PREPARATION AND CHARACTERIZATION OF FLUCONAZOLE TOPICAL NANOSPONGE HYDROGEL

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ABSTRACT

Objective: The study aimed to develop a polymeric nanosponge-based hydrogel system for enhanced topical application of fluconazole, an antifungal drug.

Methods: Nanospheres were formulated using the emulsion solvent diffusion method using various polymers like hydroxypropyl methylcellulose, ethylcellulose and Eudragit RS 100. Polyvinyl alcohol and ethanol were used to prepare the aqueous and dispersed phases. Nanospheres were dispersed in an appropriate amount of gelling agent Carbopol 940 to get nanosponge gel. Drug-polymer interaction has been carried out by FTIR spectroscopy. The prepared nanospheres were evaluated for various tests like production yield, drug entrapment efficiency, compatibility and SEM studies. The nanosponge hydrogel was tested for pH, drug content, spreadability, in vitro diffusion and kinetic studies.

Results: The drug entrapment efficiency of fluconazole nanospheres was found in the range of 93.43±0.84% to 80.8±0.36% for all formulations, respectively. The spreadability of prepared nanospheres gel formulation was in the range between 5.20±0.19 to 7.187±0.85.

Particle size analysis showed that the average particle size of fluconazole nanospheres formulated using ethyl cellulose (F5) was found to be 334 nm. The zeta potential was found to be -10.4 mV, indicating the formulated fluconazole nanospheres (F5) had moderate stability. FTIR and DSC studies of pure drug and nanospheres suggested that the formulations were stable and there was no chemical interaction with polymer and other excipients. The optimised fluconazole topical nanosponge hydrogel (FGS) released 90.90% drug in 8 h.

Conclusion: Fluconazole topical nanosponge hydrogel could be successfully prepared by emulsion solvent diffusion method. Fluconazole topical nanosponge hydrogel showed promising results under in vitro condition and thus, there exists a scope for evaluation of the developed nanosponge hydrogel for further pharmacokinetic studies, using appropriate test models.

Keywords: Fluconazole, Nanospheres, HPMC, Eudragit RS100, Ethylcellulose, Emulsion solvent diffusion method

INTRODUCTION

Topical and delivery systems aim to deliver active ingredients across the skin. Both systems share manufacturing concerns and risks. Topical preparations can replace needles, offering benefits like avoiding hepatic metabolism and gastric degradation [1]. Conventional topical systems such as ointments and creams are less effective for skin permeation due to their poor absorption of substances upon application to the skin, sustaining the release even after the retention of nanospheres particles on the skin surface [10]. Skin diseases rank 4th globally in nonfatal disease burden. Fluconazole is used to prevent and treat a variety of fungal and yeast infections like candidiasis, blastomycosis, coccidioidomycosis, cryptococcosis, histoplasmosis, dermatophytosis, and pityriasis. It falls under the category of medications known as azole antifungals. Its mechanism involves inhibiting the growth of fungi responsible for the infections. Fluconazole, classified under BCS class II is available in tablet, powder, suspension, cream and gel forms [11]. In the present study, an attempt was made to prepare fluconazole nanosponge hydrogel for efficient delivery of the drug into the skin.

MATERIALS AND METHODS

Fluconazole was obtained as a gift sample from Tagoor Laboratories Pvt Ltd, Hyderabad, Telangana. Hydroxypropyl methyl cellulose, Eudragit RS 100, Propylene Glycol, Ethyl Cellulose were purchased from SD Fine Chemicals Pvt Ltd, Mumbai. Carbopol 940 and Poly Vinyl Alcohol were procured from HiMedia Laboratories Pvt Ltd, Thane, Maharashtra, and Qualigen Fine Chemicals, Umbarde, Maharashtra, respectively.

Preparation of fluconazole nanosponges by emulsion solvent diffusion method

Fluconazole-loaded nanospheres were prepared by using the emulsion solvent diffusion method. Four different polymers,
polvinyl alcohol (PVA), hydroxypropyl methylcellulose (HPMC), ethyl cellulose (EC) and Eudragit RS100 (ERS100) with different ratios were used for nanosponge formulation. Two phases were used, one is organic and the other is the aqueous phase. The organic phase contains a drug and polymer mixture in 30 ml ethanol and the aqueous phase contains PVA and 100 ml distilled water. The aqueous phase was added in a drop wise manner in the organic phase on a magnetic stirrer at 5000 rpm. After two hours of stirring the prepared solution was vacuum dried using a vacuum pump for 20 min, nanosponges were collected by filtration method and dried in an oven at 40 °C for 24 h [12, 13]. A total of twelve formulations were prepared. The composition of fluconazole nanosponges is shown in table 1.

**Evaluation of fluconazole nanosponges**

**Production yield**

Production yield was determined by calculating the initial weight of raw material and the final weight of drug-loaded nanosponges using the formula [14]:

\[
\text{Production yield} = \left(\frac{\text{Practical mass of nanosponge}}{\text{Theoretical mass of nanosponge}}\right) \times 100 \quad \ldots \ldots \quad (1)
\]

**Drug entrapment efficiency (%)**

Accurately weighed nanosponges (10 mg) were suspended in 100 ml of phosphate buffer pH 7.4 solution and later filtered through filter paper, the absorbance of filtrate after appropriate dilution was measured at 261 nm using a UV-visible spectrophotometer [15]. Entrapment efficiency of nanosponges was calculated by using the formula:

\[
\text{Drug entrapment efficiency} = \left(\frac{\text{Actual drug content in nanosponge}}{\text{Theoretical drug content}}\right) \times 100 \quad \ldots \ldots \quad (2)
\]

**Particle size**

The average mean diameter and size distribution of the optimized nanosponge was found by the Dynamic Light Scattering method using the Malvern Zetasizer at 25 °C. The dried nanosponges were dispersed in water to obtain proper light scattering intensity for fluconazole nanosponges [16].

**Zeta potential**

Zeta potential is a measure of surface charge. The surface charge (electrophoretic mobility or movement velocity of the particles in an electric field and the particle charge) of the nanosponge were determined by using a Zeta sizer (Malvern Instruments Ltd.) having zeta cells, polycarbonate cell with gold plated electrodes. The nanosponges were diluted 10 times with distilled water as a medium for sample preparation and analysis. Zeta potential is essential for characterization of the stability of nanosponges [17].

**Scanning electron microscopy (SEM)**

SEM analysis was performed to determine the microscopic characters (shape and morphology) of prepared fluconazole nanosponges. Nanosponges were prepared and dried well to remove the moisture content and images were taken using scanning electron microscopy (Hitachi X650, Tokyo, Japan) in different magnifications.

**Fourier transform infrared spectroscopy (FTIR)**

FTIR study was performed to override the possibility of interaction between drug and polymer. FTIR studies were performed on pure drug fluconazole, nanosponge formulation F4 (HPMC), F8 (EC) and F12 (ERS100). The nanosponges were mixed with potassium bromide (KBr) in 1:90 ratios and compressed in the form of a pellet by 15 tons pressure. The FTIR spectrum was recorded in the wavelength range of 4000 to 400 cm⁻¹ [19].

**Differential scanning calorimetry (DSC)**

A differential scanning calorimeter was used to obtain DSC peaks of pure drug and the prepared nanosponge F5 (EC). The DSC thermogram was obtained by sealing the drug or formulation in hermetically in an aluminium pan and kept under nitrogen purging (atmosphere). The samples were scanned from room temperature to 300 °C and with 10 °C rise/min [20].

**Preparation of fluconazole topical nanosponge hydrogel**

A sufficient amount of gelling agent Carbopol 940 was dissolved and soaked overnight in a suitable quantity of water to get a good dispersion. Later propylene glycol as a penetration enhancer was added. In another beaker, fluconazole nanosponges (equivalent to 80 mg drug) were dispersed in water. This was added to the previous beaker and stirred on a magnetic stirrer at 400 rpm for 20 min [21, 22]. The formulae of nanosponge gels are shown in table 2.

**Characterization of fluconazole nanosponge hydrogels**

**pH**

The pH of the formulation was measured using a digital pH meter (Digisun Electronics, Hyderabad). The pH of the topical gel formulation should be between 6 to 8.

**Drug content**

1 gm of the gel was dissolved in 100 ml of phosphate buffer pH 7.4 and a sample (5 ml) was taken from this solution and diluted to 25 ml. Fluconazole concentration was determined by measuring the absorbance at 261 nm using a UV-visible Spectrophotometer (Shimadzu, UV2600) [23].

\[
\text{Drug content} = \left(\frac{\text{Actual drug content}}{\text{Theoretical drug content}}\right) \times 100 \quad \ldots \ldots \quad (3)
\]

### Table 1: Formulation of fluconazole nanosponges

<table>
<thead>
<tr>
<th>S. No</th>
<th>Ingredients</th>
<th>Formulation codes</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>F1</td>
</tr>
<tr>
<td>1</td>
<td>Fluconazole (gm)</td>
<td>1</td>
</tr>
<tr>
<td>2</td>
<td>PVA (gm)</td>
<td>5</td>
</tr>
<tr>
<td>3</td>
<td>HPMC (gm)</td>
<td>0.6</td>
</tr>
<tr>
<td>4</td>
<td>Ethylcellulose (gm)</td>
<td>-</td>
</tr>
<tr>
<td>5</td>
<td>Eudragit RS100 (gm)</td>
<td>-</td>
</tr>
<tr>
<td>6</td>
<td>Ethanol (ml)</td>
<td>30</td>
</tr>
<tr>
<td>7</td>
<td>Distilled water (ml)</td>
<td>100</td>
</tr>
</tbody>
</table>

### Table 2: Formulation of fluconazole nanosponge hydrogel

<table>
<thead>
<tr>
<th>S. No</th>
<th>Ingredients</th>
<th>Formulation codes</th>
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</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>FG1</td>
</tr>
<tr>
<td>1</td>
<td>Fluconazole nanosponge (mg)</td>
<td>80</td>
</tr>
<tr>
<td>2</td>
<td>Carbopol 940 (mg)</td>
<td>70</td>
</tr>
<tr>
<td>3</td>
<td>Propylene glycol (ml)</td>
<td>0.5</td>
</tr>
<tr>
<td>4</td>
<td>Distilled water (ml)</td>
<td>10</td>
</tr>
</tbody>
</table>

19
Spreadability test

A sliding plate apparatus was used to determine the spreadability of the gel by measuring the diameter of 1 gm gel between horizontal plates after 1 min. Around 125 gm of standardized weight was tied to the upper plate. An excess gel is placed between two glass slides and a 1000 gm weight is placed on them for 5 min, to compress the sample to a uniform thickness. The bottom slide is anchored to the apparatus and weights are placed in the pan. The time in seconds needed to separate the two slides is taken as a measure of spreadability [24, 25]. A shorter time interval indicates better spreadability. Spreadability was determined by using a formula.

\[ S = \frac{M \cdot L}{T} \]  

Where,

S= Spreadability
M= weight tied to the upper slide.
L= length of a glass slide.
T= Time taken to separate two slides (s).

**In vitro diffusion study**

An in vitro diffusion study of nanosponge hydrogel formulations was performed through the cellulose membrane by using a Franz diffusion cell. The receptor compartment was filled with 7.4 pH phosphate buffer and maintained at 32±0.5 °C. The medium was continuously stirred by the help of a magnetic stirrer. 1 gm of the gel was placed over the cellulose membrane. Sampling was done every 30 min. 1 ml sample was withdrawn and diluted with 25 ml pH 7.4 phosphate buffer. The withdrawn sample was replaced with the same amount of buffer to maintain the sink condition. Samples were analyzed for fluconazole content using a UV spectrophotometer at 261 nm [26].

**Kinetic study**

The release mechanism was assessed by fitting the release data to different kinetic equations, including first-order, zero-order, Higuchi and Korsmeyer-Peppas models. Subsequently, the corresponding r² values were determined for each model [27].

**RESULTS AND DISCUSSION**

**Preparation of fluconazole nanosponges by emulsion solvent diffusion method**

Fluconazole-loaded nanosponges were prepared by using the emulsion solvent diffusion method. The selection of polymers for the formulation of fluconazole nanosponges was based on the trial batches carried out by using different polymers such as HPMC, ethyl cellulose and Eudragit RS 100. Drug: polymer ratio was selected based on the literature. By changing the polymer concentration, nanosponge optimization was done. A total of twelve formulations (F1-F12) of fluconazole nanosponges were prepared. The details are given in table 1 and nanosponges are shown in fig. 1. The nanosponges obtained with HPMC were quite transparent, whereas with ethyl cellulose, they were off-white and with Eudragit RS 100 yellowish coloured.

**Evaluation of fluconazole nanosponges**

**Production yield**

The percentage yield of F1 to F12 batches was observed in a wide range from 41.01±0.17% to 77.53±0.62%. The percentage yield was minimum for formulation F1 (41.01±0.17%) and maximum for formulation F8 (77.53±0.62%). From the results, it was observed that as the concentration of polymer increased, percentage yield also increased. The production yield may vary due to the change in polymer concentration. The yield obtained with ethyl cellulose as a polymer was much higher when compared with HPMC and Eudragit RS 100 (table 3) and this result is similar to the finding of Jadhav KR et al (2020) [26].

**Drug entrapment efficiency (%)**

The percentage entrapment efficiency was in the range of 52.3±0.84% to 80.8±0.36% for all formulations. Highest % entrapment efficiency was shown by F4-80.8±0.36% and least by F9-52.3±0.84. It is observed that as the polymer concentration increases the % entrapment efficiency also increases. The change in percentage entrapment efficiency may be due to the changes in the polymer concentration and difference in the degree of cross-linking. The results are similar to the findings by Jelvehgari M et al. (2017) [26] and are shown in table 3.

**Particle size**

Particle size analysis was performed by the zeta size for optimized fluconazole nanosponge F5. The particle size was found to be between 150 nm to 343.2 nm. The average particle size of nanosponge F5 was 334 nm. The zeta size distribution of ethyl cellulose-fluconazole nanosponges is depicted in fig. 2.

**Zeta potential**

Zeta Potential was determined using the Malvern zeta-sizer instrument. Zeta potential analysis is carried out to find the surface charge of the particles to know their stability during storage. The magnitude of zeta potential is predictive of the colloidal stability. Nanoparticles with zeta potential value greater than +25 mV or less than -25 mV typically have high degrees of stability. When particles in suspension exhibit a substantial negative or positive zeta potential, they repel each other, mitigating any inclination for aggregation. Conversely, particles with a low zeta potential lack the repulsive force necessary to deter their aggregation and subsequent

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**Fig. 1: Fluconazole nanosponges developed with different polymers**

A) HPMC  
B) Ethyl cellulose  
C) Eudragit RS 100
Flocculation. For fluconazole nanosponges F5 zeta potential was found to be -10.47 mV. These values indicate that the formulated nanosponge F5 has a high degree of stability. Zeta potential of fluconazole nanosponges is shown in fig. 3.

![Fig. 2: Particle size of fluconazole nanosponge formulation F5](image)

**Table 3: Evaluation of fluconazole nanosponges**

<table>
<thead>
<tr>
<th>S. No.</th>
<th>Nanosponge codes</th>
<th>Production yield (%)</th>
<th>Drug entrapment efficiency (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>F1</td>
<td>41.0±0.17</td>
<td>68.6±0.32</td>
</tr>
<tr>
<td>2</td>
<td>F2</td>
<td>44.0±0.10</td>
<td>69.1±0.55</td>
</tr>
<tr>
<td>3</td>
<td>F3</td>
<td>51.5±0.52</td>
<td>73.3±0.42</td>
</tr>
<tr>
<td>4</td>
<td>F4</td>
<td>64.7±0.32</td>
<td>80.8±0.36</td>
</tr>
<tr>
<td>5</td>
<td>F5</td>
<td>63.0±0.87</td>
<td>68.1±0.36</td>
</tr>
<tr>
<td>6</td>
<td>F6</td>
<td>70.4±0.47</td>
<td>69.8±0.25</td>
</tr>
<tr>
<td>7</td>
<td>F7</td>
<td>75.1±0.79</td>
<td>72.2±0.84</td>
</tr>
<tr>
<td>8</td>
<td>F8</td>
<td>77.5±0.62</td>
<td>75.2±0.11</td>
</tr>
<tr>
<td>9</td>
<td>F9</td>
<td>43.0±0.34</td>
<td>52.3±0.84</td>
</tr>
<tr>
<td>10</td>
<td>F10</td>
<td>43.6±0.15</td>
<td>62.1±0.74</td>
</tr>
<tr>
<td>11</td>
<td>F11</td>
<td>55.7±0.39</td>
<td>66.9±0.69</td>
</tr>
<tr>
<td>12</td>
<td>F12</td>
<td>66.8±0.91</td>
<td>78.0±0.15</td>
</tr>
</tbody>
</table>

*Results are given in mean±SD, n=3*

![Fig. 3: Zeta potential of fluconazole nanosponge formulation F5](image)

![Fig. 4: SEM images of fluconazole nanosponge formulation F5](image)
Scanning electron microscopy

Fig. 4 shows the surface and cross-sectional SEM pictures of the optimized fluconazole nanosponge (F5).

Scanning electron micrographs of the prepared nanosponges at different magnifications showed that the nanosponges were porous with a rough surface morphology. The spongy and porous nature of nanosponges was observed in the SEM images and it could be due to the inward diffusion of fluconazole in the ethylcellulose polymeric surface of nanosponge during the fabrication which could be interrelated with the results Al-Suwayeh et al. (2014) [30].

Fourier transform infrared spectroscopy

FTIR is a versatile tool in pharmaceutics and biopharmaceutics. It has a wide field of applications ranging from the characterization of drug formulations to the elucidation of kinetic processes in drug delivery. In the present study, IR Spectra of pure drug, formulations F4, F8 and F12 are studied in detail. The IR spectra of pure drug and formulations F4, F8 and F12 were used in order to ascertain whether there is any interaction of the drug with excipients. The IR spectrum of the drug gives important signals for functional groups and various bonds in the expected IR region, indicating that the drug is a pure sample. HPMC, Ethyl cellulose and Eudragit RS 100 are the different polymers which are used in the development of required formulations F4, F8, and F12 respectively. In the spectra of all these formulations, the drug has shown characteristic absorption bands in the almost same positions with negligible variation in comparison with IR of the pure sample; this suggests that the drug is in normal form even in the presence of excipients and has not undergone any kind of interactions. Thus, there is no decomposition of the drug and polymer in the developed formulations. The spectra are shown in fig. 5–8.
Differential scanning calorimetry

Differential scanning calorimetry is used to study the thermal stability of the sample. It is mainly useful in the characterization of drug substances and drug products. It is a speedy, simple and consistent technique which allows fast estimation of pharmaceutical drug substances, polymorphic form, excipients compatibility, endotherm and exotherm and impacts of additives on the crystallization of drug substances. Thus, it is an important thermo analytical method and serves as an industrial quality control technique. Drug excipient compatibility studies are an important part of the development of new formulations. DSC study ensures that whether the interaction occurs between a drug and excipients that could affect the properties, stability and efficacy of active ingredients. The thermogram of the pure drug fluconazole is an endothermic curve which showed that the drug starts melting at 139.90 °C. This endothermic peak absorbed at 144.85 °C refers to an endothermic reaction by melting (fig. 9). The actual melting point of the pure drug Fluconazole is 138 °C. Thus, experimentally determined melting point of the pure drug by DSC thermogram matches with the theoretical melting point of the drug. The thermogram of formulation F5 (fig. 10) containing drug and excipients exhibited a broad endothermic peak where the melting of the mixture starts at 178.20 °C, which is approximately taken as 179 °C. This suggests that the drug underwent a transition from a crystalline state to a state of molecular dispersion within the nanosponge. Also, this could confirm encapsulation and interaction of the drug with the nanosponge structure.

Fig. 8: FTIR of fluconazole nanosponge F12 (Eudragit RS 100)

Fig. 9: DSC thermogram of pure drug fluconazole

Fig. 10: DSC thermogram of optimized fluconazole nanosponge F5 (Ethyl cellulose)
Characterization of fluconazole topical nanosponge hydrogel

pH

The pH of fluconazole topical nanosponge hydrogel was determined by using the pH meter. The pH of all formulations was in the range compatible with the normal pH range of the skin (table 4). Hence the preparation is supposed to be non-irritant.

Drug content

It was observed that the drug content in the prepared nanosponges gel was satisfactory and the drug was uniformly distributed in all the formulations. The percentage of drug content was found to be between 91.63±0.93 to 64.10±0.13 (table 4).

Spreadability test

Spreadability is an important characteristic of topical formulation, and it’s responsible for correct dosage transfer to the target site. Spreadability is a factor to be considered in the formulation of gel. The spreadability of the prepared nanosponges gel formulation was in the range between 5.20±0.19 to 7.18±0.85 gm. cm/s, indicating the gel could be easily smeared over the applied affected skin area (table 4).

Table 4: Evaluation of fluconazole topical nanosponge hydrogels

<table>
<thead>
<tr>
<th>S. No.</th>
<th>Nanosponge hydrogel codes</th>
<th>pH**</th>
<th>Drug Content (%)*</th>
<th>Spreadability (gm. cm/s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>FG1</td>
<td>6.8</td>
<td>64.10±0.13</td>
<td>5.20±0.19</td>
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<td>FG2</td>
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<td>5.62±0.26</td>
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<td>3</td>
<td>FG3</td>
<td>7.1</td>
<td>69.70±0.69</td>
<td>5.27±0.37</td>
</tr>
<tr>
<td>4</td>
<td>FG4</td>
<td>7.0</td>
<td>65.37±0.72</td>
<td>5.80±0.44</td>
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<tr>
<td>5</td>
<td>FG5</td>
<td>6.9</td>
<td>91.63±0.93</td>
<td>6.80±0.59</td>
</tr>
<tr>
<td>6</td>
<td>FG6</td>
<td>6.7</td>
<td>78.72±0.85</td>
<td>7.00±0.66</td>
</tr>
<tr>
<td>7</td>
<td>FG7</td>
<td>6.8</td>
<td>77.88±0.69</td>
<td>6.60±0.71</td>
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<tr>
<td>8</td>
<td>FG8</td>
<td>6.7</td>
<td>79.92±0.12</td>
<td>6.00±0.86</td>
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<tr>
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<td>FG9</td>
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<td>FG10</td>
<td>6.8</td>
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<td>FG11</td>
<td>6.6</td>
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<td>FG12</td>
<td>6.8</td>
<td>71.63±0.77</td>
<td>7.18±0.85</td>
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</tbody>
</table>

*Results are given in mean±SD, n=3. **mean of three determinations.

In vitro diffusion study

The in vitro diffusion study showed that formulations FG1, FG5 and FG9 gave the best release of 82.26%, 90.90% and 79.86% drug at the end of 8 h. The formulations with HPMC as polymers released 82.26% (FG1) to 65.60% (FG4) of drug at the end of 8 h. It was observed that increasing the concentration of HPMC decreased the release rate of fluconazole, because HPMC forms a strong viscous gel on contact with aqueous media with the gel controlling delivery of drug. Usually, water-soluble drugs are released primarily by diffusion of dissolved drug molecules across the gel layer. The extent of polymer swelling and the hydration of the microstructure formed within the gel layer vary with the degree of polymer interaction with hydrating media [31]. The formulations with EC as polymers released 90.90% (FG5) to 71.23% (FG8) of the drug at the end of 8 h. The porous matrix formed by EC was conferred for sustained and progressive release of the drug. Slow diffusion of water inside the hydrophobic EC leads to the release of the drug for prolonged periods at a controlled rate. The formulations with Eudragit RS 100 as polymers released 79.86% (FG9) to 71.26% (FG12) of the drug at the end of 8 h. The nanosponges formulated from Eudragit RS 100 showed no burst effect and the drug was released in a controlled manner. It hence shows that the drug is not weakly adsorbed at the surface of nanosponges [32]. The order of release was EC-HPMC-Eudragit RS100. The fluconazole topical nanosponge hydrogel FG5 prepared with ethyl cellulose was selected as the optimum formulation due to its high drug release and comparatively higher yield.

It was observed that the drug release decreased with an increase in the amount of polymer for each type used. It was found that the drug-polymer ratio had a significant effect on the release pattern of the drug. As the drug-polymer ratio increased the release of drug from the nanosponges decreased. This may be attributed to an increase in the thickness of nanosponges, which may decrease drug release. The in vitro diffusion results are shown in fig. 11.

Fig. 11: In vitro diffusion study of fluconazole nanosponge hydrogels FG1–FG12*. N=3, error bars were omitted
Kinetic study

The values of the correlation-coefficient (r²) for all the selected formulations were high enough to evaluate the drug release behavior. The kinetic results revealed that the selected formulations followed zero order, as correlation-coefficient (r²) values (0.9658-0.9958) of zero order are higher than that of first-order values (0.7655–0.9876). The data plotted as per Higuchi kinetics gave fairly linear plots with correlation-coefficient values between (0.9820-0.9848) for all the formulations. The drug release was proportional to the square root of time, indicating that the drug release from nanosponge hydrogel was diffusion-controlled. In the Korsemayer–Peppas model the n values ranged between 0.6797 and 0.7950 suggesting a non-Fickian (Anomalous) release mechanism. This implies a combination of diffusion and chain relaxation mechanism in drug release.

CONCLUSION

Fluconazole topical nanosponge could be prepared by emulsion solvent diffusion method using polymers HPMC, Ethyl cellulose and Eudragit RS 100. It was observed that the concentration of polymer affected the percentage yield and entrapment efficiency. Zeta potential values indicate that the formulated fluconazole nanosponge (F5) had a high degree of stability. SEM images showed the spongy and porous nature of nanosponges. The prepared nanosponge hydrogels gave promising results concerning the release of fluconazole. It was found that the polymer type and the drug polymer ratio had a significant effect on the release pattern of the drug. Fluconazole topical nanosponge hydrogels showed promising results under in vitro conditions and thus, there exists a scope for evaluation of the developed nanosponge hydrogel for further pharmacokinetic studies using appropriate test models.

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Nil

AUTHORS CONTRIBUTIONS

The corresponding author, Sarfaraz Md wrote the manuscript and assisted in the experiments. Shaili Kamal and Arshad Husain performed the experimental works and H. D. Doddavaya supervised the work. All authors discussed the results and contributed to the final manuscript.

CONFLICT OF INTERESTS

Declared none

REFERENCES