar results were obtained in previous investigations, where the biological AgNPs at low concentrations showed high inhibitory activity against pathogens like *E. coli, E. faecalis, S. aureus,* and *S. pneumoniae.* They concluded that the AgNPs with smaller sizes provide a larger surface area for interaction with the microbial cell membrane, adding to the alterations in some primary functions of bacteria, such as permeability and cell respiration [39].

Table 1: MIC and MBC values of Streptomyces sp. PG12 AgNPs against pathogenic bacteria, *values are in mean three replicates±SD

S. No.	Name of the pathogen	MIC (µg/ml)	MBC (µg/ml)
1.	S. aureus	128	128
2.	E. faecalis	32	32
3.	S. pneumoniae	32	64
4.	E. coli	64	64

Anticancer activity of AgNPs

The cytotoxicity of AgNPs was evaluated on A549 and MCF-7 cell lines by employing both MTT assay and ROS expression study with 6.25 to 100 μ g/ml AgNPs concentrations and IC₅₀ concentration at different time intervals, respectively. The AgNPs showed dose-dependent activity on A549 (fig. 9 A-G) and MCF7 (fig. 10 A-G) cell lines in the MTT assay; the increase in ROS generation with an increase in treatment time of AgNPs significantly inhibits the growth by causing damage to the tumour cells of A549 cell line (fig. 11 A-G) and MCF-7 cell line (fig. 12 A-G) was observed. The anticancer activity was evaluated by the conversion of MTT to purple-colored formazan by mitochondrial succinate dehydrogenase, and the cytotoxicity was determined by the concentration of AgNPs required to minimize the survival of tumour cells by 50% (IC₅₀). The biosynthesized AgNPs exhibited considerable cytotoxic potential on A549 and MCF-7 cell lines with the IC₅₀ values of 69.04 µg/ml and 138.3 µg/ml, respectively, and was evidenced by the shrinkage of cancer cells.



Fig. 9: MTT assay of *Streptomyces* sp. PG12 AgNPs on A549 cell line; A) control, B) standard, C) 6.25µg/ml D) 12.5µg/ml, E) 25µg/ml, F) 50µg/ml, and G) 100µg/ml of AgNPs



Fig. 10: MTT assay of *Streptomyces* sp. PG12 AgNPs on MCF7 cell line; A) control, B) standard, C) 6.25µg/ml D) 12.5µg/ml, E) 25µg/ml, F) 50µg/ml, and G) 100µg/ml of AgNPs

Similarly, previous studies conducted on the anticancer activities of AgNPs against MCF7 [40], CT26 [41], Caco-2 [42], and HePG2 [43] cell lines reported that the AgNPs showed dose-dependent activity on selected cell lines. The AgNPs treated cells were different from the untreated cells in cell shape, size, and other morphological details. In the ROS assay study, the treated cells were somewhat granular shaped with less well-defined colony morphology. The appearance of dissembled gaps between neighboring cells were characteristics of apoptosis, and

the fluorescent green coloured cells represented necrotic cancer cells, which were dead due to the induced ROS. According to the previous reports, ROS are free radicals that play a vital role in the living system by causing cell damage and cell death. The rapidly increased ROS generation was due to the nature of surface, size, shape, and intensity of AgNPs. The overproduction of ROS induces apoptosis in several types of cancers, and cell death occurs due to an imbalance between pro and anti-oxidants levels inside the tumour cells [23, 44, 45].



Fig. 11: ROS-mediated anticancer activity of *Streptomyces* sp. PG12 AgNPs on A549 cell line at different intervals of time; A) 0 min, B) 5 min, C) 15 min, D) 30 min, E) 60 min, F) 90 min, and G) 120 min of treatment time



Fig. 12: ROS-mediated anticancer activity of *Streptomyces* sp. PG12 AgNPs on MCF7 cell line at different intervals of time; A) 0 min, B) 5 min, C) 15 min, D) 30 min, E) 60 min, F) 90 min, and G) 120 min of treatment time

CONCLUSION

The current study describes the synthesis of easy, non-toxic, and environmentally friendly extracellular AgNPs from Actinomycete isolate Streptomyces sp. PG12 culture supernatant. The primary indication regarding AgNPs synthesis was the colour change in the reaction mixture and confirmed by the UV-Vis. the absorption peak at 421 nm. The biochemical functional groups involved in the capping, stabilization, and reduction of AgNPs were determined by FTIR analysis, and SEM and TEM microscopy ascertained the surface morphology. The spherical AgNPs were present in the range of 20 to 32 nm with the smallest particle size of 18.91 nm. The biogenic AgNPs showed increased antimicrobial activity with lower MIC and MBC values. The biosynthesized AgNPs exhibited considerable anticancer activity, increasing ROS generation with increased exposure time on A549 and MCF-7 cell lines. The present results suggest that, Streptomyces sp. PG12 can be used for the mass synthesis of biological AgNPs, which could treat multidrug-resistant (MDR) microbial infections and as an anticancer agent after successful in vivo studies.

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Nil

AUTHORS CONTRIBUTIONS

SN designed the concept and supervised the experiments. PSS carried out the experimental analysis and wrote the manuscript. MPB wrote and edited the manuscript. All the authors read and approved the manuscript.

CONFLICT OF INTERESTS

The authors declare that there are no conflicts of interests.

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