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Original Article

THE METAL COMPLEXES OF 1-(PHENYLAMINO)-4, 4, 6-TRIMETHYL-3,4-DIHYDROPYRIMIDINE-2-(1*H*)-THIONE: PREPARATION, PHYSICAL, SPECTROSCOPIC STUDIES AND ANTIBACTERIAL PROPERTIES

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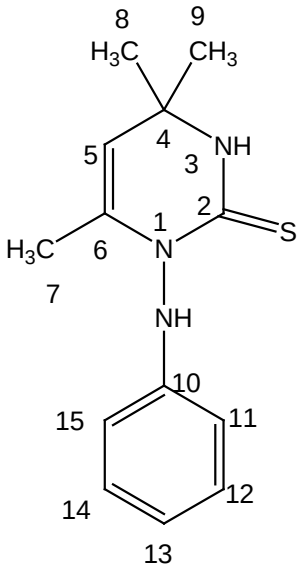


Fig. 1: Structure of Ligand (Hpmpt)

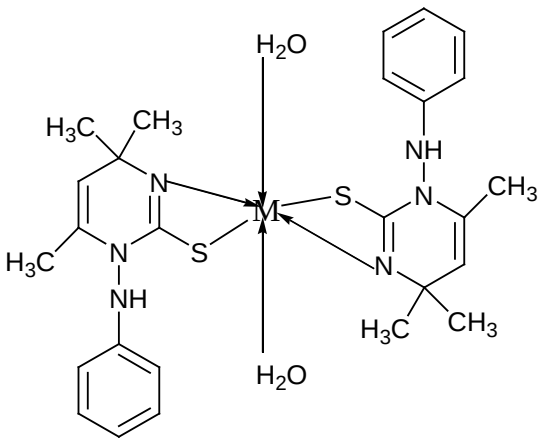


Fig. 2: Structure of $[M(pmpt)_2(H_2O)_2]$ (M(II) = Mn, Co, Ni, Cu)

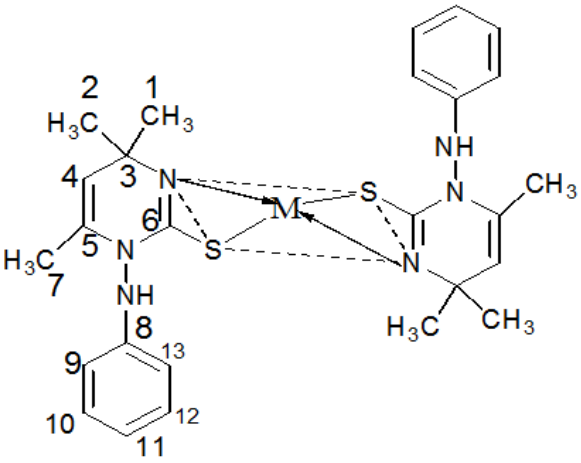


Fig. 3: Structure of $[Zn(pmpt)_2]$, $[Pd(pmpt)_2]$, $[Cd(pmpt)_2]$

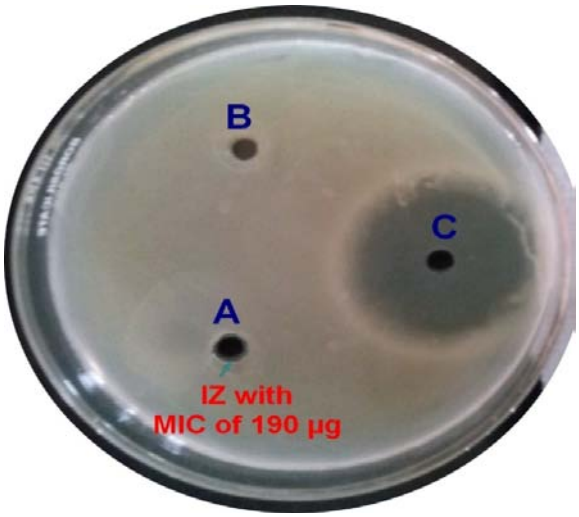
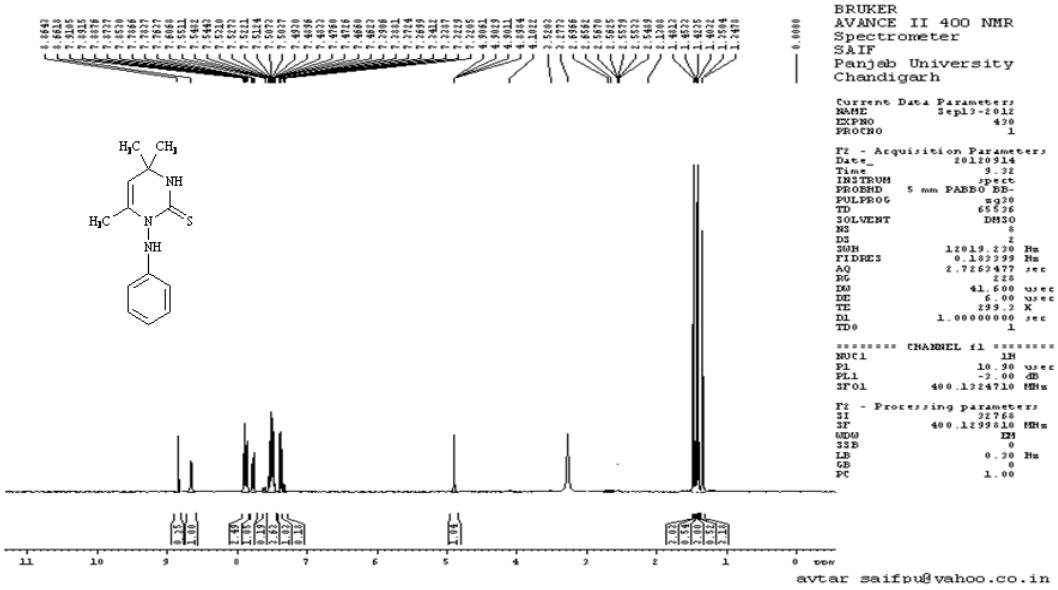
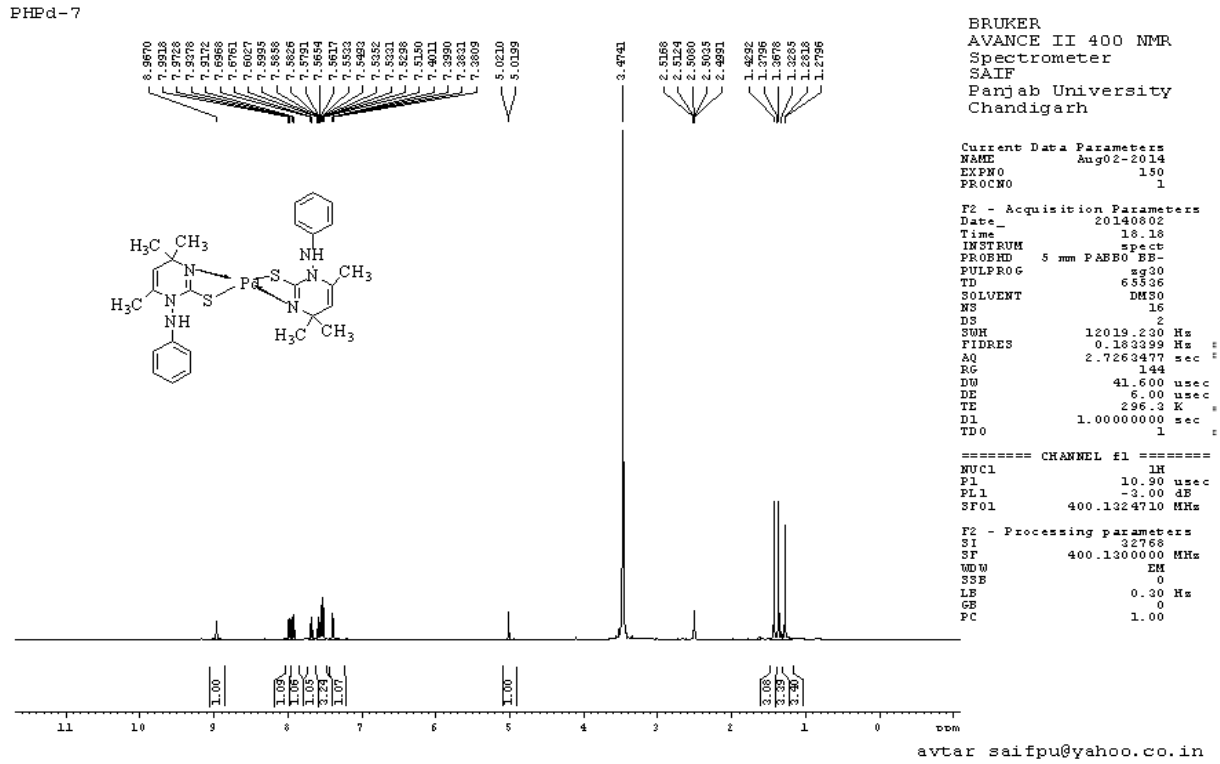


Fig. 2: Agar plate exhibiting the (A) Inhibition zone produced revealing the MIC value of 190µg for the compound, (B) No inhibition zone produced when DMSO alone is poured in the well and (C) Inhibition zone produced when similar quantity i. e. 190µg of chloramphenicol is poured in well

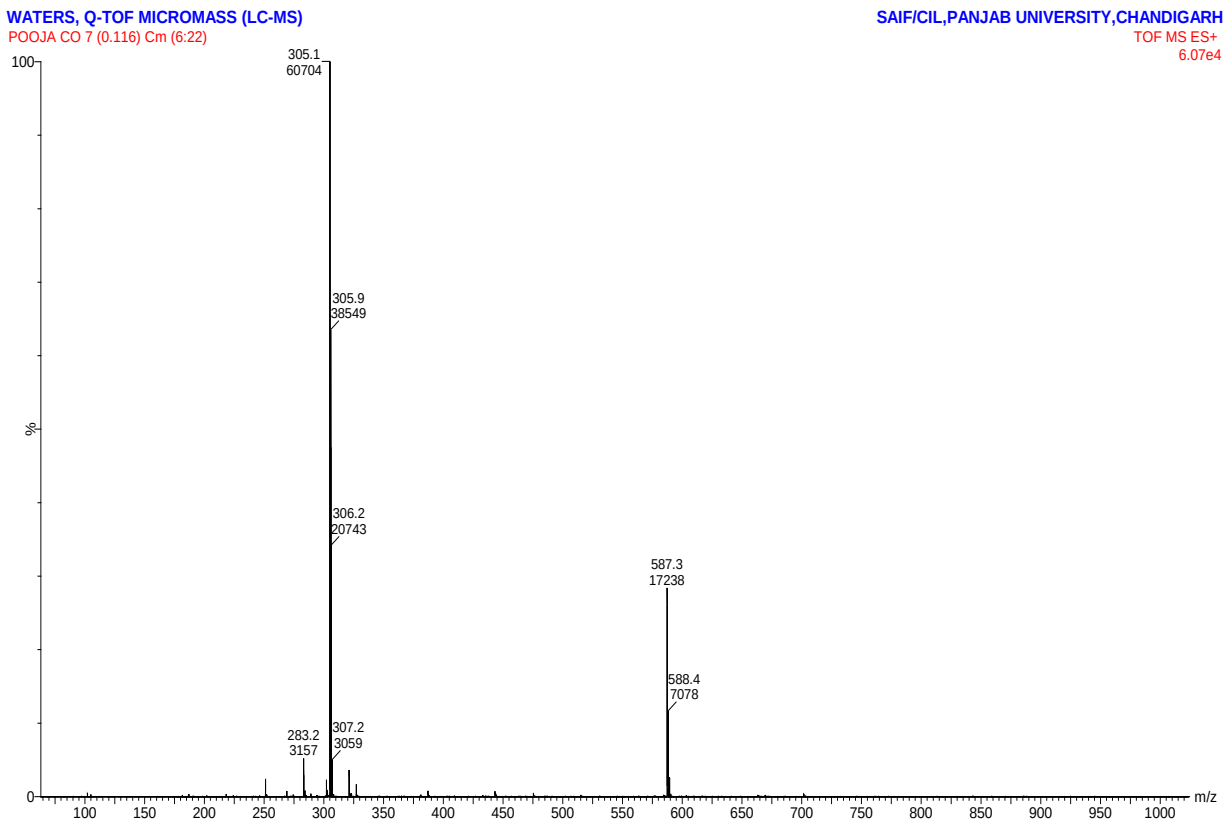
NMR of Ligand



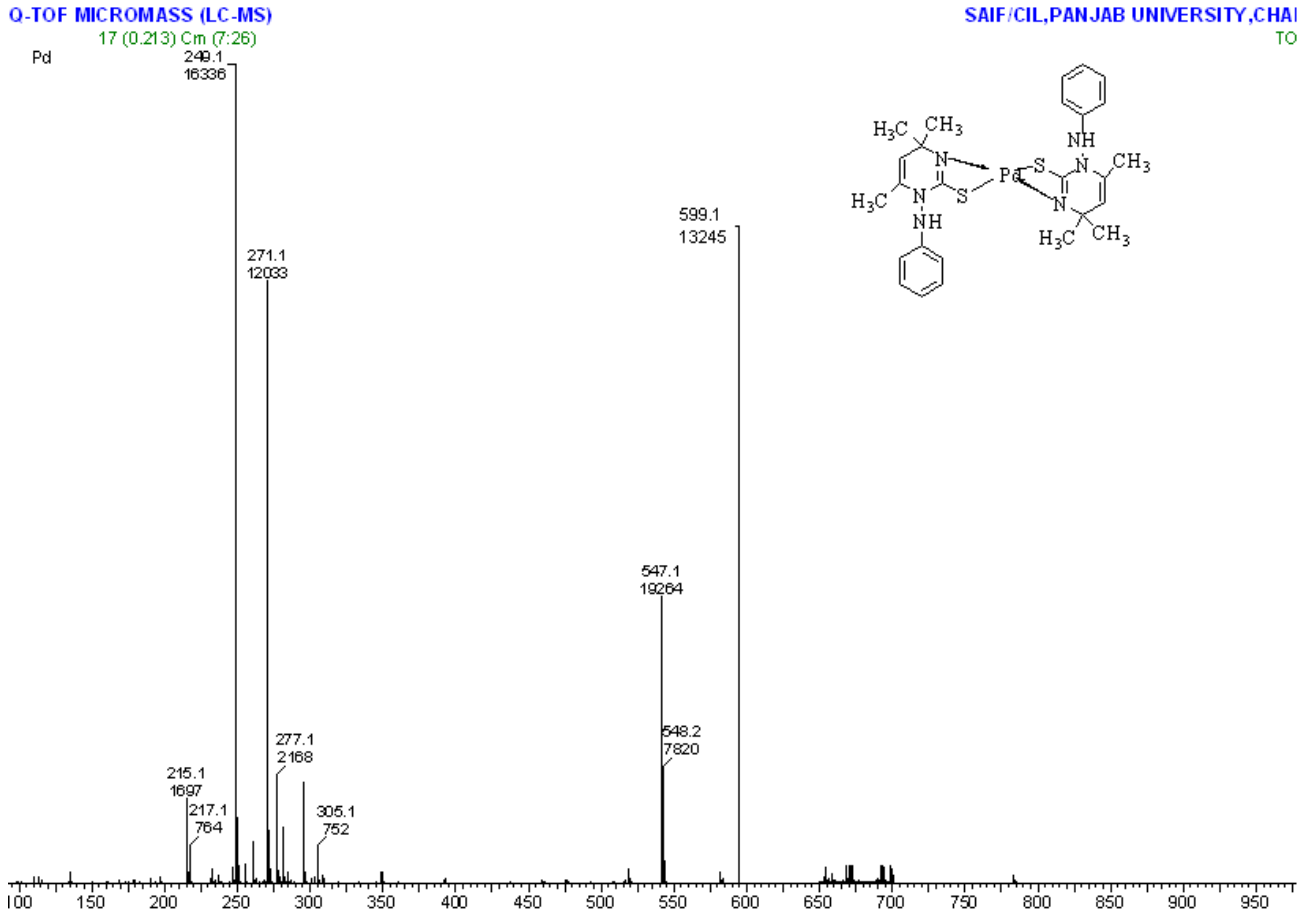
NMR of Pd complex



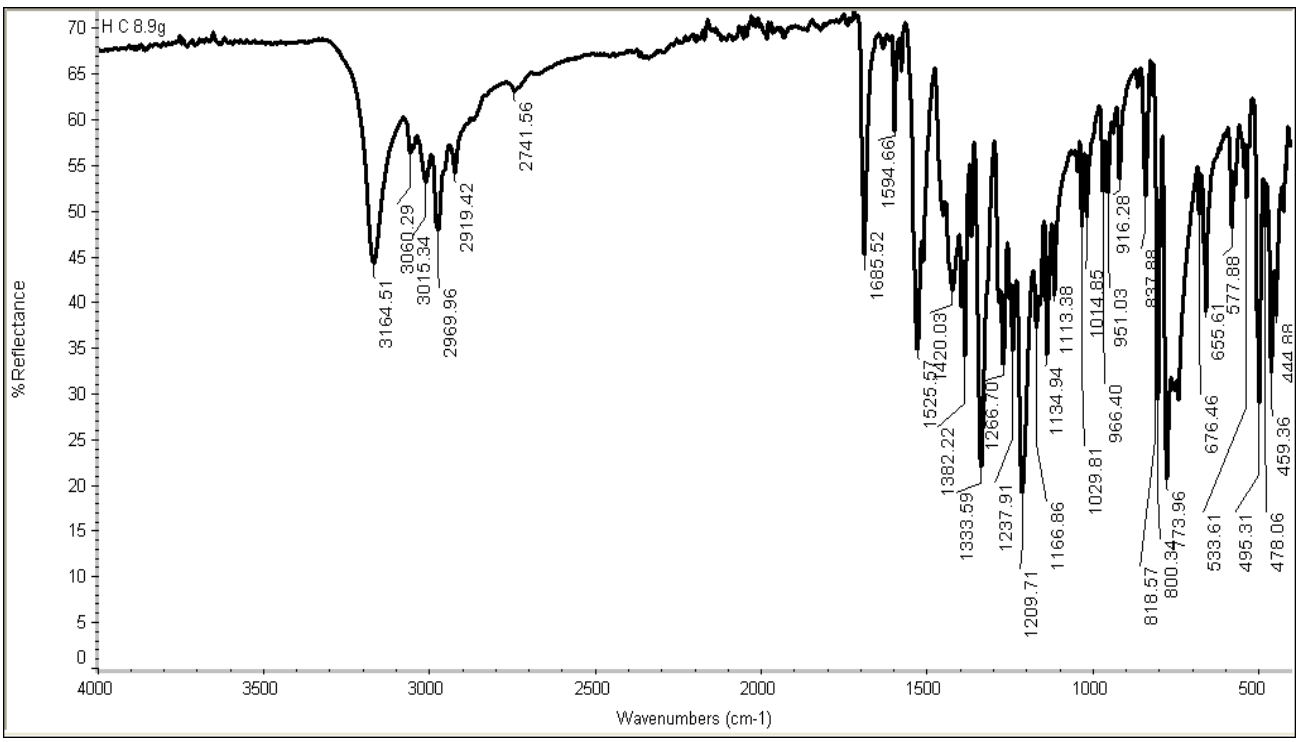
Mass of Co complex



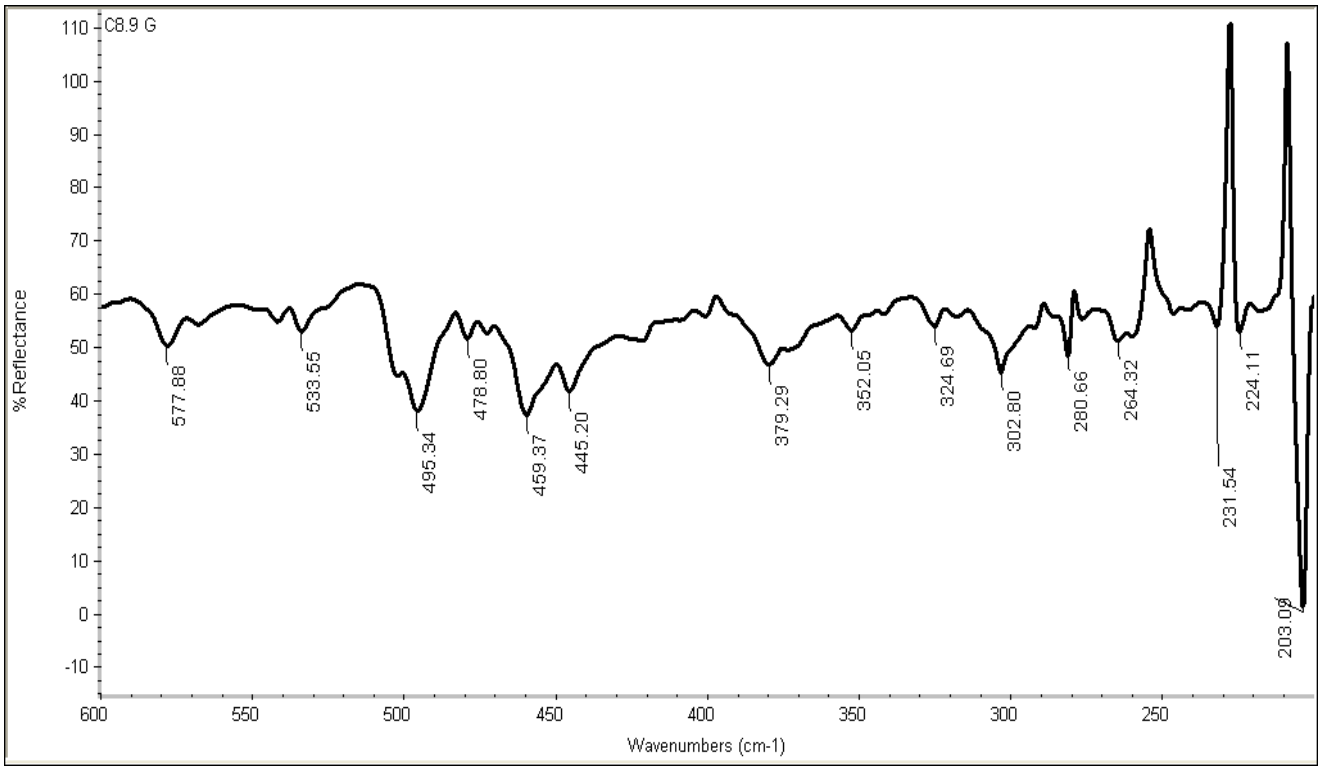
Mass Spectrum of Pd complex



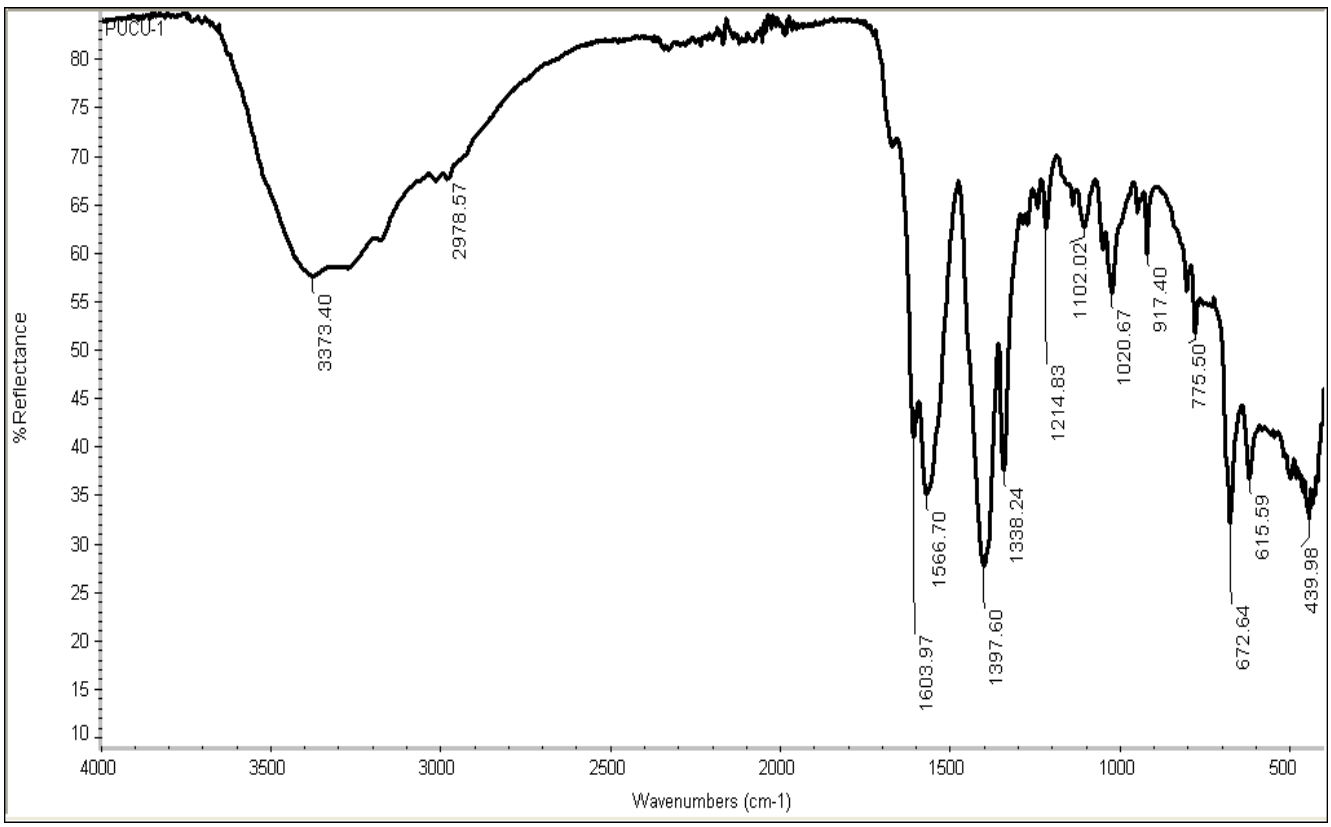
IR of Ligand (upper range)



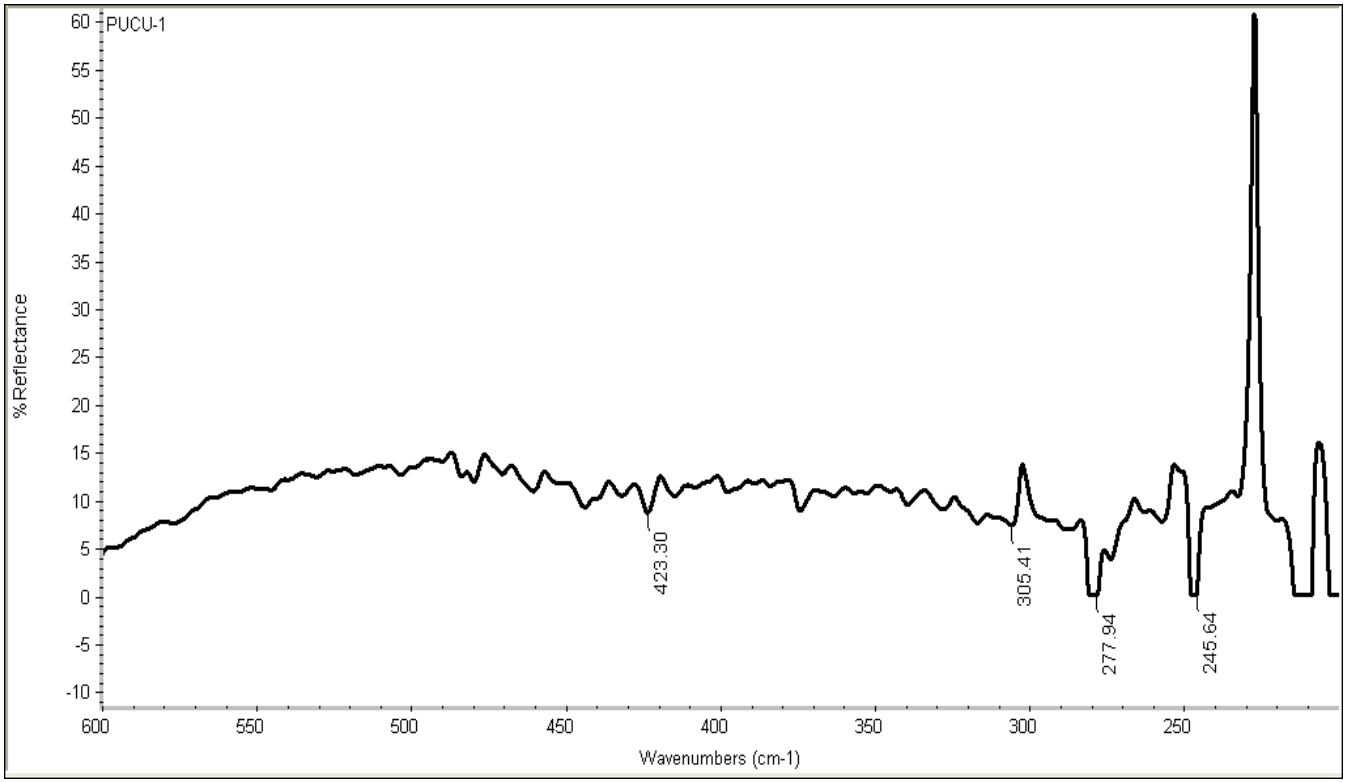
IR of Ligand (low range)



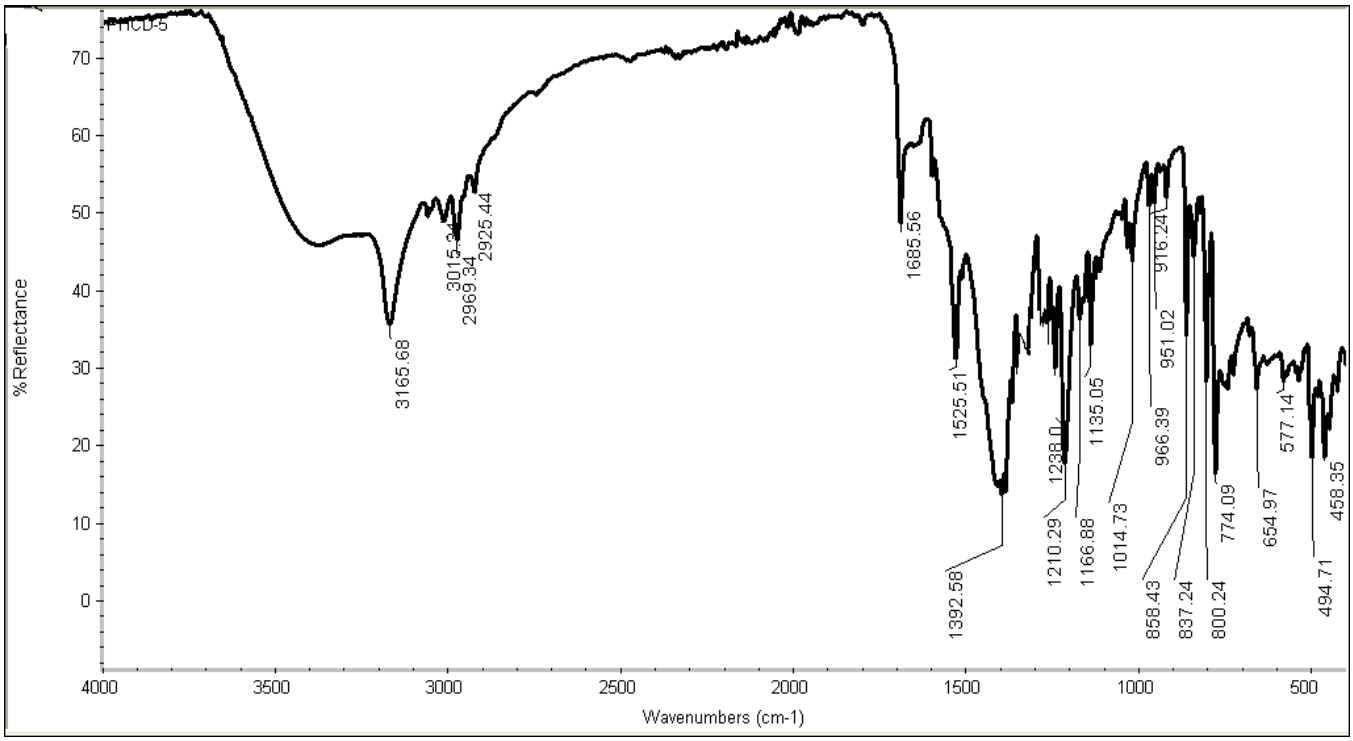
IR of Cu complex (upper range)



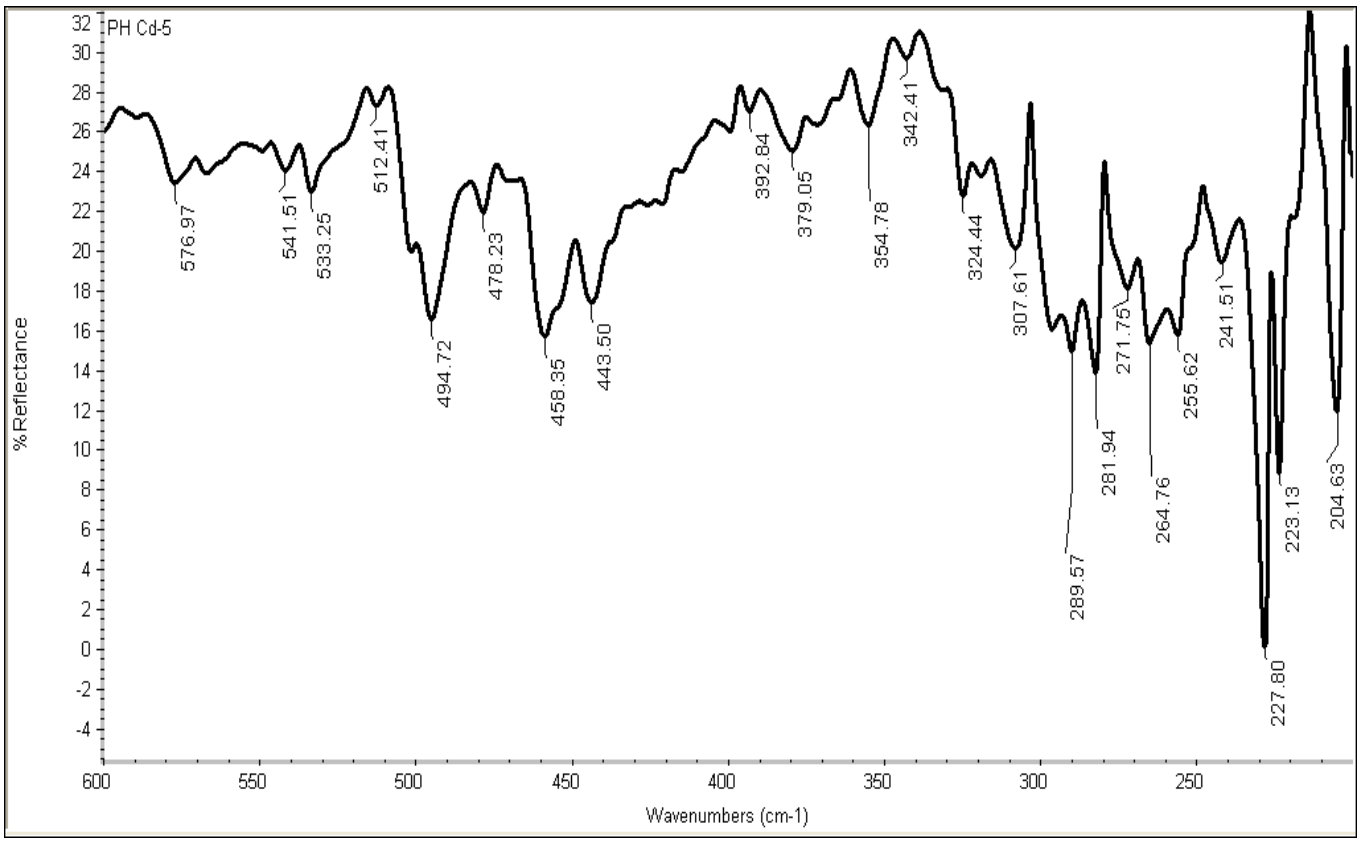
IR of Cu complex (Low range)



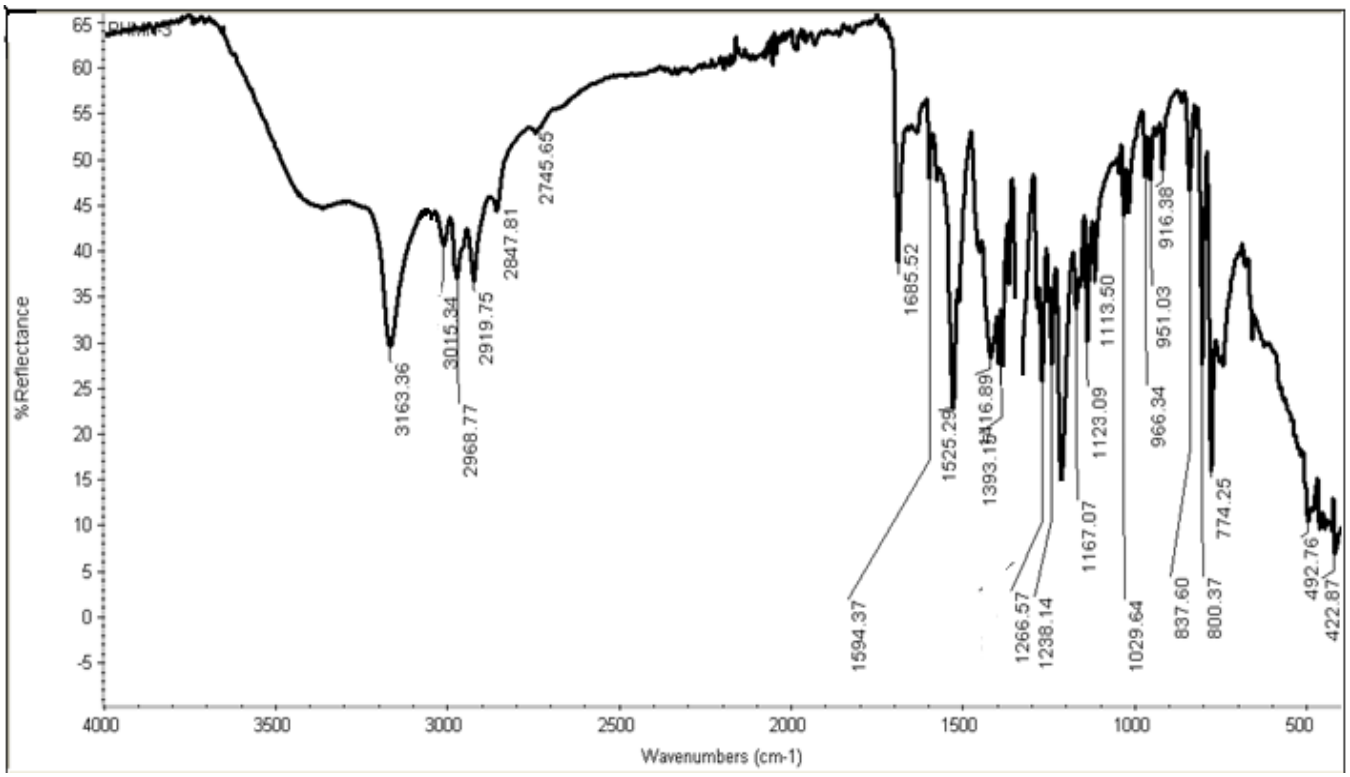
IR of Cd complex



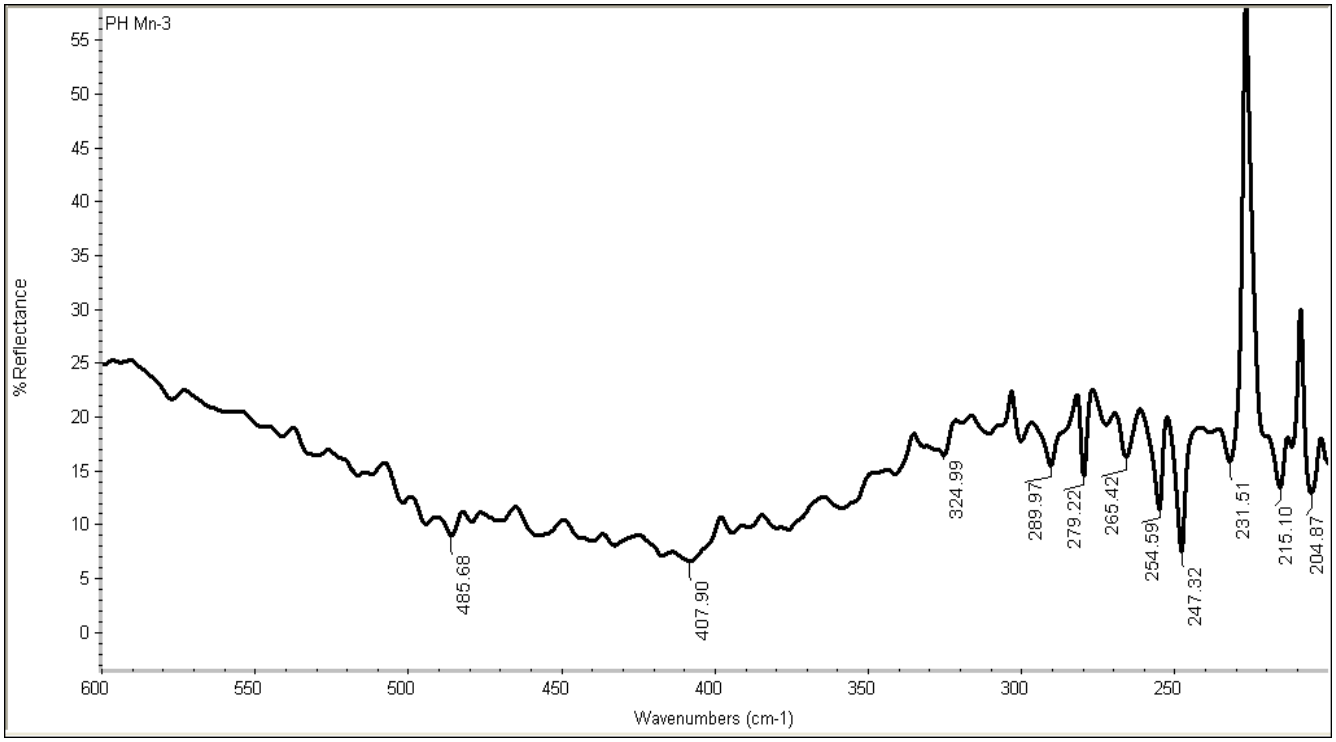
IR of Cd complex (low range)



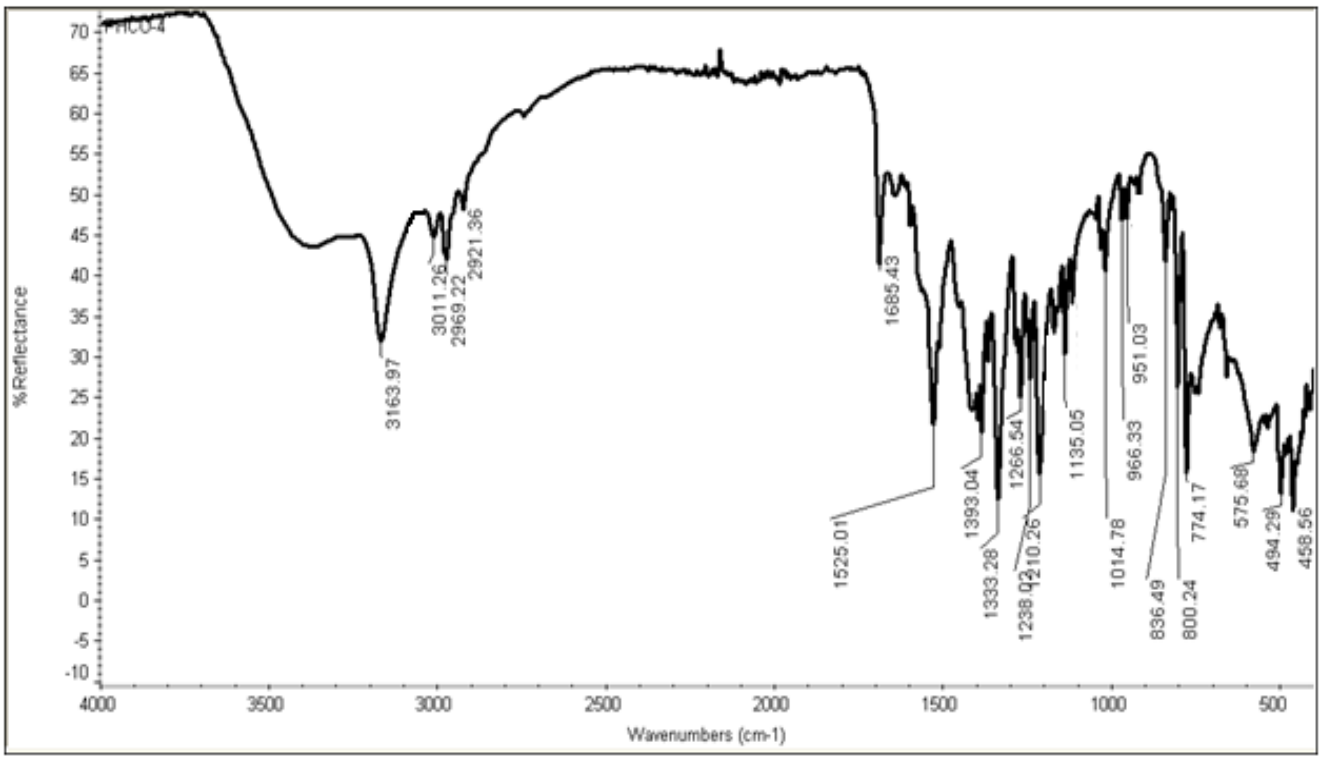
IR of Mn complex (upper range)



IR of Mn complex (Low Range)



IR of Co complex (upper range)



IR of Co complex (Low range)